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Determination of the pore structure of porous materials using electrical conductance

Brett William Gunnink
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electrical conductance**

Gunnink, Brett William, Ph.D.

Iowa State University, 1987

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Determination of the pore structure of porous materials
using electrical conductance

by

Brett William Gunnink

A Dissertation Submitted to the
Graduate Faculty in Partial Fulfillment of the
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INTRODUCTION

The pore-volume distribution of a porous solid as a function of pore size has become progressively more important in understanding chemical and physical behavior of porous materials. The purpose of this dissertation is to introduce a new method for determining pore size distributions and demonstrate the significance of this new methodology to the field of civil engineering materials. This method has been called conductometric phase transition porosimetry (CPTP).

The methodology combines the plastic ice model for solid-liquid phase transitions of pore water with relationships between electrical conductance and temperature, and conductance and pore structure. With measurements of the electrical conductance and the corresponding temperature of a material subjected to a cycle of capillary freezing and thawing, a pore size distribution is obtainable.

It is well-known that the liquid to solid and solid to liquid phase transitions of water in capillaries occurs at temperatures below 0°C , the normal melting point of bulk water. The plastic ice model gives a thermodynamic relationship between the effective pore radii and the temperatures at which these phase transitions occur. For the solid to liquid phase transition, the radius given by this relationship is the pore body radius, while for the liquid to solid transition it is the radius of the pore constriction or neck.

In addition, many porous materials consist of a nonconducting solid phase which has a void system consisting of randomly intersecting

capillaries of various sizes. If the material is saturated, the pore solution generally will exhibit electrolytic conductance which results from the mobility of ions in the pore solution. In CPTP, it is assumed that the conductance through the solid phase, including ice in frozen pores, is negligible when compared to the electrolytic conductance of the unfrozen pore solution. Thus, when the temperature of a mass of saturated porous material is lowered from 0°C , the pore solution will freeze and a decrease in conductance will be observed. Capillaries will begin to freeze with larger sizes freezing at higher temperatures in accordance with the plastic ice theory.

An automated Conductometric Phase Transition Porosimeter has been built and used to determine the pore size distribution of a porous Vycor glass sample and a comparative analysis of these results with those obtained by mercury porosimetry and phase transition porosimetry is presented.

In addition, since perhaps the greatest advantage this method of porosimetry has over other available methods is the ability to test much larger samples, example results obtained from the testing of 4-inch diameter by 4-inch high portland cement mortar and concrete cylinders are also presented. For the cement mortar tested, the pore size distribution obtained is compared with the pore size distribution obtained from mercury porosimetry.

The ability to evaluate the pore size distributions of heterogeneous conglomerate materials such as portland cement concrete should provide a

valuable new tool for investigating the physical and chemical behavior of these materials.

REVIEW OF LITERATURE AND THEORY

Pore Structure Analysis Methods

As stated in the introduction, the pore-volume distribution of a porous solid as a function of pore size has become progressively more important in understanding chemical and physical behavior of porous materials. In order to understand and predict the macrobehavior and properties of an engineering material, it is often necessary to investigate the microstructure of the material. For many porous engineering materials, including soil, concrete, stone and brick, the void space of the material greatly effects the mechanical properties of the material. Thus, an analysis of the microstructure of these materials must include the determination of the pore-size distribution to accurately predict their macro-behavior.

Two independent methods which have found large scale application exist for pore size distribution analysis. They are capillary condensation and mercury porosimetry.

Capillary condensation

The first, capillary condensation, is also often referred to as gas or vapor absorption. For this method, total pore volume and pore-size distributions are determined from gas-adsorption isotherms. The pore volume is the volume of condensed adsorbate at the saturation vapor pressure. When the pressure is reduced, the adsorbate evaporates from the larger capillaries first as dictated by the Kelvin equation. The Kelvin equation can be expressed as (Gregg and Sing, 1967):

$$RT \ln \left(\frac{p}{p_0} \right) = -\gamma V_L \left(\frac{1}{r_1} + \frac{1}{r_2} \right) \quad (1)$$

where p is the vapor pressure and p_0 is the normal vapor pressure of the adsorbate. Also, T is the temperature (expressed in $^{\circ}\text{K}$), γ and V_L are the surface tension and the molar volume of the adsorbate in liquid form, respectively, and R is the molar gas constant. The terms r_1 and r_2 are the principal radii of curvature of the meniscus. The effective radius or Kelvin radius, r_k , is defined to be equal to the pore radius, r_p , minus the film thickness of adsorbed vapor in the pores, t (see Figure 1a). Considering adsorption and desorption separately for an open-ended cylindrical pore as depicted in Figure 1a and 1b, and assuming the solid phase is completely wettable, then the appropriate values for r_1 and r_2 in equation 1 are r_k and ∞ during adsorption. During adsorption, for cylindrical pores with openings at only one end and for all cylindrical pores which remain filled with adsorbate during desorption, the appropriate values for r_1 and r_2 are r_k . From this, it can be shown that for a given volume, (p/p_0) during adsorption will be greater than it is for desorption. This difference is one source of the hysteresis loop that typically occurs between adsorption and desorption curves. Figure 2 illustrates typical adsorption and desorption isotherms.

Another source of hysteresis occurs in bottle-necked pores (see Figure 1c). With this type of pore, during adsorption condensation occurs in the neck of the pore first in accordance with the Kelvin equation and later in the body. However, during desorption evaporation in the pore body is inhibited by the presence of liquid in the pore neck

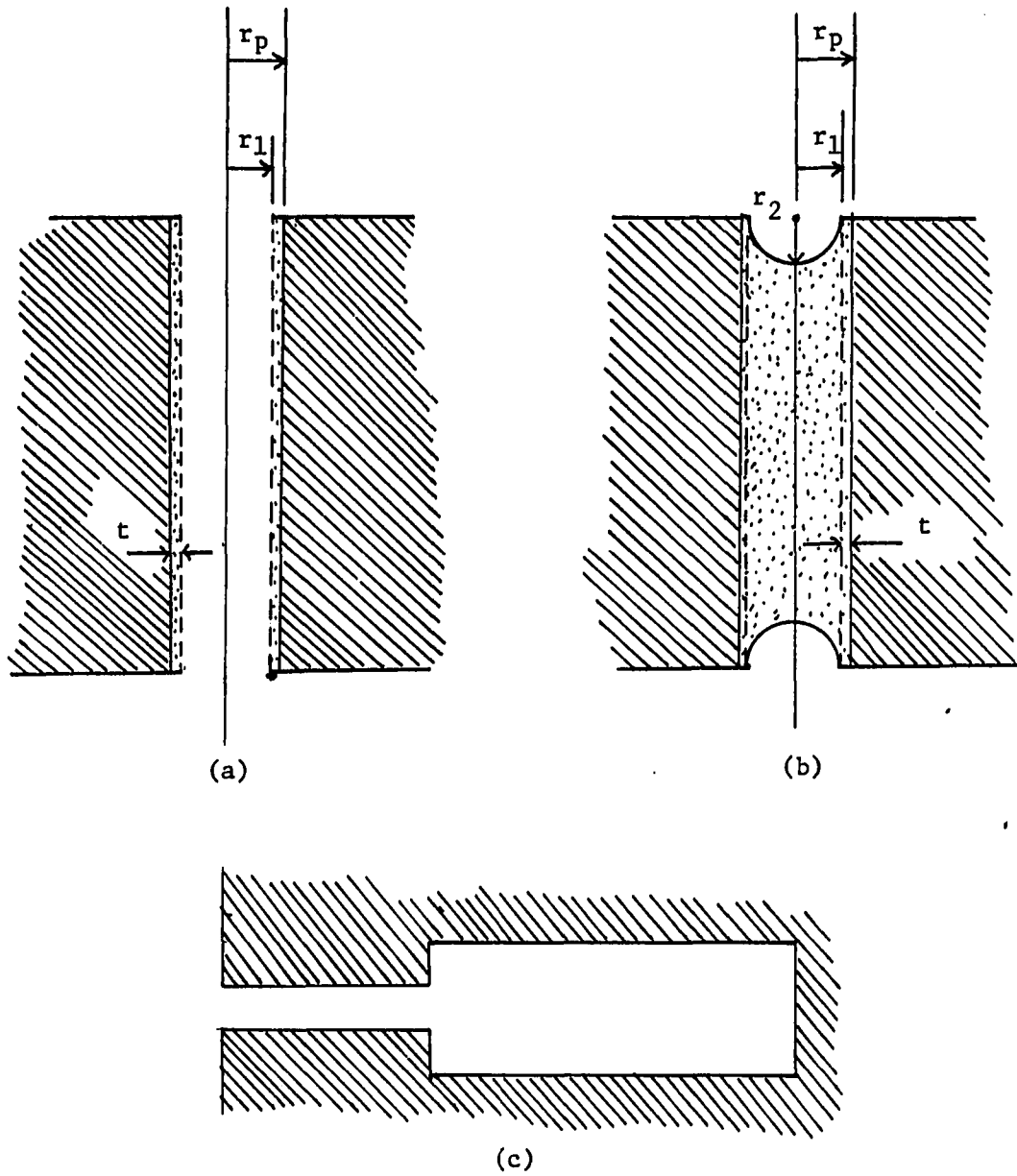


Figure 1. Illustration of capillary condensation: (a) adsorption in, and (b) desorption out of a cylindrical pore with two open ends and (c) example of a bottle-necked pore.

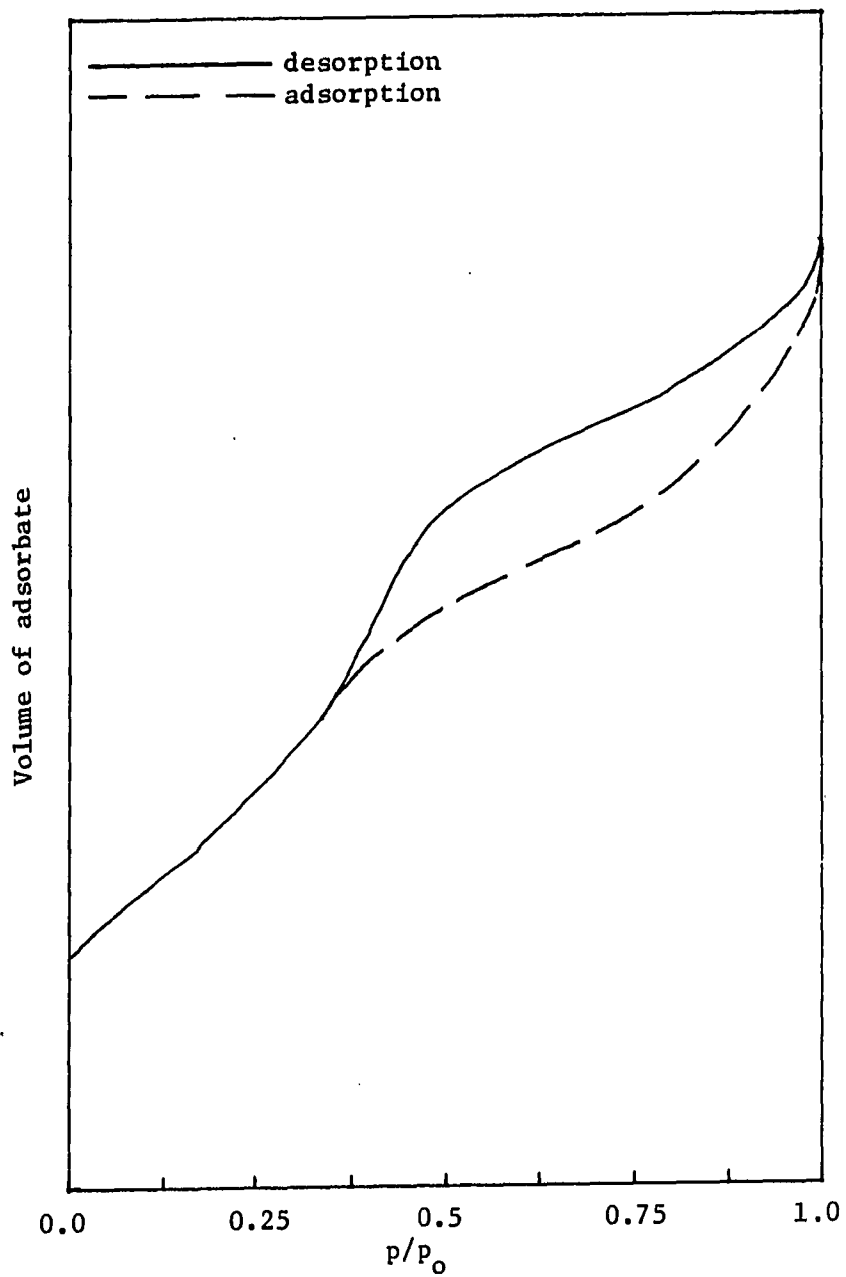


Figure 2. Example of adsorption and desorption isotherms

and will not occur until evaporation occurs in the neck in accordance with the Kelvin relationship. Due to diverse views over the explanation of the hysteresis loop for a particle pore system, the identification of the true equilibrium branch of sorption isotherms of porous solids has long been a subject of debate (O'Neil, 1985; Brunauer et al., 1967a; Brunauer, 1943).

Using the Kelvin equation, the value of r can be calculated for any given point on the isotherm. If the amount adsorbed on the walls is neglected, then the volume at that point on the isotherm would be equal to the volume of all the pores which have radii up to and including r . A graph of volume versus radius could be made and the pore size distribution would result. When allowances for the thickness of the adsorbed film, t , are made, the relevant Kelvin radius is equal to $r_p - t$. Then, a determination of the adsorbed film thickness becomes necessary.

Wheeler (1951) first proposed the sorption isotherm method and considered the thickness to be equal to the monolayer capacity as calculated from BET theory (Brunauer et al., 1938). A method introduced by Barrett et al. (1951) and expanded by others (Pierce, 1953; Cranston and Inkley, 1957) has found universal acceptance and is known as the BJH method. The appropriate values for film thickness are still a matter of debate. This is evident in the work of O'Neil (1985), in which parameters affecting the calculation of pore volume distributions are investigated.

Brunauer and co-workers (Brunauer et al., 1967a; 1967b; Mikhail et

al., 1968a; 1968b; Hagymassy et al., 1969; Skalny et al., 1971) broadened the range of pore sizes measurable by adsorption isotherms and eliminated some uncertainties by introducing two new methods for the analysis of isotherm data. The micropore analysis method (also referred to as the MP method) allows the size distributions of pores with radii smaller than 16 angstroms to be determined. The second method, the corrected modeless method, introduced the concept of hydraulic radius which is defined as the volume of pores divided by the surface area of pores. If this method is used, no assumption need be made regarding pore shape. The Kelvin radius in the Kelvin equation is replaced with hydraulic radius and the adsorbed film thickness is based on the value of the BET parameter.

Enüstün (1981) points out that vapor pressures of various liquids in capillaries of a few micron in diameter are considerably smaller than those predicted by the Kelvin equation and states that these deviations cannot be attributed entirely to multilayer adsorption at the solid/vapor interface. He states that additional deviations are primarily due to increases in the liquid/vapor interfacial tension near the solid surface. This causes perturbations in the liquid/vapor meniscus such that the meniscus is not spherical but rather the radius of curvature is greater near the sides of a capillary than it is near the center of the capillary.

A researcher must evaluate the net effect of the aforementioned uncertainties for a particular pore system in order to identify the true equilibrium branch for that system if indeed one does exist. The desorption branch has found large-scale acceptance in the literature as

being representative of the size of the pore constrictions, while the adsorption branch has generally been assumed to represent the size of the pore bodies.

Currently, commercially available sorption instruments are capable of determining pore size distributions for pores ranging from smaller than 16 angstroms to approximately 5000 angstroms and costs range from about \$25,000 for simple models to about \$100,000 for top of the line models.

Mercury porosimetry

The second method, mercury injection porosimetry, was suggested by Washburn (1921) and later developed by Ritter and Drake (1945) and Drake (1949), and further developed by Winslow and Shapiro (1959). Since most materials are non-wettable by mercury, a minimum pressure, p , is required to force mercury through a constriction. The pressure required for intrusion into a circular pore of radius r is given by the Young and Laplace equation and may be written as (Allen, 1975):

$$p = \frac{2\gamma \cos \theta}{r} \quad (2)$$

where γ is the surface tension of mercury and θ the angle of contact between the meniscus and the walls of the capillary.

The pore-volume distribution is determined by measuring the volume of mercury intruded into a sample as the pressure is increased. Measured values for pressure are converted to pore radii using equation 2.

As with sorption isotherms, the intrusion and extrusion branches of

mercury penetration tests do not follow the same path giving rise to a hysteresis loop. For mercury porosimetry, the hysteresis problem is further complicated because the hysteresis loop does not close for many materials due to entrapment of mercury. Figure 3 illustrates a typical mercury penetration cycle for which the hysteresis loop does not close.

Recently two theories have been advanced to explain the hysteresis and the entrapment of mercury after extrusion. Hill (1960) states that intrusion advances axially while extrusion advances radially. Assuming this premise to be correct, Hill shows that the extrusion pressure should be equal to half the intrusion pressure for a cylindrical pore. Hill also presents experimental evidence in support of this theory. This theory also predicts that if the neck radius is less than half the body radius for a system of essentially cylindrical pores, then entrapment of mercury will occur in the pore bodies.

The second theory has been introduced by Lowell and Shields (1984). They state that after intrusion, mercury atoms fall into the potential field of the walls of the capillary and they acquire a lower free energy than possessed during intrusion. Therefore, extrusion occurs at a lower pressure with a lower contact angle than during extrusion. The theory also states that extrusion stops when the interfacial free energy near a pore constriction equals the pore potential. This in turn results in the entrapment of mercury in the pore bodies.

Additional controversies exist with regard to mercury porosimetry beyond the effects of constrictions on intrusion and extrusion. These include debates about the effect of kinetic hysteresis due to hydraulic

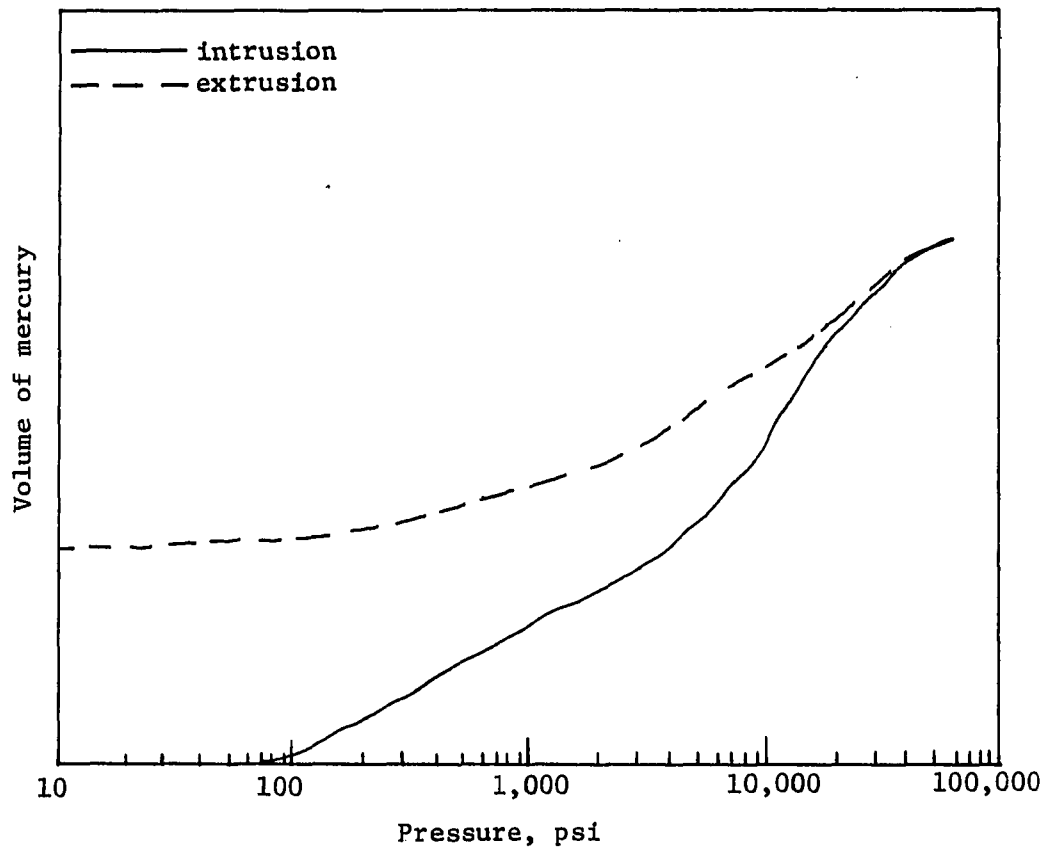


Figure 3. Example of intrusion and extrusion branches from mercury porosimetry

drag, the effect of the magnitude of the contact angle, and the effect of differences between contact angles for advancing and retreating menisci (Conner et al., 1984).

In spite of the aforementioned problems associated with hysteresis loops, the intrusion cycle has generally been assumed to represent the size of the pore constrictions and the extrusion cycle, the pore bodies. Often, due to problems associated with entrapment of mercury, the extrusion cycle is ignored.

In an effort to better interpret the information obtained from the extrusion cycle, Cebeci et al. (1978) refined a technique applied by Hill (1960) and Caro and Freeman (1961) in which, after extrusion, a specimen is intruded a second time. Cebeci et al. (1978) suggest this technique to distinguish the distribution of uniform pores from that of pores with mercury entrapping constrictions.

Mercury porosimetry has been reviewed in more detail by Rootare (1968). Currently commercially available instruments are capable of measuring pore size distributions for pores ranging from 18 to 10,000 angstroms and like capillary condensation instruments are quite expensive, with prices starting at around \$40,000.

The purpose of this section is not to provide a complete literature review of these two powerful methods of porosimetry, but rather to point out that the measurement of the pore structure of a material is not a routine analysis. Thus, when researchers refer to materials of known pore size distributions, they are generally overstating the analysis and should actually be using terminology such as pore size distribution as

determined by mercury porosimetry or nitrogen sorption. Researchers should fully understand the nature and limitations of the analysis technique used to be able to give correct interpretations.

Phase transition porosimetry

Recently, a method was developed based on observation of the liquid-solid phase transition point of pore water in a porous sample at the saturated state by dilatometric measurements (Enüstün et al., 1985a; 1985b; Eckrich et al., 1986). This method has been referred to as ice porosimetry or phase transition porosimetry. A similar method has been proposed by Brun et al. (1977). This method has been referred to as thermoporometry and is based on calorimetrically measured freezing points of water in a saturated porous material. Brun assumes freezing is initiated by homogeneous in-situ nucleation. It is in this latter respect that phase transition porosimetry differs from thermoporometry. It has been shown previously (Enüstün et al., 1978; 1985a; 1985b) that the plastic ice model put forward by Everett (1961) is applicable to freezing and melting of the pore water in a saturated porous material. A discussion of the plastic ice theory is included later in this dissertation. With phase transition porosimetry, the process of the phase change is followed by measuring the volume change of the sample. The principles of phase transition porosimetry are discussed in detail by Enüstün et al. (1985a; 1985b) and Eckrich et al. (1986).

Commercially produced phase transition porosimeters are not yet available, but laboratory prototypes are capable of determining pore size

distributions for pores ranging from about 12 to 5000 angstroms.

Importance to Civil Engineering Materials

The determination of pore size distributions is important in the characterization of a wide variety of materials. Rootare (1968) cites published papers reporting pore size distribution data for a variety of materials including paper, soft woods, textiles, leather, PVC resins, macroreticular ion exchange resins, carbons, coal, coke, membrane filters, Florida filter clays, clay building bricks, petroleum reservoir rocks, phosphate rocks and triple super phosphate, refractory materials, porous iron, aluminum oxides, catalysts and other porous materials.

In the field of civil engineering materials, investigations of the pore structure of portland cement paste and mortar, aggregates, and soils have been undertaken.

Mikhail et al. (1964) and Bodor et al. (1970) introduced and developed procedures for the analysis of the pore structure of hydrated cement pastes from nitrogen sorption isotherms. Odler et al. (1972) and Hagymassy et al. (1972) introduced experimental techniques for determining the pore size distribution of hydrated portland cement pastes by water vapor adsorption. Litvan (1976) states specific surface areas of hydrated portland cement pastes computed from water adsorption isotherms are generally much larger than those calculated from nitrogen isotherms. This difference is attributed by Litvan, not to deficiencies of the measuring methods, but to real difference evoked by the particular drying technique used. Drying techniques explored by Litvan included

drying over a desiccant; vacuum drying; storage in methyl alcohol and/or pentane, followed by vacuum distillation at room temperature, evaporation, or vacuum distillation above the critical point of the adsorbate. Elapsed time for the different combinations used ranged from 6.5 hours to 134 days and results measured by nitrogen adsorption varied from 69.6 to 249.4 m/g. Water sorption surface area sample preparation included allowing the sample to equilibrate at approximate atmospheric relative humidity for 133 days followed by vacuum drying. The water surface area from the desorption isotherm was 286 m/g, while for adsorption, 182 m/g. These results illustrate the importance of sample preparation on surface area determination and thus also on pore distribution determinations for hydrophilic materials like portland cement, soils, clays, and some aggregates. Any method for determining the pore size distributions of these materials which minimizes sample disturbance is more likely to produce results more directly applicable to engineering problems.

Winslow and Diamond (1970) used mercury porosimetry to study the evolution of porosity in portland cement paste and Diamond (1971) presented a critical comparison of mercury porosimetry and capillary condensation pore size distributions of portland cement pastes in which he favored mercury porosimetry. This set off a rather heated debate as to the merits of the two methods (Beaudoin, 1972; Diamond, 1972; Mikhail et al., 1975). Regardless of the relative merits or demerits of the two methods, both necessitate the drying of the sample prior to testing.

Lemish et al. (1958) and Hiltrop and Lemish (1960) first applied

mercury porosimetry to measure the pore size distribution of aggregates and attempted to associate it with the frost durability of concrete. They did not find any close correlation between pore size distribution and service records but stated that knowledge of a rock's pore size distribution might be of value in attempting to predict its service record as an aggregate.

Recent work by researchers at Purdue University illustrates the difficulties encountered when trying to relate measurements of properties of a component of portland cement concrete to behavior of portland cement concrete. In this case, researchers attempted to relate pore size distributions of coarse aggregates to the field service records of concrete containing these aggregates. Kaneuji (1978) introduced an empirical "Expected Durability Factor" which was used to predict the freeze-thaw durability of laboratory concrete specimens. The expected durability factor is defined as:

$$EDF = K_1/PV + K_2(MD) + K_3 \quad (3)$$

where: EDF = Expected Durability Factor;

PV = mercury intruded pore volume for pores larger than 45 angstroms in diameter, expressed in cc/g;

MD = median diameter of pores larger than 45 angstroms in diameter, expressed in μm ;

K_1 , K_2 , and K_3 are empirical constants with values of 0.579, 6.24, and 3.04, respectively.

Based on a few comparisons with active field performance, Kaneuji set

tentative borderlines for classifying coarse aggregate as durable, nondurable, or marginal. Lindgren (1980) attempted to expand this correlation by testing aggregates taken from 52 highway cores and 5 rock samples. The approach involved determining an average EDF value for aggregate extracted from the concrete cores. Determination of the average value involved selecting representative samples of the various aggregates within a concrete core, determining the EDF for each representative aggregate, determining the percentage of each representative aggregate present, selecting a threshold between durable and nondurable aggregates, and finally selecting what percentage of nondurable aggregate present in the core is detrimental. Nondurable cores were defined as having an EDF value of 50 or less for more than ten percent of the aggregate. Although the author claimed good correlation between serviceability ratings and average durability factor, closer examination of the data indicates 10 of the 52 cores evaluated would be incorrectly evaluated as durable using this approach and four incorrectly evaluated as nondurable.

While the assumption made by these authors that deterioration of portland cement concrete pavements in temperate climates is due largely to the coarse aggregate in the concrete is not a bad one, examination of the pore structure of the coarse aggregate alone precludes the possibility that the deterioration might be due to interactions between the mortar and the aggregate and other factors. Also, with the expanding use of new admixtures, new types of portland cement, and cement substitutes in concretes being placed today, assuming coarse aggregate to

be the sole source of freeze-thaw deterioration in the future is questionable. The ability to evaluate the pore size distribution of a concrete mixture and subsequent changes in this distribution when the concrete is exposed to a set of simulated or actual environmental conditions should be a valuable aid in predicting the serviceability of concrete structures made using that mixture. Available porosimetry methods are incapable of doing this due to their inability to determine pore size distribution of samples large enough to be considered representative of a highly heterogeneous material such as portland cement concrete. The porosimeter introduced in this dissertation, however, is capable of testing large samples, as will be shown later in this document.

Diamond (1970) and Sridharan et al. (1971) first demonstrated that mercury porosimetry could be used to determine the pore size distributions of soils and provide a better understanding of engineering modifications of soil such as compaction. The interpretation of pore size distributions of soil by conventional techniques is hindered by the same limitations mentioned earlier for portland cement concrete. Specifically, the inability to test representative size samples and the effects of sample preparation make interpretation of results difficult. Diamond (1970) infers this when he states that the development of techniques for water removal which do not involve changes in pore structure are necessary and suggests that for weak saturated clays, critical-region drying or freeze-drying may be practical.

In summary, the advent of conductometric phase transition

porosimetry promises significant improvements over conventional porosimetry techniques for the determination of the pore size distribution of civil engineering materials. These include minimal pore structure disturbance due to sample preparation and the ability to determine the pore size distributions of samples much larger than by other methods, thus minimizing sampling errors.

Capillary Freezing and Melting

Two models have been proposed to explain the mechanism of freezing and thawing in capillaries. They are the homogeneous nucleation model and the plastic ice model.

For the homogeneous nucleation model, it is assumed that ice crystals are formed in capillaries in the same manner as in the freezing of bulk water. Crystal growth continues until a capillary wall obstructs the growth. The remaining capillary space is occupied by unfrozen pore water or solution. The theory is consistent with the observed relationship between temperature and capillary freezing in that the smaller the capillary, the lower the freezing temperature. Enüstün et al. (1965) explored the validity of the homogeneous nucleation model experimentally by investigating the electrical conductance of a sample of porous glass saturated with ammonium nitrate solution at temperatures below the normal freezing point and found freezing points predicted by this model to be inaccurate. In discussing this theory, Enüstün et al. (1965) and Hesstvedt (1964) point out that freezing by homogeneous nucleation is a rate process beyond the treatment of equilibrium

thermodynamics and Enüstün later rejects the model in favor of the plastic ice model first rigorously developed by Everett (1961). Continued investigations (Enüstün et al., 1978; 1985a; 1985b) of capillary freezing and melting led to the confirmation of the plastic ice model and subsequently to the development of phase transition porosimetry discussed earlier in this dissertation.

Early development of the plastic ice model can be traced to research done during investigations of frost action in soils by several researchers. Penner (1957), while investigating soil moisture tension and ice segregation, concluded that moisture tensions developed as a result of freezing point depressions and that higher tensions developed in soils with small pores than in soils with large pores because the freezing point decreases with the radius of curvature of the ice/water interface. Penner (1959) and Miller et al. (1960) pointed out that equilibrium thermodynamics could be used to provide a relationship between the freezing point in capillaries and the suction pressure developed in these capillaries. These studies led to the work of Everett (1961) and Everett and Haynes (1965) who developed a rigorous equilibrium thermodynamics formula for capillary freezing and melting. From this beginning, the plastic ice model has been extensively utilized in investigations of frost action in soils.

The plastic ice model relationship between capillary freezing point and capillary size can be developed from fundamental equilibrium thermodynamic principles as follows.

Consider two cylinders, each closed by a piston and joined by a

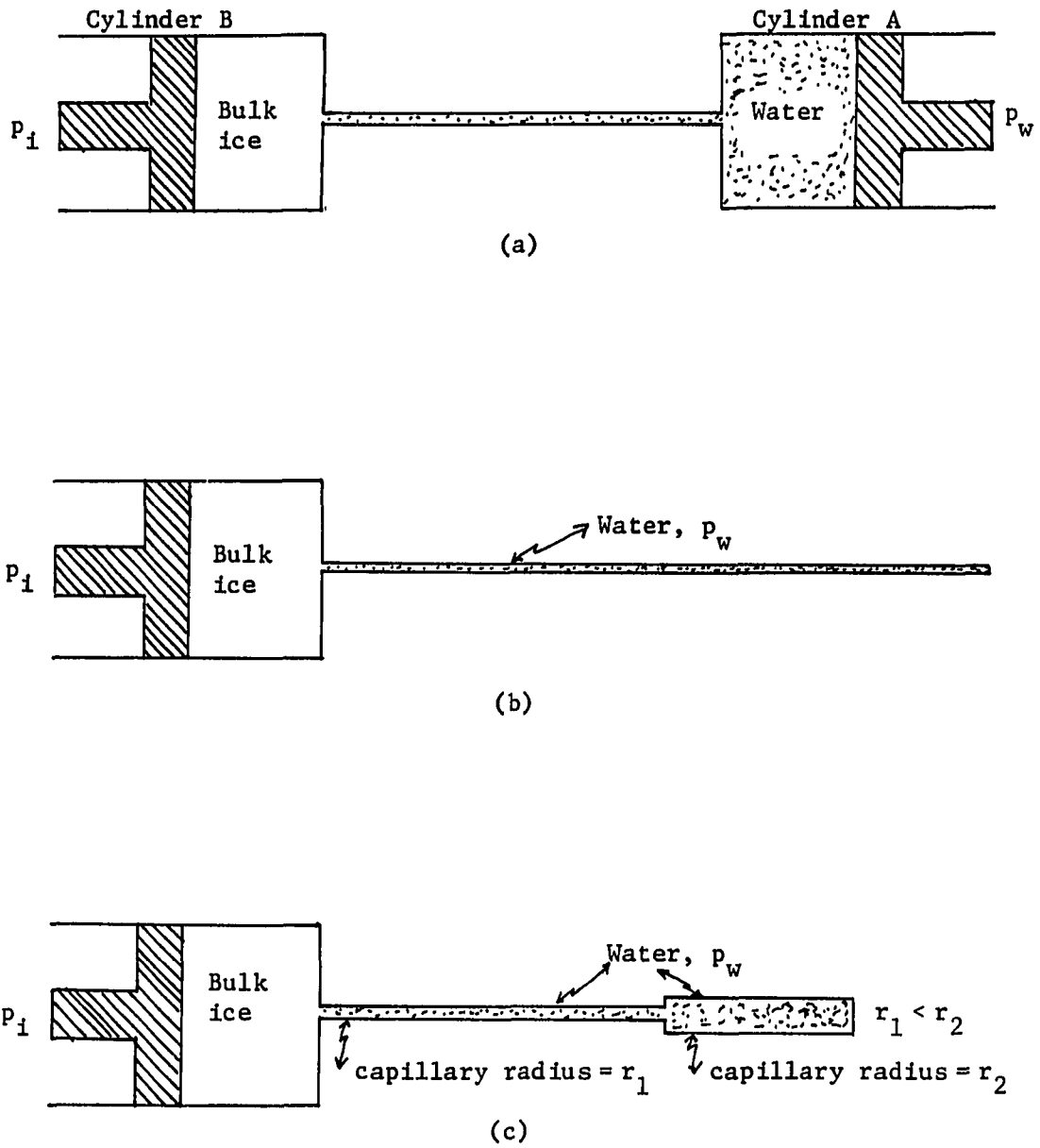


Figure 4. Schematics of capillary freezing and melting

capillary tube (see Figure 4a). If the temperature in both of the cylinders is lowered below 0°C and freezing is nucleated in cylinder B only, the accompanying expansion is taken up by the movement of the pistons. The well-known Laplace equation gives a relationship for the pressure difference across a meniscus and can be expressed as:

$$p_i - p_w = \frac{2\gamma_{i,w}}{r_{i,w}} \quad (4)$$

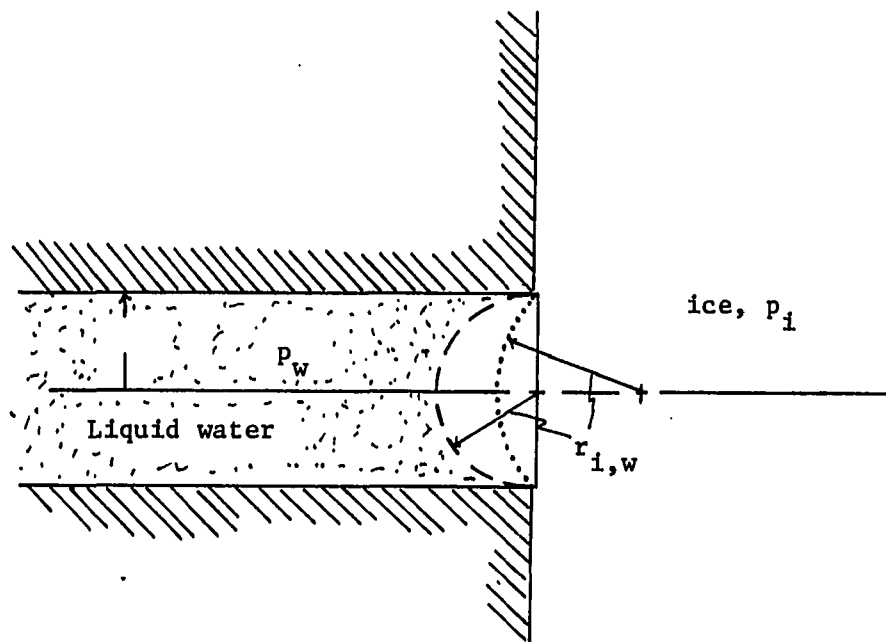
where p_i is the pressure on the ice side of the meniscus, p_w is the pressure on the liquid water side of the meniscus, $\gamma_{i,w}$ is the ice/water interfacial tension, and $r_{i,w}$ is the radius of curvature of the spherical meniscus. As long as the pressures exerted by the pistons are equal to those of equilibrium, the ice/water interface will be a plane and there will be no tendency for ice to penetrate the capillary (see Figure 5). Also, since the chemical potential of the liquid water will be greater than that of the ice, the growth of the bulk ice crystal will continue until all the liquid water is withdrawn from cylinder A. With soils, this type of behavior can occur and is the driving mechanism behind frost heave. The chemical potential of the liquid water, $\mu_w(T, p_w)$, at temperature T ($^{\circ}\text{K}$) and pressure, p_w , can be expressed as:

$$\mu_w(T, p_w) = \mu_w^{\circ}(T_0, p_0) - \int_{T_0}^T S_w^{\circ} dT + \int_{p_0}^{p_w} V_w^{\circ} dp \quad (5)$$

and the chemical potential of the ice $\mu_i(T, p_i)$ as:

$$\mu_i(T, p_i) = \mu_i^{\circ}(T_0, p_0) - \int_{T_0}^T S_i^{\circ} dT + \int_{p_0}^{p_i} V_i^{\circ} dp \quad (6)$$

where:



- $p_i = p_w \quad r_{i,w} = \infty$
- $p_i - p_w < 2\gamma/r_{i,w} \quad r_{i,w} > r$
- $p_i - p_w = 2\gamma/r_{i,w} \quad r_{i,w} = r$
ice will advance into capillary

Figure 5. Equilibrium interface between ice and water

T_0 = melting point of ice in $^{\circ}\text{K}$;
 p_0 = atmospheric pressure;
 $\mu_w^{\circ}, \mu_i^{\circ}$ = chemical potentials of liquid water and ice at T_0 and p_0 ,
 respectively;
 p_w, p_i = pressures of liquid water and ice, respectively;
 S_w°, S_i° = molar entropies of liquid water and ice at T_0 and p_0 ,
 respectively; and
 V_w°, V_i° = molar volumes of liquid water and ice at T_0 and p_0 ,
 respectively.

If the capillary is close-ended (see Figure 4b) and the pressure of the ice remains at p_0 due to movement of the piston as the temperature of the system is lowered, the transfer of liquid water to ice lowers p_w below p_0 ; thus, the radius of curvature of the ice/water interface will increase in accordance with the Laplace equation (see Figure 5), and the pressure in the liquid will be $p_0 - (2\gamma/r_{i,w})$. Realizing that $\mu_i^{\circ} = \mu_w^{\circ}$ and that at equilibrium $\mu_i = \mu_w$, equations 4, 5, and 6 can be combined as:

$$- \int_{T_0}^T S_i dT = - \int_{T_0}^T S_w^{\circ} dT + \int_{p_0}^{p_0 - (2\gamma/r_{i,w})} V_w^{\circ} dp \quad (7)$$

Neglecting the pressure dependence of V_w° , equation 7 can be rewritten as:

$$2 V_w^{\circ}/r_{i,w} = \int_{T_0}^T (S_w^{\circ} - S_i^{\circ}) dT \quad (8)$$

Also, from thermodynamics,

$$s_w^0 - s_i^0 = (L^0/T_0) + \int_{T_0}^T ((C_{pw} - C_{pi})/T) dT \quad , \quad (9)$$

where L^0 is the normal heat of fusion of ice at T_0 and p_0 , and C_{pw} and C_{pi} are the molar heat capacities at constant pressure of liquid water and ice, respectively. Combining equations 8 and 9 results in:

$$2 V_w^0 / r_{i,w} = L^0 (T_0 - T) / T_0 - \int_{T_0}^T \int_{T_0}^T ((C_{pw} - C_{pi}) / T) dT dT \quad . \quad (10)$$

Enüstün et al. (1965) pointed out that the double integral portion of this equality can be considered to be negligible. Finally, realizing that the density of water, ρ_w , is equal to the ratio of its molecular weight to its molar volume, V^0 , and that the heat of fusion of ice per unit mass, λ , is equal to the molar heat of fusion, L^0 , divided by the molecular weight of ice, equation 10 can be reduced to:

$$r_{i,w} = \frac{2\gamma T_0}{\rho_w \lambda (T_0 - T)} \quad (11)$$

Equation 11 correlates the equilibrium temperature, T , with the radius of curvature of the ice/water interface, $r_{i,w}$, in a capillary. When the radius of curvature of the meniscus equals the radius of curvature of the capillary, r (see Figure 5), the water in the capillary will freeze.

Figure 4c shows a bottle-necked capillary. During capillary freezing, the phase transition in the capillary will be controlled by the radius of the pore constriction as the ice cannot intrude into the larger pore body until freezing is initiated in the pore constriction. During

melting, however, the constriction will melt first in accordance with the plastic ice model as will the body, but at a higher temperature.

Hesstvedt et al. (1964) gave the following relationship between the ice/water interfacial tension, γ , and temperature where:

$$\gamma = \gamma_0 + k (T - T_0) \quad (12)$$

where k is a dimensional constant with a value of $0.25 \text{ ergs cm}^{-1} \text{ K}^{-1}$ and γ_0 is the value of γ at T_0 . Hesstvedt also gives a value for γ_0 of 29 ergs cm^{-2} .

For conductometric phase transition porosimetry, it is helpful to develop a relationship between pore radius, r and the negative inverse of the absolute temperature, X . With this information and substituting the relationship for ice/water interfacial tension given in equation 12, equation 11 can be rewritten as:

$$r = \frac{-2T_0 (k - \gamma_0 X + kT_0 X)}{\rho_w \lambda (1 + T_0 X)} \quad (13)$$

where r , T_0 , γ_0 , k , ρ_w , and λ are as defined earlier and $X = (-1/T)$, where T is the phase transition temperature in $^{\circ}\text{K}$.

The numerical values of the various constants used in the computations contained in this dissertation are as follows:

$$\begin{aligned} T_0 &= 273.16 \text{ }^{\circ}\text{K}; \\ \rho_w &= 1.000 \text{ gm cm}^{-3}; \text{ and} \\ \lambda &= 333.3 \text{ J gm}^{-1}. \end{aligned}$$

These values can be obtained from standard physical and chemical handbooks.

A possible concern with applying plastic ice model relationships to conductometric phase transition porosimetry is the effect of solute concentration in pore water upon capillary freezing and melting. It is well-known that solute concentration affects the freezing point of bulk pore solution. It may also affect the ice/water interfacial tension. Enüstün et al. (1978) investigated these effects of solute concentration on capillary freezing of four porous Vycor glass samples saturated with distilled water, and 0.02, 0.04, and 0.06 molar ammonium nitrate solutions. They monitored the variation of electrical conductance with temperature during a cycle of capillary freezing and thawing for each of the four samples and from examinations of inflection points of all curves concluded that the presence of solute had no significant effect on the location of inflection points and thus no significant effect on capillary freezing and thawing. These experiments were limited to solutions with relatively low solute concentrations saturating a porous material having a relatively small (approximate modal pore body size = 45 angstroms) and narrow pore size distribution. The effect of solute concentration on the freezing point of solutions in bulk and capillary pores saturated with the solution can be evaluated using fundamental thermodynamic relationships. In the derivation of the plastic ice model equation relating pore freezing temperatures and pore radii (equation 11), it was stated that at equilibrium the chemical potential of water, μ_w , would equal the chemical potential of ice, μ_i . For a nonvolatile solute that does not form a solid solution with a freezing solvent, at the freezing point the chemical potentials of the pure frozen solvent and the

contaminated liquid are equal. The resultant equilibrium relationship can be expressed as (Atkins, 1982):

$$\mu_i = \mu_w^* = \mu_w + RT \ln(x_a) \quad , \quad (14)$$

where μ_w^* is the chemical potential of the pore solution, R is the gas constant, and x_a is mole fraction of solvent in the pore solution. Following the same steps presented earlier results in an equation which relates pore freezing temperatures, pore radii, and solute concentration. This relationship can be expressed as:

$$r = \frac{2\gamma T_o}{\rho_w \lambda (T_o - T) - (\rho_w T_o / M) RT \ln(x_a)} \quad . \quad (15)$$

Column 3 of Table 1 lists values of r calculated for several different values of x_a and T using equation 15. Concentrations selected are .02, .06, .1 and 1 molar ammonium nitrate solutions. It can be seen from the examination of the values listed in Table 1 that for the solute concentrations and pore sizes investigated by Enüstün et al. (1978) (.02 to .06 molar), the effect of the solute is negligible as they had observed experimentally. However, for increased solute concentrations and especially for larger pore sizes, this is not the case. For example, at a temperature of -1.0°C , pure water in pores with radii larger than 471 angstroms will freeze. If the pores are saturated with a .1 molar solution of ammonium nitrate, only solution in pores larger than 578 angstroms will freeze. Equation 14 provides the relationship necessary to calculate pore radii from pore freezing points if the solute

concentration is known. At concentrations and temperatures for which no pore radius is indicated in column 3 of Table 1, no freezing will occur as these temperatures are above the freezing point of the bulk solution.

Another relationship can be used to calculate pore radii from pore solution freezing points if the bulk solution freezing point is known. The relationship is nearly the same as the one expressed in equation 11, except the bulk solution freezing point, T_o^* , is substituted for the normal freezing point of water, T_o . The appended relationship is:

$$r_{i,w} = \frac{2\gamma T_o^*}{\rho_w \lambda (T_o^* - T)} \quad (16)$$

Column 4 of Table 1 gives values for r calculated for several different values of T_o^* and T . The values of T_o^* selected are those equal to the bulk solution freezing points of .02, .06, .1, and 1 molar ammonium nitrate solutions. These values were calculated using a relationship given by Atkins (1982), where

$$T_o - T_o^* = (R (T_o)^2 / H_{fus}) x_b \quad (17)$$

where H_{fus} is the molar enthalpy of fusion for the solvent and x_b is the mole fraction of the solute. Values for T_o^* calculated using equation 16 and a value of $6.008 \text{ kJ mol}^{-1}$ for the molar enthalpy of fusion of water are given in Table 2. Comparative examination of the data listed in columns 3 and 4 of Table 1 indicate the relationships expressed in equations 15 and 16 give the same results.

At this point, another advantage phase transition porosimetry

Table 1. Effect of solute concentration on calculation of pore radii from pore solution freezing points for ammonium nitrate solution

Molarity	Temperature, °C	Pore radius, angstroms	
		Equation 14	Equation 15
0	- 0.10	4749.4	4749.4
0	- 0.50	788.1	788.1
0	- 1.00	471.2	471.2
0	-10.00	43.44	43.44
0	-20.00	19.67	19.67
0	-30.00	11.75	11.75
.02	- 0.10	7565.1	7564.1
.02	- 0.50	840.2	840.4
.02	- 1.00	489.4	489.5
.02	-10.00	43.59	43.61
.02	-20.00	19.70	19.71
.02	-30.00	11.76	11.76
.06	- 0.10	--	--
.06	- 0.50	967.8	968.6
.06	- 1.00	530.2	530.7
.06	-10.00	43.91	43.95
.06	-20.00	19.77	19.79
.06	-30.00	11.78	11.80
.1	- 0.10	--	--
.1	- 0.50	1141.1	1142.2
.1	- 1.00	578.4	579.2
.1	-10.00	44.23	44.31
.1	-20.00	19.84	19.88
.1	-30.00	11.81	11.81
1.0	- 0.10	--	--
1.0	- 0.50	--	--
1.0	- 1.00	--	--
1.0	-10.00	52.82	53.69
1.0	-20.00	21.51	21.91
1.0	-30.00	12.42	12.69

Table 2-3. Variation of bulk solution freezing point with solute concentration for ammonium nitrate solutions

Molarity	Bulk solution freezing point (°C)
0.00	0.0000
0.02	-0.0371
0.06	-0.1114
0.10	-0.1855
1.00	-1.8528

methods have over mercury porosimetry can be illustrated. The pressure differential across the meniscus of intruding mercury for a pore with a radius of curvature equal to 18 angstroms is approximately 60,000 psi. Due to the much lower interfacial tension across an ice/water meniscus, the pressure differential across the meniscus of intruding ice for a pore with a radius of curvature of 18 angstroms (-21.5 °C) is approximately 3800 psi. Thus, mechanical sample degradation due to induced pressures during the determination of pore size distributions will be much less for phase transition porosimetry methods than for mercury porosimetry. However, viscous forces that develop due to flow of ice are higher than those due to flow of mercury because of the higher viscosity of ice. Therefore, rapid freezing and repeated freezing and thawing cycles may cause destruction of samples in phase transition porosimetry.

It is pertinent at this point to discuss some limitations of the plastic ice model. Theoretically, the model provides a relationship for evaluating the size of infinitesimally small pores. However, for pores with extremely small radii, interactions between the solid porous

material and the liquid solution may cause the radius of the penetrating ice to differ significantly from the radius of the penetrated pore. Further, as the size of a pore begins to approach the size of water molecules, the formation of a meniscus and thus applicability of the Laplace equation becomes questionable. The size of pores at which these effects become appreciable is debatable. However, the calculation of pore sizes from their freezing points assuming these effects to be negligible still provides valuable information about the pore structure of materials with extremely small pores. Researchers should bear in mind that the calculated pore radii for these small pores may provide only an index of the size of the pore which, although directly related to the pore size, may not be equal to the actual pore size.

Principles of Electrical Conductance of Porous Materials

Many porous materials consist of a nonconducting solid phase and a void system of randomly intersecting capillaries of various sizes. If the material is saturated, the pore solution generally will exhibit electrolytic conductance, which results from the mobility of ions in the solution.

Archie (1942), Winsauer et al. (1951), and Perkins et al. (1956) used a parameter called resistivity factor in their investigations of the resistivity of porous geomaterials. Consider a cylindrical container of volume V having a cross-sectional area, A , and length, L , made of an insulating material, but with ends of a conducting material and filled with a solution of conductivity, κ . The conductance, C , of the container

is then

$$C = \kappa(A/L) = \kappa(V/L^2) \quad . \quad (18)$$

Let the same container be filled with a porous material saturated with the same solution. The conductance will now be considerably less since the mean length of path traversed by the current is longer and the cross-sectional area which is available to current flow is smaller.

Previous researchers defined the conductance of porous material, C' , as

$$C' = \kappa(A'/L') \quad (19)$$

where L' is the effective tortuous path length traversed by the electrical current and A' is the effective conducting cross-sectional area of the material available to current flow. The resistivity factor, F , by definition is equal to C/C' .

Winsauer et al. (1951) and later Perkins et al. (1956) expressed a relationship between resistivity factor, F , and pore structure where

$$F = \tau/\psi \quad , \quad (20)$$

where τ is the tortuosity, defined as the ratio of L' to L , and ψ is the ratio of A' to A . Also, these investigators related the resistivity factor to the tortuosity, τ , and porosity, η , of the sample by

$$F = \frac{\tau^2}{\eta} \quad . \quad (21)$$

In a commentary of the paper by Perkins et al. (1956), Wyllie and de Witte (1956) pointed out that this is true only if porosity equaled tortuosity times ψ , or otherwise stated, the pore volume, V_p , was equal

to the product of A' and L' , which is only true for pores with uniform cross-sectional areas.

This can be shown by considering a pore system consisting of one pore with a nonuniform cross-sectional area. If the pore is a right angle truncated cone, a relationship between the conductance of the pore and the pore geometry can be developed as follows. Figure 6 illustrates the geometry of such a cone. The resistance of an infinitesimally small section of the cone, dR , can be expressed as

$$dR = \frac{\rho dl}{\pi r^2} = \frac{\rho}{\pi} \frac{dl}{(\ell \tan\theta + r_o)^2} \quad (22)$$

where dl , r , r_o , and θ are as defined in Figure 6 and ρ is the resistivity of the pore solution. Integrating over the length of the pore, L , gives the following relationship,

$$R_c = \frac{\rho}{\pi} \int_0^L \frac{dl}{(\ell \tan\theta + r_o)^2} \quad (23)$$

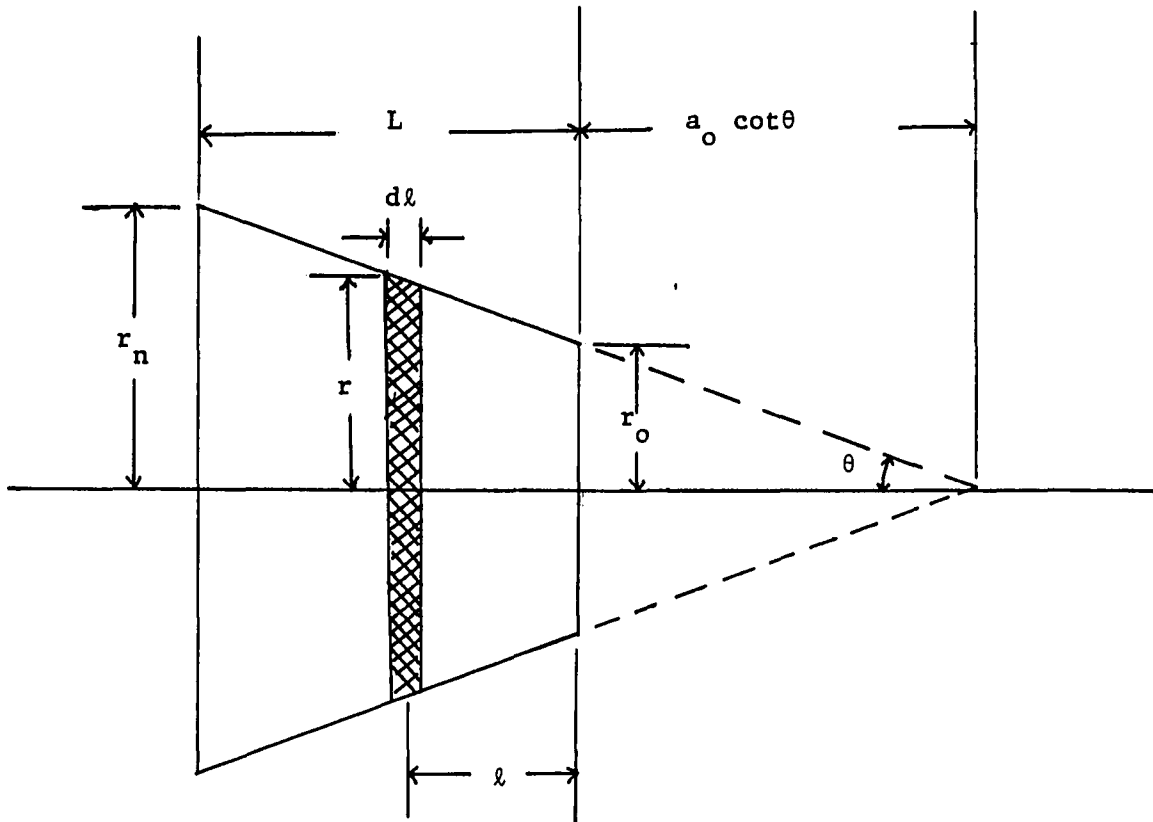
where R_c is the resistance of the cone. Realizing that $\tan \theta$ equals the quantity, $(r_n - r_o)/L$, and solving the integral yields

$$R_c = \frac{\rho}{\pi} \frac{L}{r_o r_n} \quad (24)$$

Since conductance, C , is the reciprocal of resistance, R and conductivity, κ , is the reciprocal of resistivity, ρ , equation 24 can be rewritten as

$$C_c = \frac{\kappa(\pi r_n r_o)}{L} \quad (25)$$

where C_c is the conductance of the cone. Examination of equation 25 with respect to equation 18 indicates that for a pore system consisting of a



$$r = (l + r_o \cot \theta) \tan \theta = l \tan \theta + r_o$$

Figure 6. Geometry of right angle truncated cone

single right angle truncated cone, A' must equal $\pi r_n r_o$ and L' must equal L . Therefore, in order for the relationship expressed in equation 21 to be correct, the volume of the pore must be equal to $L\pi r_n r_o$. Mensuration tables give the volume of a right angle truncated cone, V_c , as

$$V_c = L \frac{\pi}{3} r_o r_n \left(\frac{r_o}{r_n} + 1 + \frac{r_n}{r_o} \right) . \quad (26)$$

From this, it is clear that the volume of the cone will equal $L\pi r_n r_o$ only when $r_n = r_o$, or otherwise stated, only when the pore has a uniform cross-sectional area.

For conductometric phase transition porosimetry, the conductance of a porous material, C' , is redefined as

$$C' = \kappa V_p / (L^2 \omega) \quad (27a)$$

where κ and L are as defined earlier (equation 18), V_p is the pore volume and ω is a dimensionless pore geometry factor. At this point, some relationships between pore geometry factor and pore geometry will be developed to give some physical significance to this new factor.

An expression for the conductance of a pore system consisting of x number of parallel conducting pores can be developed from equation 27a and can be expressed as

$$C' = \kappa \frac{V_p}{\omega L^2} = \kappa \sum_{j=1}^{j=x} \frac{V_{pj}}{L^2 \omega_j} , \quad (27b)$$

where C' , κ , V_p , L , and ω are as defined earlier and V_{pj} and ω_j are the pore volume and pore geometry factor for individual conducting pore j . From this relationship, it can be seen that if the pore geometry factor for each parallel conducting pore, ω_j , is the same, then the pore

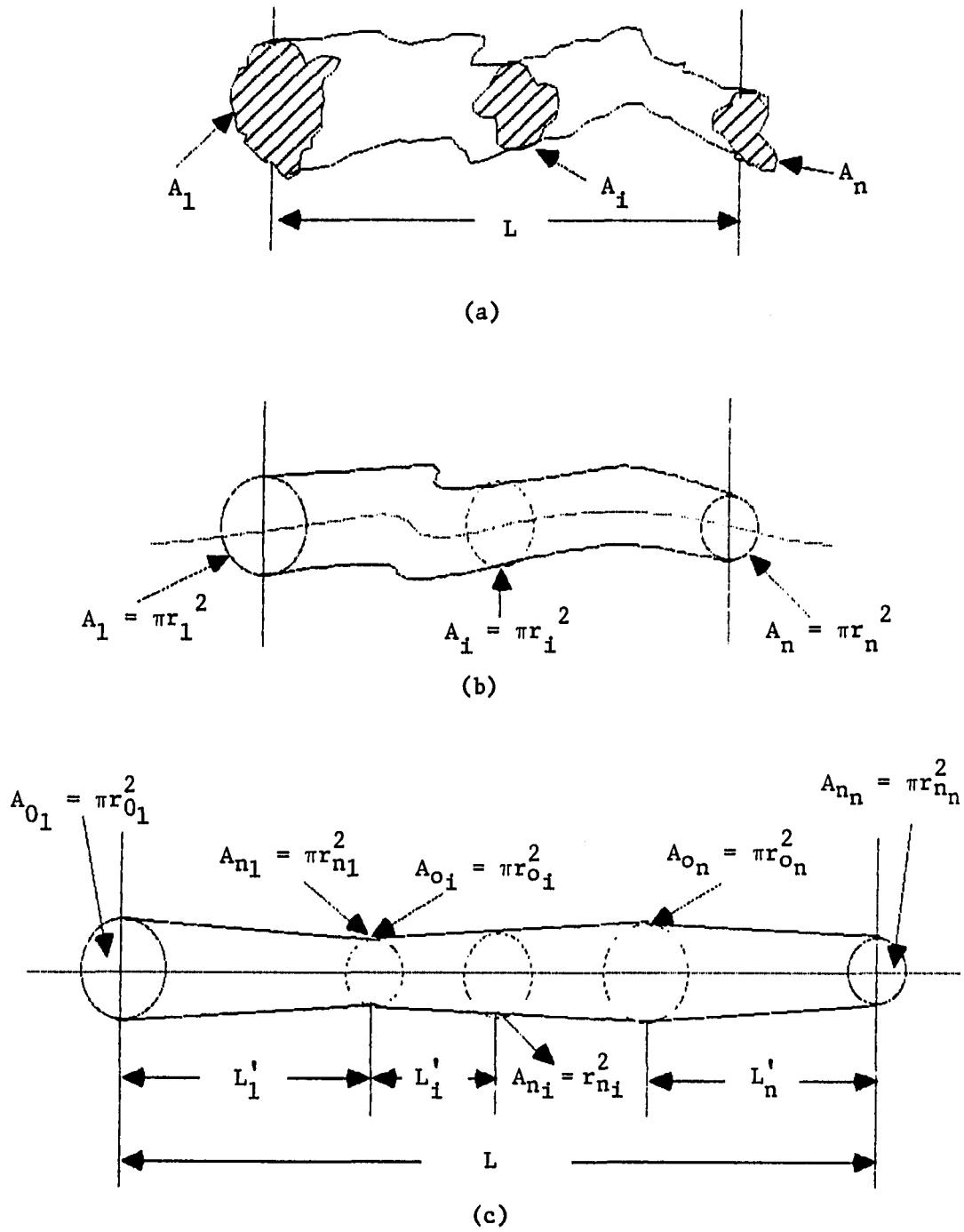


Figure 7. Right angle truncated cone modeling of irregular pore

geometry factor for the pore system, ω , is equal to ω_j .

Let us now consider an irregular single conducting pore channel of a porous material as shown in Figure 7a. This pore can be represented by a pore with equivalent circular cross-sectional areas as shown in Figure 7b. Such a pore would have the same volume and conductance as the original irregular pore. If this circular pore is then stretched so the curve connecting the centers of the circular cross-sections is a straight line, the pore can be represented by a series of right angle truncated cones as shown in Figure 7c without a change in the volume or conductance.

By combining equations 25, 26, and 27a, it can be shown that if the pore system of a porous material when stretched consists of a single right angle truncated cone, then the dimensionless pore geometry factor, ω , can be expressed as

$$\omega = \left(\frac{1}{3}\right)\left(\frac{L'}{L}\right)^2 \left(\frac{r_o}{r_n} + 1 + \frac{r_n}{r_o}\right) \quad (28)$$

Using the same approach, the pore geometry factor, ω , for a pore system consisting of a singular irregular pore (represented by z number of right angle truncated cones in a series connection, Figure 7) can be expressed as

$$\omega = \frac{\sum_{i=1}^{i=z} \left(L_i' r_{oi} r_{ni} \left(\left(\frac{r_{oi}}{r_{ni}} \right) + 1 + \left(\frac{r_{ni}}{r_{oi}} \right) \right) \right)}{3L^2} \left(\sum_{i=1}^{i=z} L_i' / (r_{oi} r_{ni}) \right) \quad (29)$$

The relationship expressed in equation 29 can be greatly simplified for a pore of uniform cross-sectional area. The simplified relationship can be

expressed as

$$\omega = (L'/L)^2 \quad (30)$$

where L' is the sum of the lengths of component cylinders (right angle truncated cones with uniform cross-sectional areas) and L is the distance between the ends of the conducting pores as previously defined. The quantity (L'/L) has been commonly referred to as tortuosity. Also, for a pore consisting of uniformly sized right angle truncated cones (r_{oi} and r_{ni} are constant), equation 29 can be simplified to the relationship expressed in equation 28. Further, if the connected cones lie in a straight line perpendicular to the end planes of the conducting material, L' will equal L and the pore geometry factor will be a function of the ratio of the pore neck radii to the pore body radii only. This ratio will be hereafter referred to as necking. Later in this dissertation, these two fundamental characteristics, tortuosity and necking, will be referred to in a qualitative sense in discussing the importance of the pore geometry factor to conductometric phase transition porosimetry.

Finally, considering any pore system to consist of x number of irregular conducting pore channels connected in parallel, an expression for the pore geometry factor for any pore system as modeled by an agglomeration of right angle truncated cones is

$$\omega = \frac{\sum_{j=1}^{j=x} \left(\sum_{i=1}^{i=z} L_i' r_{oi} r_{ni} \left(\left(\frac{r_{oi}}{r_{ni}} \right) + 1 + \left(\frac{r_{ni}}{r_{oi}} \right) \right) \right)_j}{3L^2} \times \left(\sum_{j=1}^{j=x} \left(\left(\sum_{i=1}^{i=z} L_i' / (r_{oi} r_{ni}) \right)^{-1} \right)_j \right)^{-1} \quad (31)$$

Equation 31 provides a rather complicated expression for the pore geometry factor for any pore system. The relationship may be seen as largely academic, illustrating the need to be able to determine the pore size distribution of a material in order to characterize the pore structure of a material. A corrected version of equation 21 which relates the measurable parameter resistivity factor, F , to porosity, η , and pore structure can be written by combining equations 18 and 27 and expressed as

$$F = \omega/\eta \quad . \quad (32)$$

Examining this relationship in light of equation 31 illustrates that values for the pore geometry factor determined from singular conductance measurements and measurements of porosity are very difficult to interpret due to the interrelated effect of tortuosity and necking on pore geometry factor. If the parallel conducting pores within a system have equivalent pore geometry factors, the pore geometry factor for the system as well as the individual pores could be evaluated using the relationship given for a single conducting pore (equation 29). This would simplify matters somewhat, but interpretation would still be difficult. Pore systems for which this would be true would contain geometrically similar pores, or otherwise stated, all pores in the system would have the same tortuosity and necking. This does not imply that all the pores in the system are identical but rather that large pores are essentially magnifications of smaller pores.

CONDUCTOMETRIC PHASE TRANSITION POROSIMETRY

In conductometric phase transition porosimetry, it is assumed that the electronic conductance through the solid phase of a porous material is negligible when compared to the electrolytic conductance of the pore solution. Thus, when the temperature of a mass of saturated porous material is raised from sub-freezing temperatures, the frozen pore solution will melt and an increase in electrical conductance will be observed. Capillaries will begin to melt with smaller sizes melting at lower temperatures in accordance with the plastic ice theory.

The theory behind conductometric phase transition porosimetry is developed from a few basic relationships. First, it is assumed the relationship between the electrical conductivity of an electrolyte and the absolute temperature of the electrolyte is an Arrhenius type relationship and can be expressed as

$$\ln \kappa = \frac{-a}{T} + b, \quad (33)$$

where:

κ = electrical conductivity;

T = absolute temperature;

a = physical constant; and

b = physical constant.

Recalling the relationship between conductance and pore volume introduced earlier as equation 27 and combining the same relationship with equation 33 results in the following equation.

$$Y = mX + d + \log_{10} \left(\frac{V_p}{L^2 \omega} \right) \quad (34)$$

where L , V_p , and ω are as defined earlier; and

$X = -1/(\text{absolute temperature})$;

$Y = \log_{10} (\text{conductance})$;

$m = \text{physical constant}$; and

$d = \text{physical constant}$.

A plot of the log of the conductance versus the negative inverse of the absolute temperature (Y versus X) for which no phase change occurs will result in a line with a slope equal to m . Below freezing temperatures, the phase changes which occur in a certain range of pores will effectively decrease the volume of conducting pores. Thus, V_p at below freezing temperatures can be considered to be the volume of conducting pores, V_{cp} .

Taking the first derivative of equation 34 yields

$$\frac{dY}{dX} = m + \frac{d(\log_{10} (V_p / (L^2 \omega)))}{dX} \quad (35)$$

Integrating equation 35 over the definite interval, X to X_0 , yields the following relationship:

$$\log_{10} \left[\frac{\frac{V_{cp}(X_0)}{L^2 \omega_{X_0}}}{\frac{V_{cp}(X)}{L^2 \omega_X}} \right] = \int_X^{X_0} \frac{dY}{dX} dX - \int_X^{X_0} m dX \quad (36)$$

The right half of equation 36 can be calculated numerically from

conductance test data and is given the variable name Z . Also, if X_0 is the value of X at the pore solution melting point, then $V_{cp}(X_0)$ is simply the total conducting pore volume, V_p .

Figure 8 illustrates graphically the calculation of the parameter Z . Plotted on this graph is the first derivative of the log of the conductance with respect to the negative inverse of the absolute temperature versus the negative inverse of the absolute temperature. The data plotted were obtained from the warming cycle of conductometric phase transition testing of porous Vycor glass (see results and discussion). The integration of these data over finite intervals, as illustrated in Figure 8, gives the parameter Z as a function of the negative inverse of the absolute temperature.

Rearranging equation 36 and referring to pore volume by radius, we obtain

$$\alpha \frac{V_{cp}(r)}{V_p} = 10^{-Z(X, X_0)} \quad (37)$$

where $V_{cp}(r)$ is the volume of the pores with radii smaller than or equal to r containing unfrozen pore solution, V_p is the total pore volume and α is the ratio of the pore geometry factor of the total pore system, ω , to the pore geometry factor of the pores smaller than r , ω_r . The plastic ice model of capillary freezing and melting as expressed earlier in equation 13 provides the necessary relationship between the negative inverse of absolute temperature, X , and pore radius.

If it is assumed that the pore geometry factor, ω , is independent of pore size, then α is equal to unity and equation 37 can be reduced to

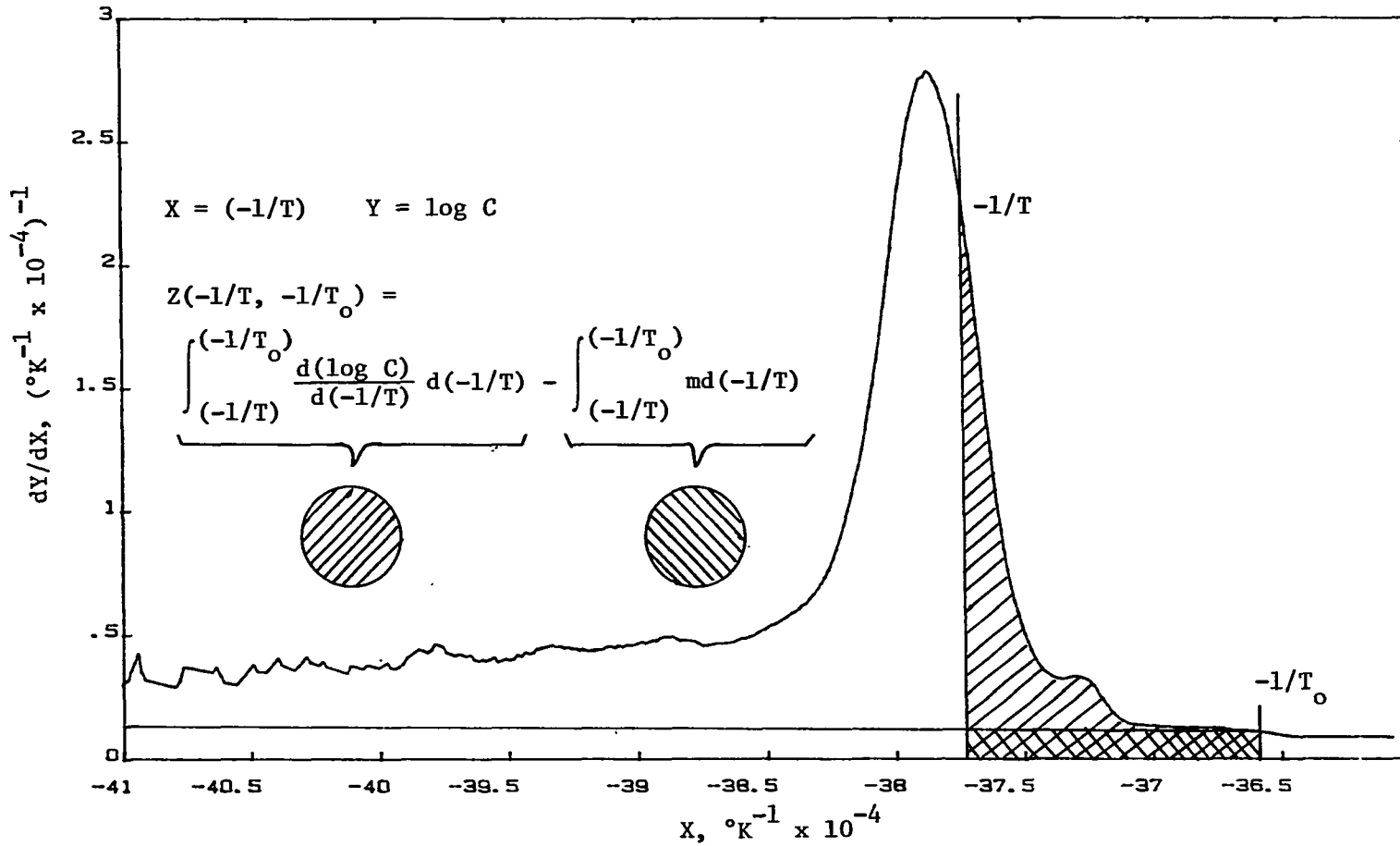


Figure 8. Illustration of the calculation of the parameter Z in conductance phase transition porosimetry

$$\frac{V_{cp}(r)}{V_p} = 10^{-Z(X, X_0)} \quad (38)$$

Assuming a to be equal to unity implies the pore system has a certain degree of homogeneity. This does not imply that all the conducting pores are exactly the same, but rather that the tortuosity and necking inherent to a given pore structure is uniform throughout the pore size distribution.

Examination of equation 37 indicates that the evaluation of the maximum and minimum pore sizes will be unaffected by variations in pore geometry factor. One can examine qualitatively the effect of varying pore geometry factors and such an examination follows.

The expression for the conductance of a pore system of x number of parallel conducting pores expressed in equation 27 can be reduced to

$$\frac{V_p}{\omega} = \sum_{j=1}^{j=x} \frac{V_{pj}}{\omega_j} \quad (39)$$

where V_{pj} and ω_j are the pore volume and pore geometry factor, respectively, for individual conducting pore j . Errors in the determination of pore size distributions of materials for which the pore geometry factor is dependent of pore size are somewhat reduced by the compensating effect of pore volume on the overall pore geometry factor of a pore system. For example, let's consider a pore system in which the pore geometry factor is a function of pore size with small pores having larger pore geometry factors. Pores containing larger volumes of pore solution will affect the overall pore geometry factor to a greater extent than pores containing smaller volumes of solution. Changing pore

geometry factors will affect the determined size distribution. It will be somewhat skewed towards the larger pores, as illustrated in Figures 9 and 10.

The plots in Figures 9 and 10 labeled actual size distribution are for a hypothetical pore system having a normalized Gaussian distribution with a standard deviation, σ , equal to 0.3 times the mean pore size, R . The pore geometry factor for a range of pore sizes was assumed to be inversely proportional to the pore radius. Calculations necessary to produce the curves labeled conductometry size distribution in Figures 9 and 10 were performed with the aid of a computer spreadsheet and the relationships expressed in equations 37 and 39. Examination of Figure 10 indicates the modal pore sizes as determined by conductometric phase transition porosimetry would be somewhat larger than the actual modal pore size. Table 4 illustrates the effect of standard deviation on this shift in modal pore size. Shown in this table are the errors in modal pore size expressed as a percentage of the actual pore size as calculated for various standard deviations. Also listed are the approximate ratios of the pore geometry factor for the largest pores in the distribution to the pore geometry factor for the smallest pores in the distribution. These ratios will hereafter be referred to as pore geometry factor ratios. The pore geometry factor ratios listed in Table 4 are the ratios of the pore geometry factor for a pore with a radius equal to the mean radius, R , minus three times the standard deviation, σ , to the pore geometry factor for a pore with a radius equal to the mean radius, R , plus three times the standard deviation, σ . Since pore geometry factors

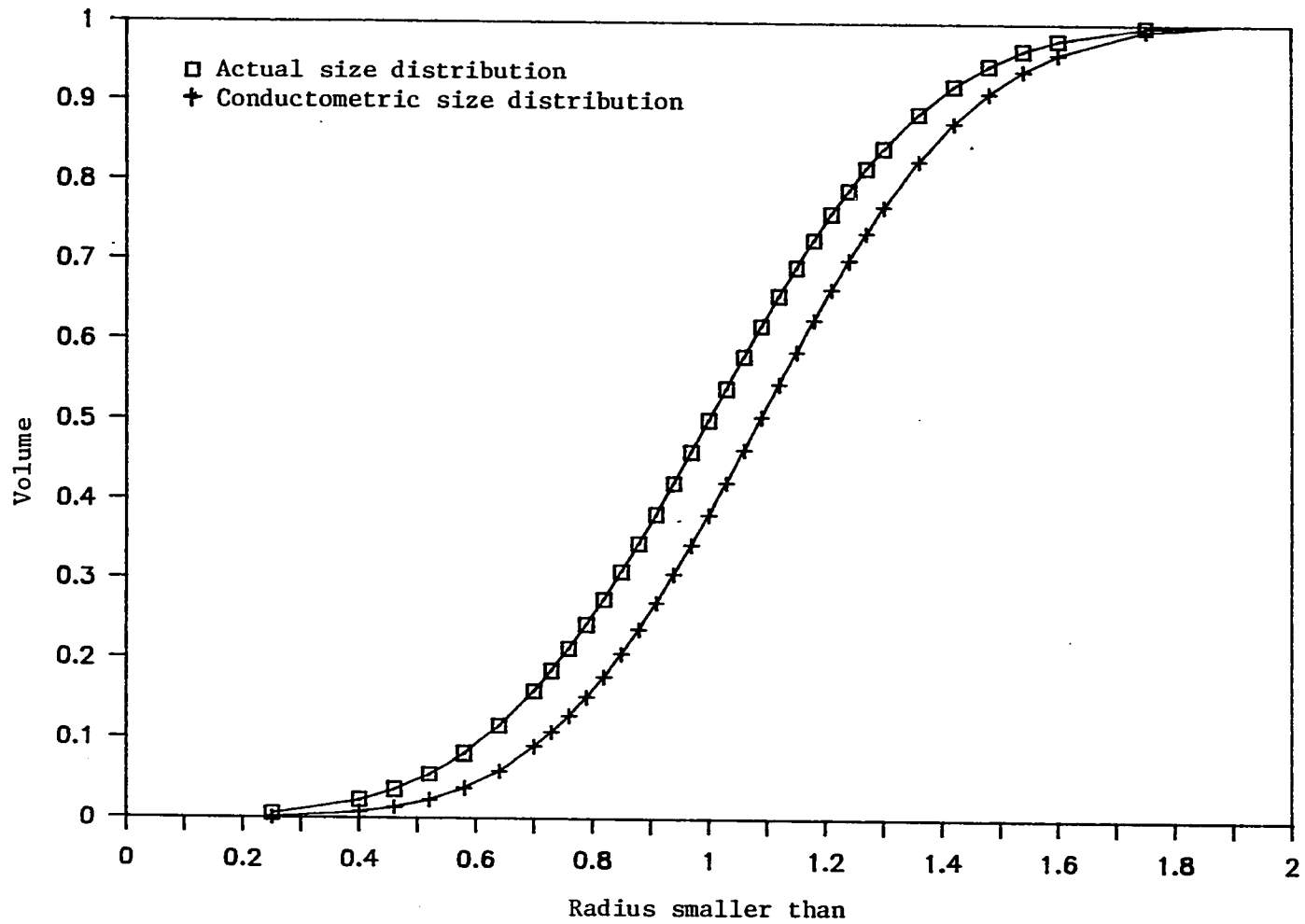


Figure 9. Illustration of error imposed by assuming that pore geometry factor is independent of pore size in CPTP

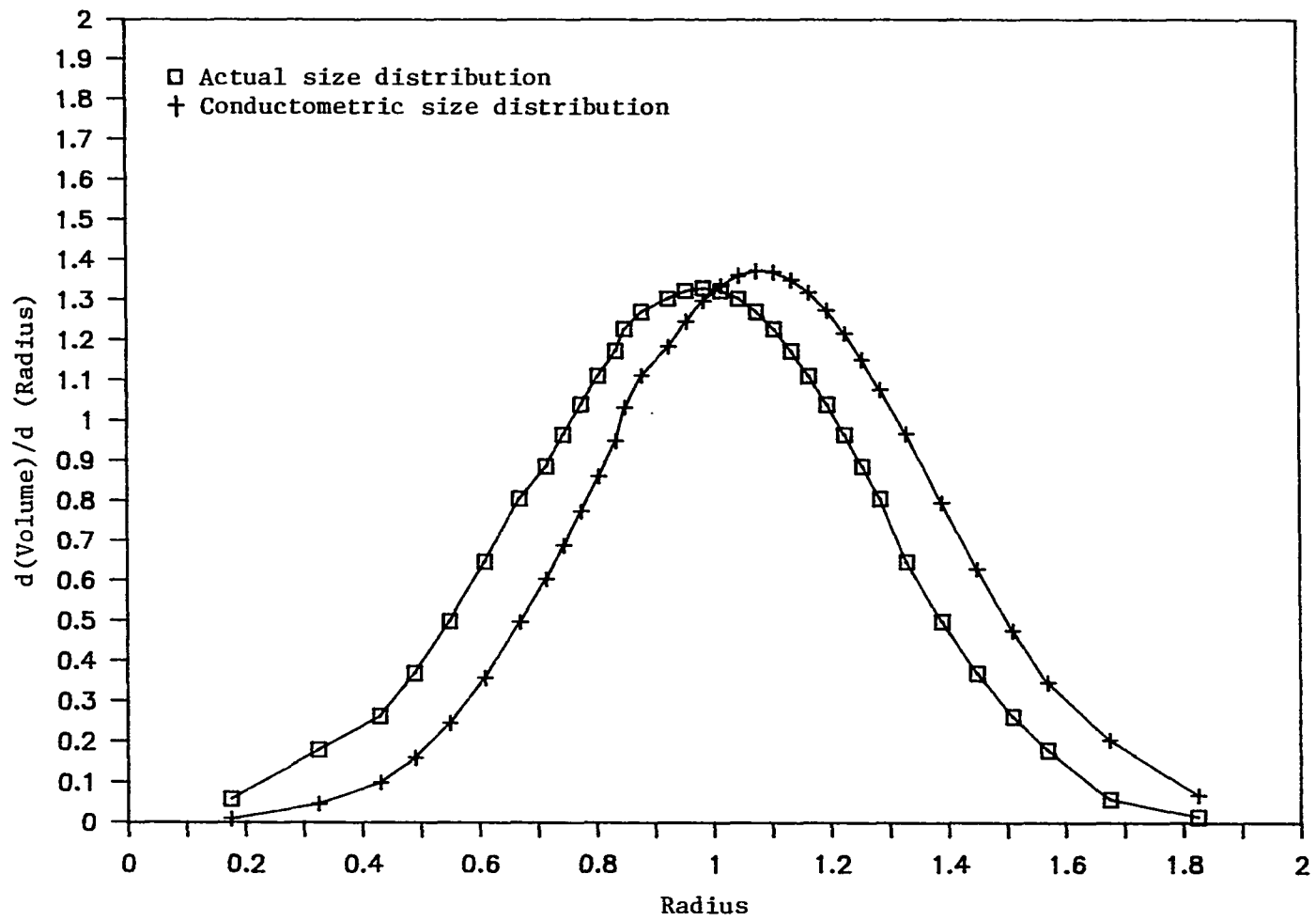


Figure 10. Illustration of error imposed by assuming that pore geometry factor is independent of pore size in CPTP

Table 4. Effect of standard deviation on shift in modal pore size when the pore geometry factor is inversely proportional to the pore radius

Standard deviation	Pore geometry factor ratio ^a	% Error in modal pore radius
0.3 R	19	10.5%
0.2 R	4	5.0%
0.1 R	1.9	0.5%

$$^a \text{Pore Geometry Factor Ratio} = \frac{\omega \text{ for pore with radius} = R - 3\sigma}{\omega \text{ for pore with radius} = R + 3\sigma}$$

Table 5. Effect of standard deviation on shift in modal pore size when the pore geometry factor is inversely proportional to the square of the radius

Standard deviation	Pore geometry factor ratio	% Error in modal pore radius
0.3 R	361	16.5%
0.2 R	16	9.0%
0.1 R	3.4	2.5%

were assumed to be proportional to the inverse of the pore radius, the pore geometry factor ratios listed in Table 4 are equal to the ratio of the radius of the largest pore to the radius of the smallest pore for the distribution. Table 5 gives the same information as Table 4 except pore geometry factor is assumed to be inversely proportional to the square of the radius. This results in larger pore geometry factor ratios. Notice that the error in modal pore size is related to both standard deviation and pore geometry factor ratios with the error increasing with increasing standard deviation and increasing pore geometry factor ratios. However, the error is bounded by the maximum pore size for a given size distribution; therefore, the effect of increasing pore geometry factor ratios diminishes as pore geometry factor ratios get larger. It seems conservative to state that errors that result from assuming pore geometry factor to be independent of pore size would be less than 20%. This would result in evaluating the modal size distribution of a pore system with an actual modal size of 100 angstroms as having a modal size distribution for 120 angstroms. For pores with uniform cross-sectional areas, the pore geometry factor of a pore is equal to the tortuosity of the pore; thus, the pore geometry factor ratio is the ratio of the conductive path length of the largest pore to the conductive path length of the smallest pore. From this, it also seems conservative to state that for many materials the error will be much less.

EXPERIMENTAL METHODS AND MATERIALS

Apparatus

The conductometric phase transition porosimeter consists of a conductance meter, a cryostat, a thermistor, the measurement circuitry, a microcomputer, and a plotter.

Conductance measurements are made with a Solomat 2009 conductivity meter. This meter's high frequency alternating current excitation and the use of stainless steel or tungsten electrodes minimize polarization effects. The instrument has a resolution of 0.1 micro-siemens (μS). Digitized conductance output from the meter is relayed to the microcomputer.

A thermistor manufactured by Thermometrics is used to measure the temperature of the sample. The thermistor is used as a variable resistor in a square wave oscillator. The oscillator's output frequency varies with corresponding changes in the resistance of the thermistor. The computer then measures the period of the square wave.

The temperature of the sample is controlled by placing the sample in a suitable container and immersing that container in a Haake cryostat filled with iso-propyl alcohol. The temperature of the cryostat is controlled by the computer through a digital to analog converter.

The computer used is an Apple IIe personal computer equipped with two input/output cards from John Bell Engineering. These cards serve as an interface between the measurement circuitry and the computer. Software was developed to allow the computer to control the temperature

of the cryostat, gather test data, perform necessary data processing, and output the processed data to either the computer's video monitor or a Hewlett-Packard plotter. Cycle starting and ending temperatures, cooling rates, and data collection intervals are all input by the user. In addition, the user can input the total number of cycles for a test sequence and designate different cycle parameters for each cycle within the sequence. These features allow the user to tailor the test sequence to a particular pore system. They also allow periodic determinations of pore size distribution to be made for samples subjected to repeated cycles of freezing and thawing. With this system, a wide range of ambient freezing and thawing conditions can be simulated.

The appendix of this dissertation contains a user's manual for the conductometric phase transition porosimeter. This manual contains instructions for operation of the porosimeter, porosimeter circuitry schematics, computer program descriptions, and computer program listings.

Experimental Procedure

Vycor glass experiment

The conductometric phase transition porosimeter was first used to test a Vycor glass sample obtained from the Corning Glass Company. This porous glass was selected because extensive pore size information is available for it (see Materials section) and thus comparative analysis with either technique is available. A 14-mm diameter and 4-mm thick disk was saturated with a 0.01 molar ammonium nitrate solution. By saturating the sample with this solution, measurable conductance was provided with

negligible freezing point depression.

For many materials, sample preparation necessary for this type of porosimetry will be minimal. It is simply a matter of embedding electrodes in a mass of the material. However, for the brittle Vycor glass, a more complex electrode configuration and containment apparatus was devised (see Figure 11). It consisted of a large glass test tube and two smaller sections of glass tubing. The Vycor disk was press fitted into one end of the larger tubing with a short piece of tygon tubing serving as a gasket. The smaller tubing was located at the opposite end. This tubing provided the inlet for the electrode connections and the thermistor. The upper end of it was filled with a desiccant and cotton to prohibit condensation inside the tube. The electrical contacts with the Vycor disk were provided by mercury placed inside the tube and tubing in contact with the top and bottom surfaces of the Vycor disk. Tungsten wires provided the electrical connections between the mercury contacts and the conductance meter. Finally, a thermistor was placed in the mercury immediately above the Vycor sample to provide necessary temperature measurements.

The assembled apparatus was then placed in the cryostat and the testing cycle was begun. The temperature of the bath was lowered from 7°C to -30°C at the rate of 3°C per hour. Conductance and temperature measurements were taken at 0.1°C intervals and these data composed the cooling portion of the test. The temperature is then raised to 7°C at the same rate. Again, conductance and temperature measurements were taken at 0.1°C intervals and these data composed the warming portion of

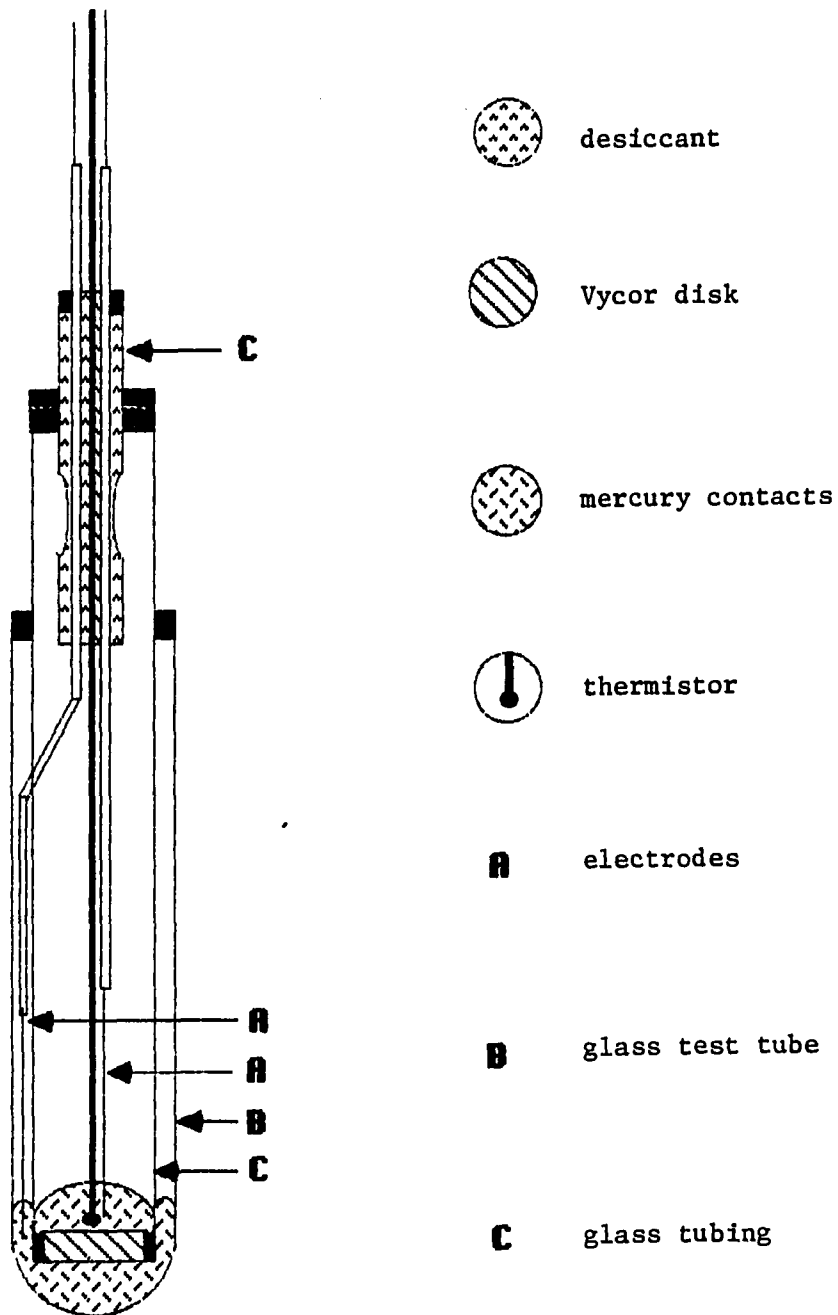


Figure 11. Electrode configuration and containment apparatus for Vycor glass sample

the test.

After some initial experimentation, a test run was made with a 5 kilo-ohm resistor connected in series with the Vycor sample. This allowed the test to be run in the more sensitive conductance range available on the conductance meter (0.1 μS resolution). Experimentally obtained values of conductance were then converted to quantitative conductance values using known relationships for conductors connected in series.

Portland cement mortar and concrete experiments

For the evaluation of pore size distributions of portland cement mortar and concrete, 4-inch diameter by 4-inch high cylinders were molded with an imbedded electrode configuration as shown in Figure 12. The electrodes are fixed in a Plexiglass bottom plate during molding. To prevent surface conductance at the bottom of submersed specimens, the bottom portions of the electrodes were embedded in small plexiglass studs. These studs, when inserted in holes cut in the mold bottom, also held the electrode in place during casting. Also, the upper portion of the electrodes were covered with nonconductive plastic shrink tubing to eliminate conductance between the exposed electrodes. Electrodes were cut from 1/8-inch diameter stainless steel rods. Finally, after curing, a 1 1/2-inch deep hole is drilled in the cylinder to accommodate a plastic thermistor housing. The thermistor is placed in this housing to obtain the necessary temperature measurements.

During testing, the cylinder is placed in a thin-walled plastic

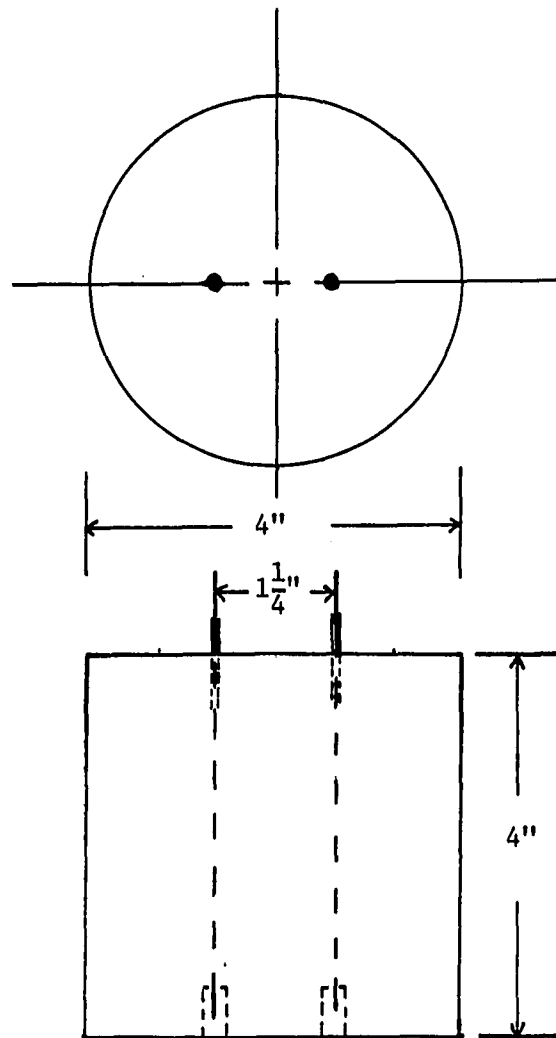


Figure 12. Electrode geometry for mortar and concrete samples

container, and the void between the walls of the cylinder and the container are filled with Ottawa sand. The sand is then saturated with water. Experimentation has shown that the silica crystals in the sand prevent supercooling of the bulk water surrounding the sample. This allows meaningful cooling and warming data to be gathered.

The test cycle for portland cement mortar and concrete cylinders was the same as the test cycle for the Vycor glass sample. For these cylinders, a 7 kilo-ohm resistor was connected in series with the cylinder to allow use of the more sensitive range on the conductance meter. Data processing routines calculated the conductance of the cylinder alone from the measured conductance as was done with the Vycor sample.

Materials

Vycor glass

As stated earlier, the Vycor glass tested was obtained from the Corning glass company. Corning gives this porous glass the designation "Vycor No. 7930." Table 6 (Enüstün et al., 1985a) lists the pore size range, porosity and specific surface of porous Vycor obtained by various researchers using various methods. Values for the conductance method listed in this table were obtained from inflection points of conductance versus temperature data gathered by Enüstün et al. (1978). From these data, these researchers were able to identify minimum, maximum and modal pore sizes. Conductometric phase transition porosimetry builds upon this promising work by calculating a complete pore size distribution from the

Table 6. Pore size range, porosity and specific surface of porous Vycor glass obtained using various methods (Enüstün et al., 1985a)

Method	Range of pore body radius, Å			Range of pore neck radius, Å			Porosity % by vol.	Specific surface, m ² /g
	Min.	Mode	Max.	Min.	Mode	Max.		
Phase trans. poros.	17	42	75-100	12	21	40	27 ^a -29 ^b	135 ^c
Mercury poros.	19	80	--	18	25	35	29 ^d	--
Conductance	17	36	60	--	24	26	--	--
TEM	15	38 ^e	85	--	--	--	--	--
Manufacturer's specification	10	--	100	--	--	--	28	150-200

^aFrom rewarming data.

^bFrom cooling data.

^cAssuming spherical pores.

^dFrom intrusion.

^eAfter converting number-size distribution to volume-size distribution.

same information. Vycor size distributions presented in this dissertation are slightly larger than those indicated by Table 6 (see Results and Discussion). This is due to the fact that the Vycor sample tested had been stored in deionized water for a period of over three years. This method of storage was suggested by the manufacturer because it would prevent contamination of the glass. It would appear that the storage in deionized water also resulted in some dissolution of silica, resulting in the aforementioned increase in pore sizes. This increase, although observable, is largely insignificant.

Portland cement mortar and concrete

Volumetric proportions for the portland cement mortar and concrete tested are given in Tables 7 and 8, respectively. The portland cement used was Lehigh Type I. The fine aggregate used was sand obtained from the Cordova quarry in eastern Iowa. The coarse aggregate used was an oolitic limestone from the Montour quarry in Iowa. This aggregate had a nominal maximum aggregate size of 3/4 inch. No admixtures were used with either the mortar or concrete mixtures. After mixing and molding, test cylinders were cured in a room temperature water bath until testing was begun. Curing time for mortar samples was 60 days and for concrete samples, 28 days. A replicate cylinder of concrete mortar was cast without electrodes. A sample from the center of this cylinder was used as a test sample for comparative analysis with mercury porosimetry.

Table 7. Volumetric mix proportions for mortar sample

Material	Volume
Cement	.194
Water	.261
<u>Fine aggregate</u>	<u>.545</u>
Total	1.000

Table 8. Volumetric mix proportions for concrete sample

Material	Volume
Cement	.116
Water	.156
Fine aggregate	.327
<u>Coarse aggregate</u>	<u>.401</u>
Total	1.000

RESULTS AND DISCUSSION

Vycor Glass

The conductometric phase transition porosimeter was first used to determine the pore size distribution of porous Vycor glass. In Figure 13, the logarithm of the conductance is plotted against the negative inverse of the absolute temperature (Y versus X) for the Vycor sample tested. The linearity of the curves prior to freezing or after melting of capillary water supports the relationship between electrical conductivity of an electrolyte and temperature as defined in equation 32. The sharp drop in conductance which occurs at approximately -8°C on the cooling curve is due to the rapid freezing of supercooled bulk solution.

Figures 14 and 15 illustrate two graphical means of presenting conductometric pore size distributions. To obtain these plots, the Y versus X data illustrated in Figure 13 are processed using equations 35, 36, and 37 and the numerical integration procedure described in Figure 8. Pore geometry factors are assumed to be independent of pore size ($\alpha = 1$ in equation 37). In Figure 14, the cumulative relative conducting pore volume is plotted against pore radius. The relative conducting pore volume at any point on the curves is the fraction of the total pore volume smaller than the corresponding pore radius for that point. The curve labeled neck size is obtained from data gathered during the cooling portion of a test and the curve labeled body size from the warming portion. The first derivative of relative conducting pore volume with respect to pore radius is plotted against pore radius in Figure 15.

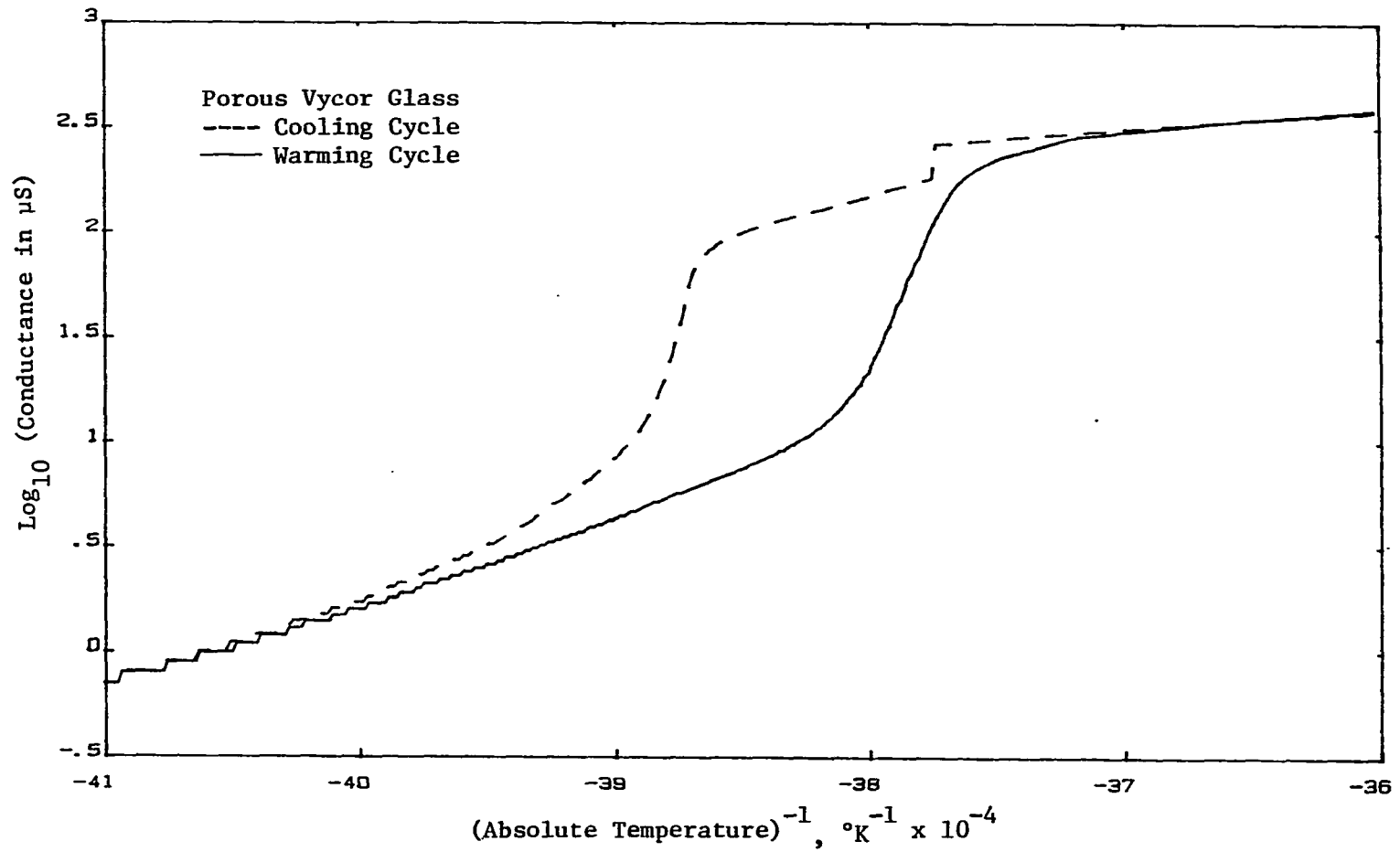


Figure 13. Logarithm of the conductance versus negative inverse of absolute temperature (Y vs. X) for Vycor glass

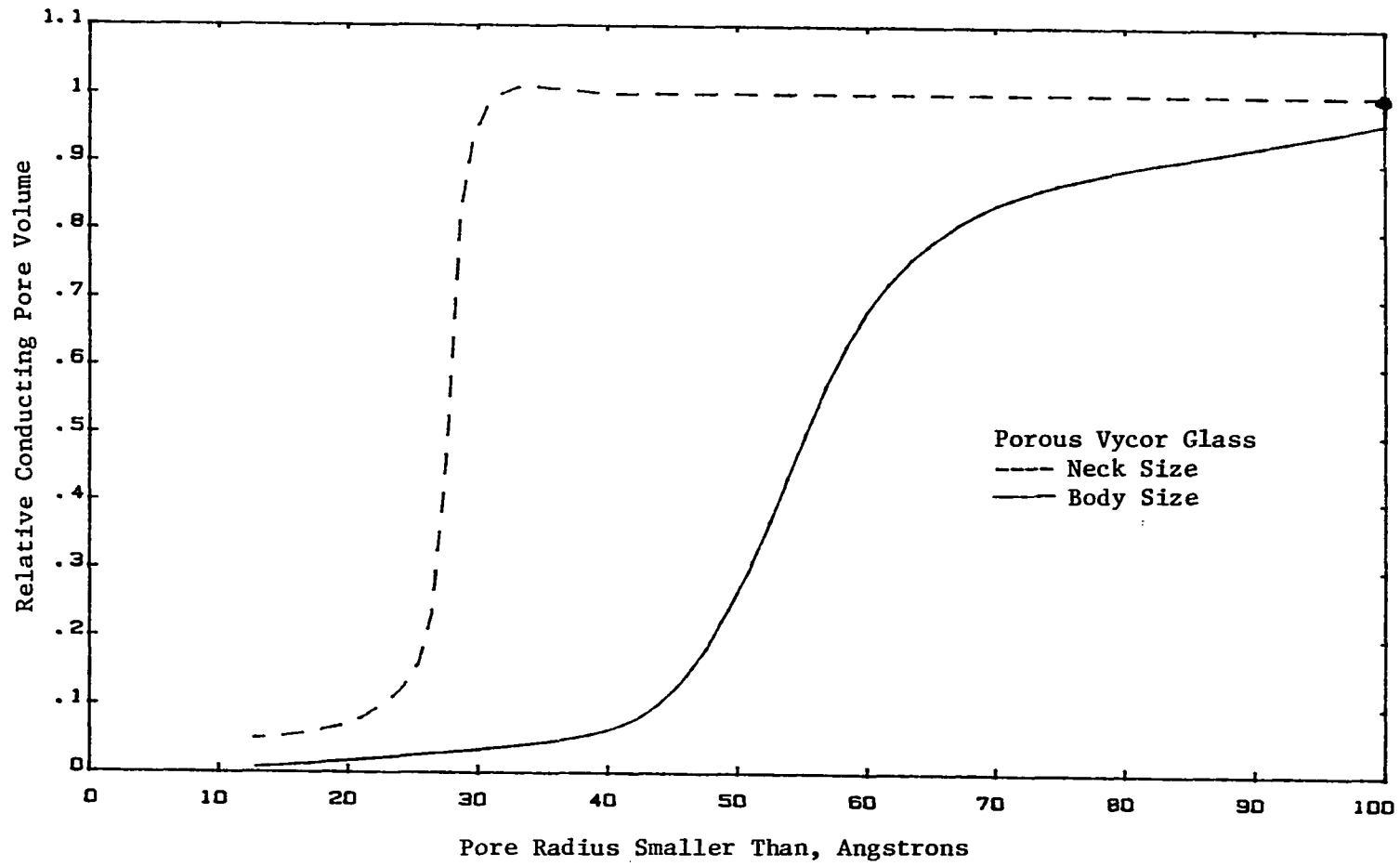


Figure 14. Conductometric pore size distribution of Vycor glass

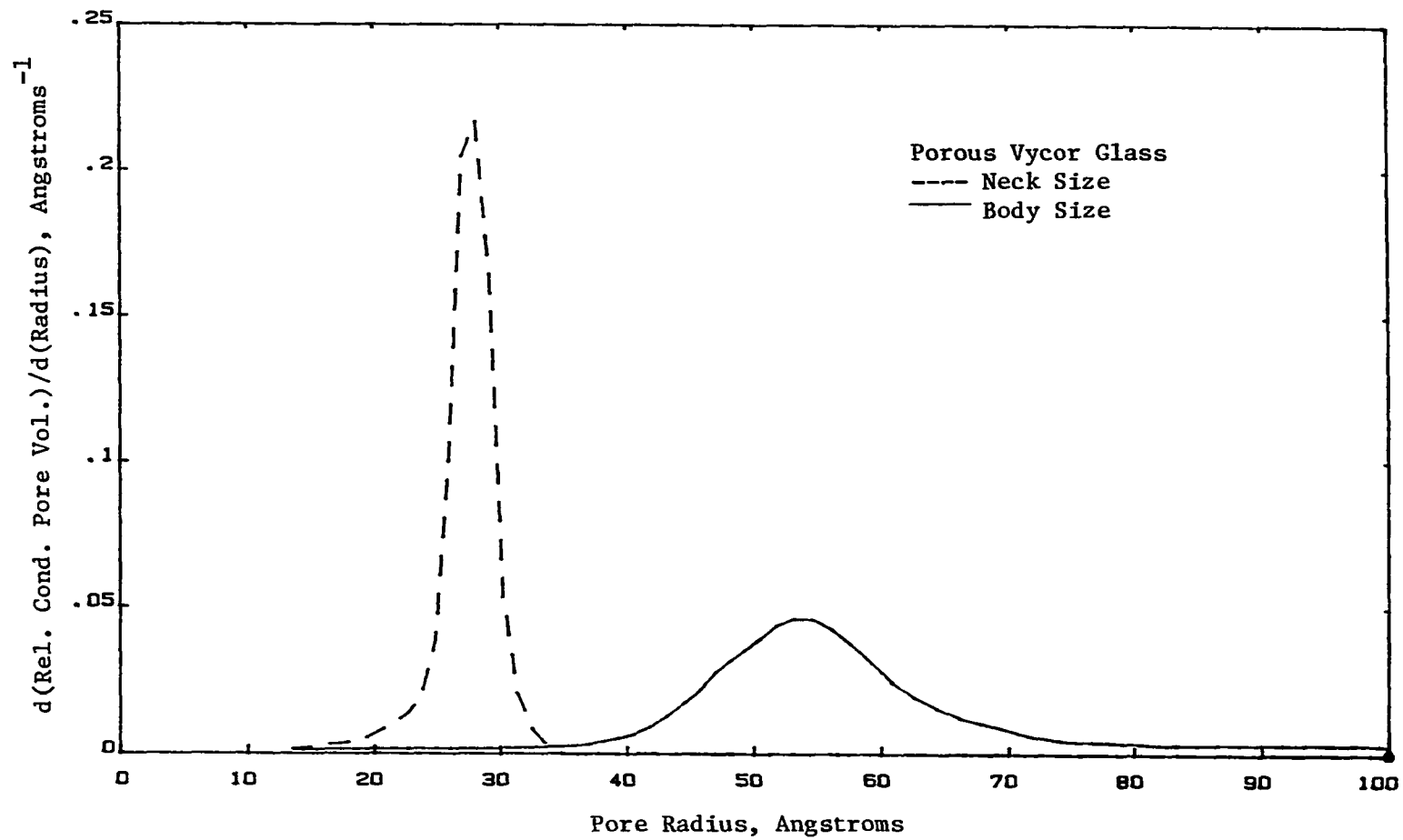


Figure 15. Conductometric pore size distribution of Vycor glass

Table 9 gives the minimum, modal and maximum pore sizes determined from this plot. Extrapolation of maximum and minimum values from distribution curves such as these is somewhat subjective. In spite of this and the aforementioned increase in pore sizes due to dissolution of silica over time, the results presented in Table 9 agree quite well with the results obtained by other researchers presented in Table 6.

Table 9. Minimum, modal and maximum pore radii in angstroms for Vycor glass as determined by conductometric phase transition porosimetry

	Minimum	Modal	Maximum
Neck size	15	29	34
Body size	35	54	75

Evaluations of these results were also made by comparing them with results obtained on the same sample by dilatometric phase transition porosimetry and by mercury porosimetry. Quantitative conductometric pore size distributions were calculated from relative pore size distributions by multiplying relative values by the total pore volume of the sample. The sample's total pore volume was determined gravimetrically to be equal to 0.21 ml. Figures 16 and 17 contrast the pore size distributions obtained from conductometric phase transition porosimetry, dilatometric phase transition porosimetry and mercury porosimetry for neck and body sizes, respectively. It can be seen from these figures that all three methods give comparable results for neck sizes. For body sizes, some

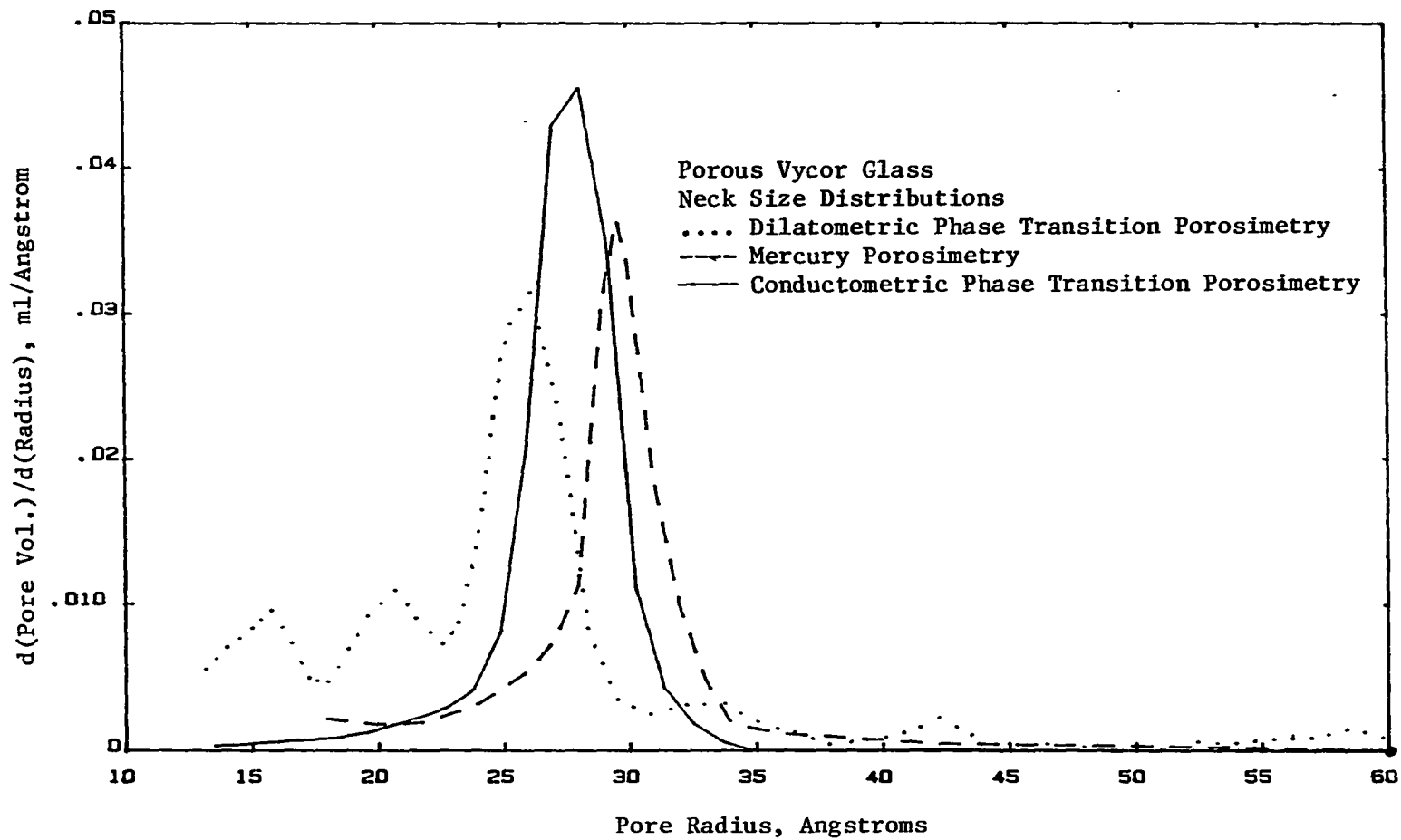


Figure 16. Comparative neck size distributions for Vycor glass

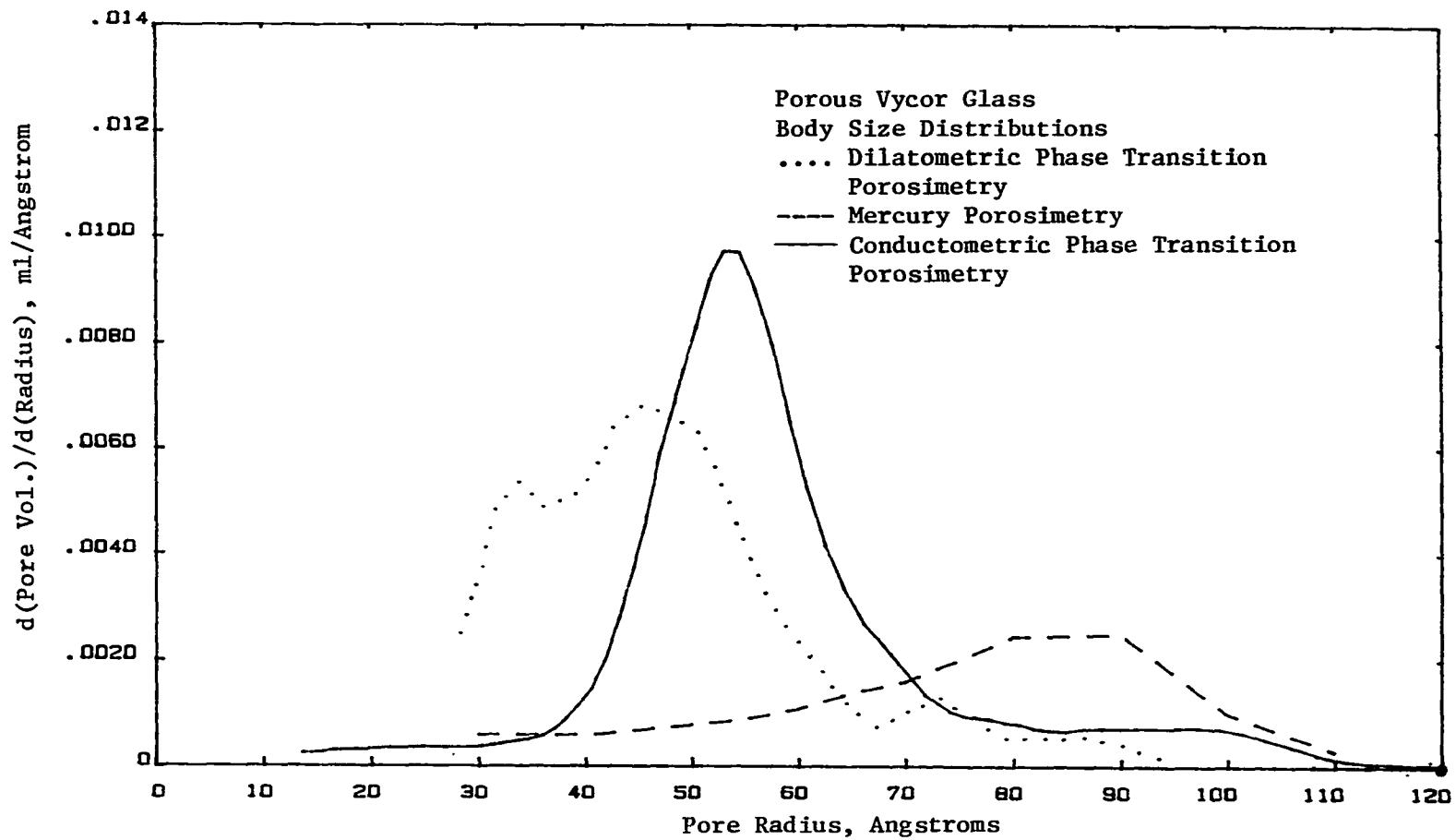


Figure 17. Comparative body size distributions for Vycor glass

differences are observable. It is believed that the differences between the body size distributions as determined by phase transition methods and that determined by mercury porosimetry are due to uncertainties involved with the entrapment of mercury during extrusion as discussed earlier.

Portland Cement Mortar

The conductometric phase transition porosimeter has been used to evaluate the pore size distribution of the 4-inch diameter by 4-inch high portland cement mortar cylinder discussed earlier.

A small portion (approximately 2 gm) was taken from the center of a replicate mortar cylinder. The pore size distribution of this portion of mortar was determined by mercury porosimetry. The mercury porosimeter sample was dried at room temperature, pressure and relative humidity prior to the evacuation process necessary with mercury porosimetry.

Resultant pore size distributions for this concrete mortar are shown in Figure 18. In this figure, percentages of the total pore volume are plotted against the logarithms of the pore radii. The wide range of pore sizes indicated by these size distributions is characteristic of hydrated portland cement (Mikhail et al., 1964). The mean pore neck size as evaluated by mercury porosimetry is nearly three times as large as the mean pore neck size as evaluated by conductometric phase transition porosimetry.

Mikhail et al. (1975), referencing data from Diamond (1971), indicates that mean pore radii of portland cement pastes obtained from water adsorption data range from 21 to 50 angstroms, while mean radii

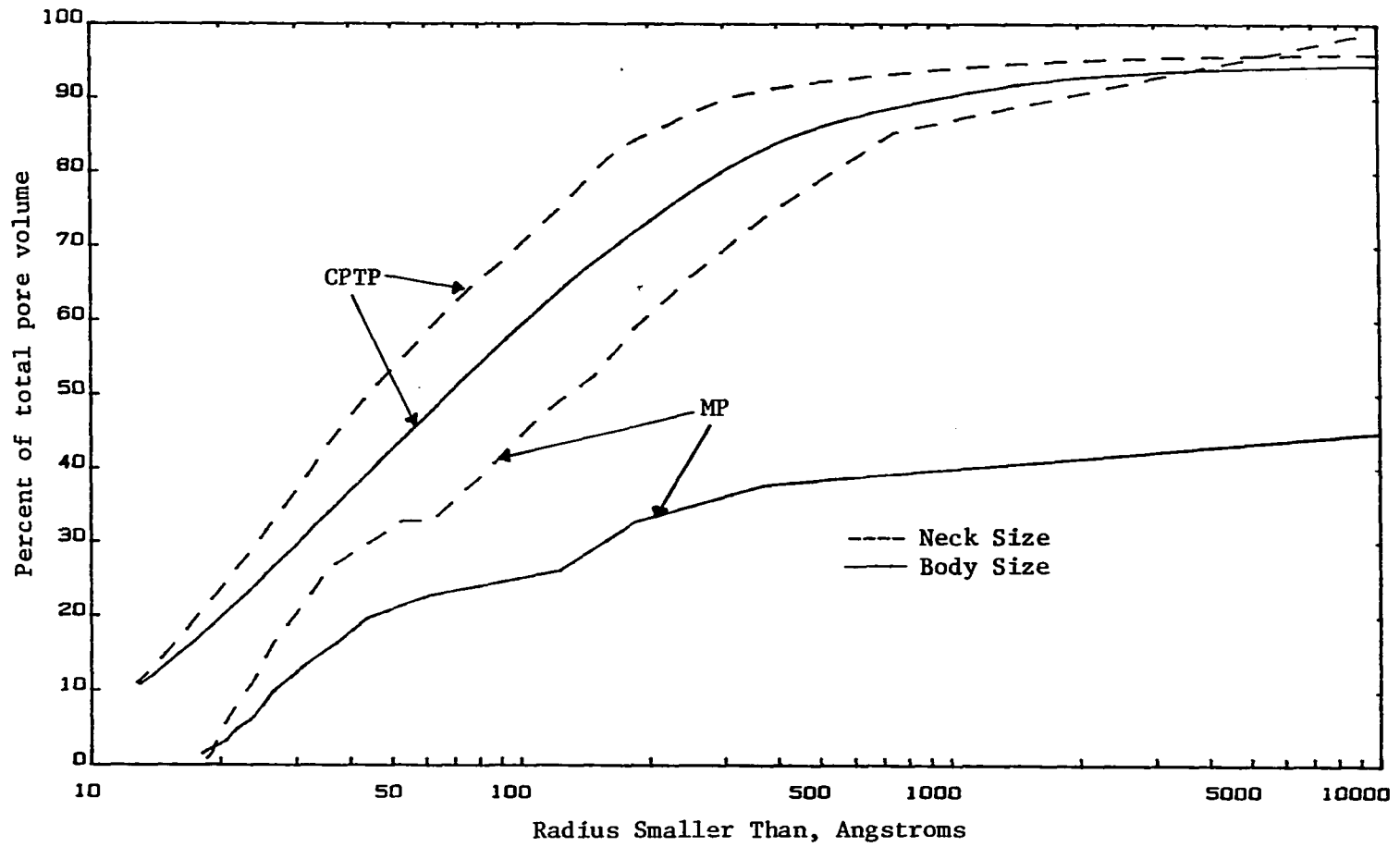


Figure 18. Pore size distribution of portland cement mortar as determined by conductometric phase transition porosimetry (CPTP) and mercury porosimetry (MP)

obtained from mercury intrusion data range from 115 to 385 angstroms. While the material tested in the present study was not portland cement paste but rather portland cement mortar, the mean radii determined for the mortar by mercury porosimetry fall within the expected range for mercury porosimetry. In contrast, the mean radii determined by conductometric phase transition porosimetry fall within the expected range for water adsorption. Mikhail et al. (1975) do not attribute these differences to any fundamental flaw in the theory behind either method, but rather to the inability of mercury porosimetry to determine complete size distributions. They reject the large mean radii and subsequent large surface areas determined by mercury porosimetry because of the presence of pores beyond the range of mercury porosimeters (approximately 18 angstroms). Indeed, they question the validity of results obtained for radii smaller than approximately 50 angstroms due to possible deformation or destruction of pores at the high pressures necessary with mercury porosimetry.

Although direct comparison of results obtained by conductometric phase transition porosimetry with results obtained by adsorption is not possible for the mortar tested, it would appear conductometric phase transition porosimetry would give results more typical of water adsorption than mercury porosimetry. Conductometric phase transition porosimetry results indicate approximately 10% of the total pore volume of the mortar is associated with pores smaller than 12 angstroms. This would be expected because the pressure gradient across the ice/water meniscus in phase transition porosimetry is much less than the gradient

across the mercury meniscus in mercury porosimetry. This, coupled with the ability of phase transition porosimetry to analyze pores down to a size of 12 angstroms, allows a more complete and accurate size distribution to be obtained.

Finally, it would appear from examination of the data presented for Vycor and mortar that the errors introduced by assuming that pore geometry factor is independent of pore size for a given pore system is not significant.

Portland Cement Concrete

Pore size distributions for the portland cement concrete cylinder were determined by conductometric phase transition porosimetry and results are shown in Figures 19 and 20. In Figure 19, relative conducting pore volume is plotted against pore radius; and in Figure 20, the first derivative of pore volume with respect to the logarithm of the radius is plotted against the logarithm of the radius. These distributions illustrate the use of the porosimeter with a material whose pore size distribution is not obtainable by other techniques. Portland cement concrete is a highly heterogeneous material and samples that can be considered to be representative of the material are much too large for analysis by other techniques.

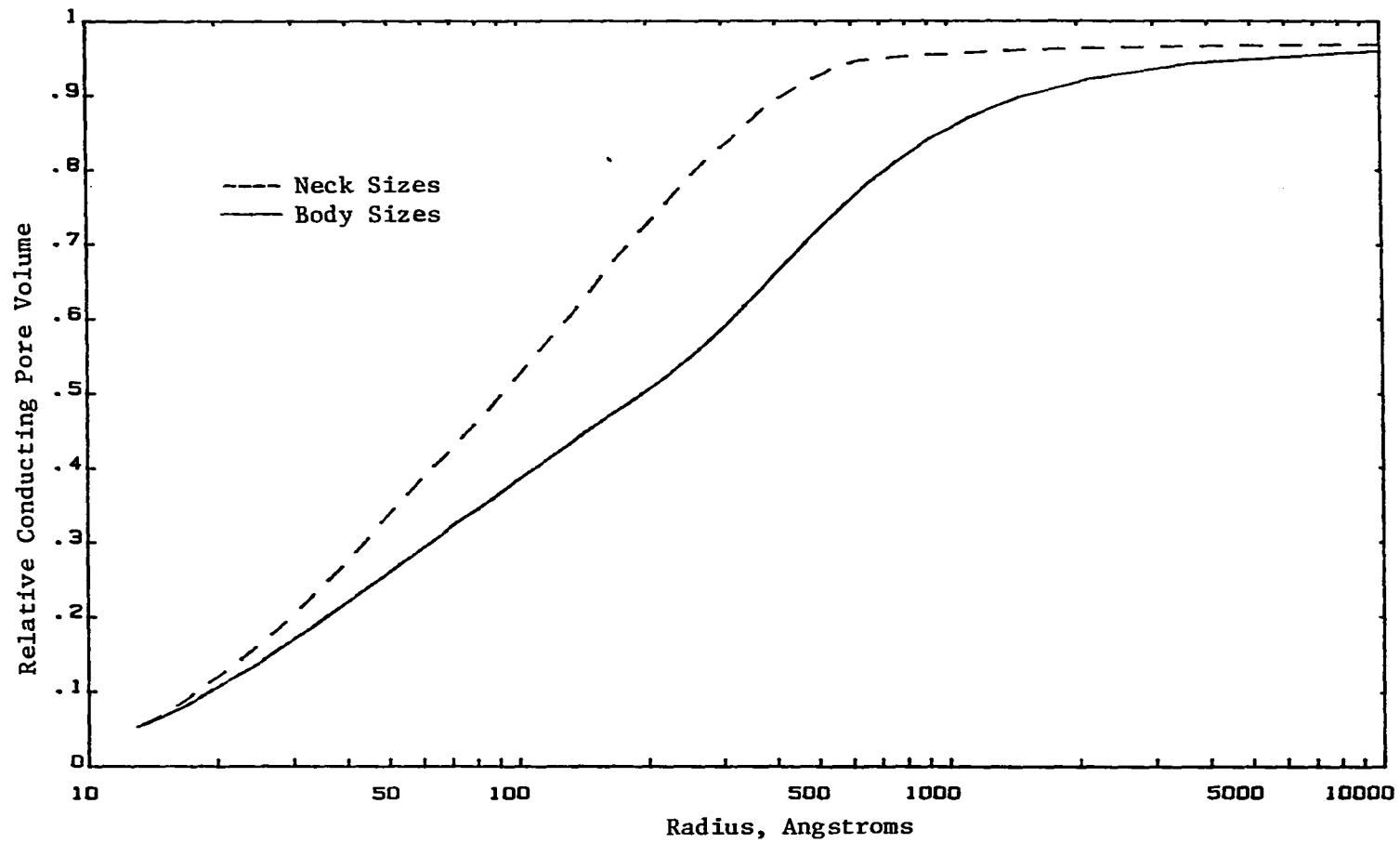


Figure 19. Pore size distribution of portland cement concrete

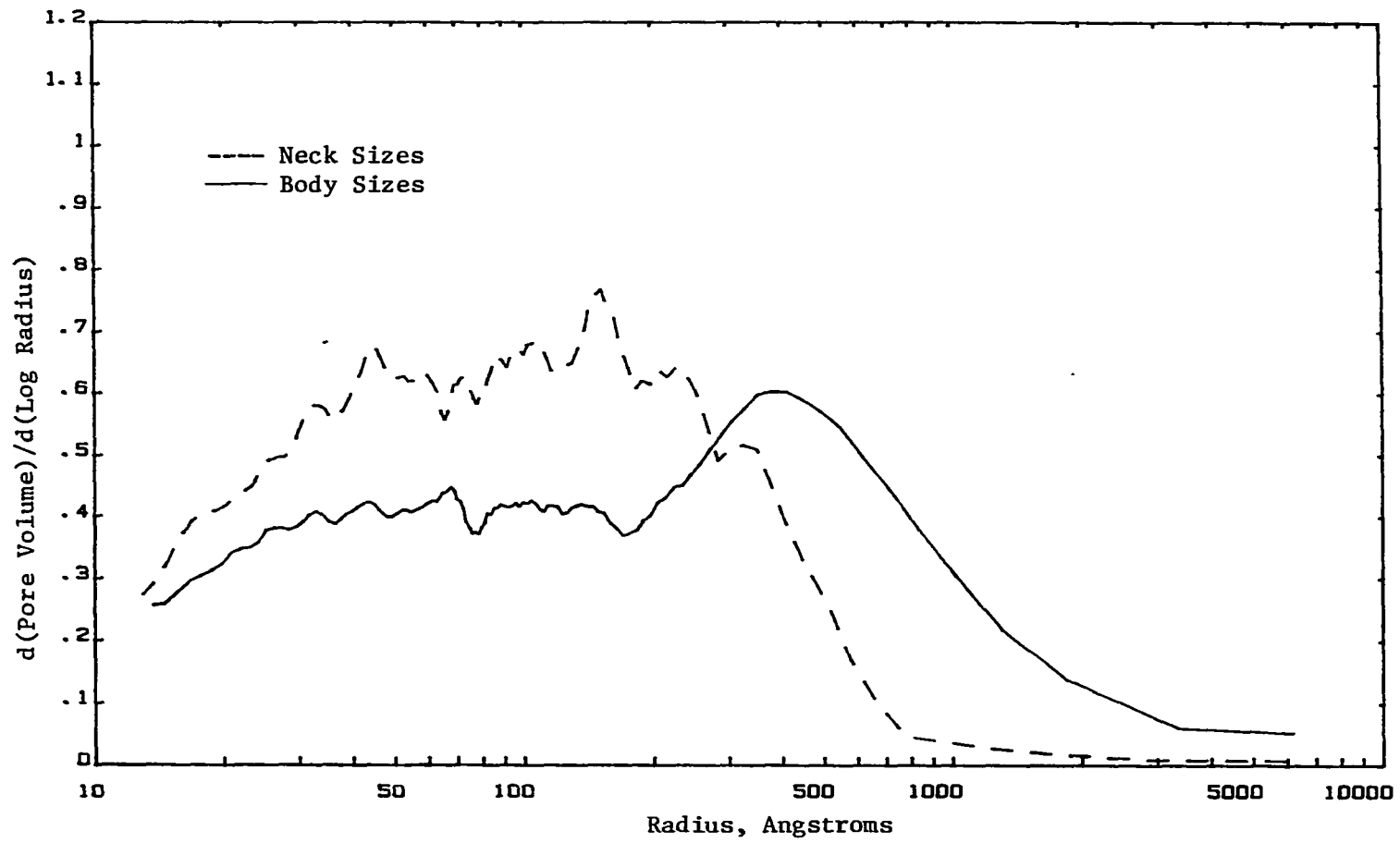


Figure 20. Pore size distribution of portland cement concrete

SUMMARY AND CONCLUSIONS

A new method for determining pore size distributions of saturated porous materials has been developed. This new methodology, referred to as conductometric phase transition porosimetry (CPTP), combines relationships correlating the volume of unfrozen pore solution to electrical conductance, the pore solution freezing temperature to pore radii, and the electrical conductivity of the pore solution to temperature. These relationships are used with measurements of the electrical conductance and temperature to obtain pore size distributions of saturated porous materials subjected to a cycle of capillary freezing and melting.

The plastic ice model of capillary freezing and melting first introduced by Everett is essential to CPTP, and this model is reviewed in this dissertation. This thermodynamically based relationship between pore water freezing temperatures and pore radii has been expanded to include the freezing of pores saturated with solutions containing nonvolatile solutes.

This dissertation also introduces the parameter called pore geometry factor and a relationship which expresses the electrical conductance of a material as a function of pore geometry factor, pore volume, pore solution conductivity, and sample geometry is introduced. This relationship is essential to CPTP and its physical significance is explored.

Based on this newly developed methodology, a conductometric phase

transition porosimeter was constructed. The porosimeter is capable of determining the pore size distributions for a broad range of pore radii ranging from 12 to 5000 angstroms. This porosimeter has been used to accurately determine the pore size distribution of porous Vycor glass. In addition, techniques for determining the pore size distributions of portland cement mortars and concretes have been introduced.

Conductometric phase transition porosimetry is free of the problems of entrapment of mercury inherent to mercury porosimetry. Also, the pressures a sample is subjected to during testing are much less than those necessary with mercury porosimetry. Therefore, deformation or destruction of pores will be significantly less with conductometric phase transition porosimetry for many materials provided the freezing and thawing rate is slow. The slow rate is necessary to prevent viscous forces from damaging the sample.

With conductometric phase transition porosimetry, samples do not have to be dried prior to testing as with mercury porosimetry and adsorption techniques. For materials which have pore structures that vary with moisture content, conductometric phase transition porosimetry provides researchers with a means of investigating these relationships.

In contrast with dilatometric phase transition porosimetry, conductometric phase transition porosimetry can be used to determine the pore size distribution of the saturated pores of a partially saturated pore system.

Conductometric phase transition porosimetry holds an advantage over other available techniques in that it allows much larger samples to be

tested. The porosimeter has been used to evaluate the pore size distributions of 4-inch diameter by 4-inch high portland cement concrete cylinders.

Potential errors associated with the assumption that pore geometry factors are independent of pore sizes for a given pore system have been shown to be insignificant in determining the pore size distributions of Vycor glass and portland cement mortar.

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APPENDIX

**CONDUCTOMETRIC
PHASE TRANSITION POROSIMETER**

**USER'S MANUAL
Volume 1
Operation Instructions**

**CONDUCTOMETRIC PHASE TRANSITION POROSIMETER
USER'S MANUAL
Volume 1**

by
Todd Millard
and
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IOWA STATE UNIVERSITY, 1987

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USER'S MANUAL
Volume 1
Operation Instructions**

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I. INTRODUCTION

Conductometric phase transition porosimetry is based on the assumption that the electronic conductance through the solid phases of a porous material is negligible when compared to the electrolytic conductance of the pore solution. Thus, when the temperature of a mass of saturated porous material is raised from sub-freezing temperatures the frozen pore solution will melt, and an increase in electrical conductance will be observed. As ice in capillaries melts at temperatures below the normal melting point of ice, the change in electrical conductance of the material with temperature will be a function of the pore structure of that material. The methodology combines the plastic ice model for solid-liquid phase transitions of pore water with relationships between conductance and temperature, and conductance and pore structure. The theory behind conductometric phase transition porosimetry is developed from a few basic relationships.

Assuming that the pore geometry of the sample can be represented by randomly intersecting spheres or cylindrical capillaries with or without constrictions, it can be shown using the plastic ice model, that the solid-liquid phase transition of water in a pore having an effective radius of r takes place at a temperature, t ($^{\circ}\text{C}$) given by:

$$t = -2\gamma T_0 / (\rho \lambda r) \quad (1)$$

where T_0 is the normal melting point of ice in $^{\circ}\text{K}$, λ is the heat of fusion of ice per unit mass, ρ is the density of water and γ is the ice/water interfacial tension at t . In melting, r is the pore body radius, while in freezing it is the radius of the pore constriction.

A relationship between the ice/water interfacial tension, γ , and temperature is used where

$$\gamma = \gamma_0 + kt \quad (2)$$

where t is the temperature in $^{\circ}\text{C}$, k is a dimensional constant with a value of $0.25 \text{ ergs cm}^{-1} \text{ K}^{-1}$, and γ_0 is the value of γ at 0°C . Also, for conductometric phase transition porosimetry it is helpful to develop a relationship between pore radius, r , and the negative inverse of the absolute temperature, X . With this information equation 1 can be rewritten as

$$r = \frac{-2T_0(k - \gamma_0 X + kT_0 X)}{\rho\lambda(1 + T_0 X)} \quad (3)$$

where r , T_0 , γ_0 , k , ρ , and λ are as defined earlier and $X = (-1/T)$, where T is the phase transition temperature in $^{\circ}\text{K}$.

The numerical values of the various constants used in computations contained in this document were as follows:

$$\gamma_0 = 29 \text{ N m}^{-1}$$

$$T_0 = 273.16 \text{ }^{\circ}\text{K}$$

$$\rho_w = 1.000 \text{ gm cm}^{-3}$$

$$\lambda = 333.3 \text{ J gm}^{-1}$$

These values were obtained from physical and chemical handbooks.

The relationship between conductance of a porous material, C' , and pore structure for a cylindrical sample of porous material with cross-sectional area, A , and length, L , is defined as

$$C' = \kappa (V_p / (L^2 \omega)) \quad (4)$$

where κ is the electrical conductivity of the pore solution, V_p is the pore volume and ω is a dimensionless pore geometry factor.

Also, it is assumed the relationship between the electrical conductivity of an electrolyte and the absolute temperature of the electrolyte is an Arrhenius type relationship and can be expressed as

$$\ln \kappa = \frac{a}{T} + b \quad (5)$$

where

κ - electrical conductivity,

T - absolute temperature,

a - physical constant, and

b - physical constant.

By combining equations 4 and 5 the following equation is derived

$$Y = mX + d + \log_{10} \left(V_p / (L^2 \omega) \right) \quad (6)$$

where L , V_p and ω are as defined earlier,

$X = -1/(\text{absolute temperature}),$

$Y = \log_{10} (\text{conductance}),$

m - physical constant, and

d - physical constant.

A plot of the \log_{10} of the conductance versus the negative inverse of the absolute temperature (Y versus X) for which no phase change occurs will result in a line with a slope equal to m . Below freezing temperatures the phase changes which occur in a certain range of pores will effectively

decrease the volume of conducting pores. Thus, V_p at below freezing temperatures can be considered to be the volume of conducting pores, V_{cp} .

Taking the first derivative of equation 6 yields

$$\frac{dY}{dX} = m + \frac{d(\log_{10}(V_p/(L^2\omega)))}{dX} \quad (7)$$

Integrating equation 7 over the definite interval, X to X_0 , yields the following:

$$\log_{10} \left[\frac{V_{cp}(X_0)}{L^2\omega_{X_0}} \frac{L^2\omega_X}{V_{cp}(X)} \right] = \int_X^{X_0} \frac{dY}{dX} dX - \int_X^{X_0} m dX \quad (8)$$

The right half of equation 8 can be calculated numerically from conductance test data and is given the variable name Z . Also, if X_0 is the value of X at the pore solution melting point (273.15 °K for water), then $V_{cp}(X_0)$ is simply the total conducting pore volume, V_p . Rearranging equation 6 and referring to pore radius instead of temperature in light of equation 1 we obtain:

$$\alpha \frac{V_{cp}(r)}{V_p} = 10^{-Z(X, X_0)} \quad (9)$$

where $V_{cp}(r)$ is the volume of the pores with radii smaller than or equal to r containing unfrozen pore solution, V_p is the total pore volume and α is the ratio of the pore geometry factor of the total pore system, ω , to the pore geometry factor of the pores smaller than r , ω_r . If it is assumed that the

shape factor, ω , is independent of pore size, then α is equal to unity and equation 9 can be reduced to

$$\frac{V_{cp}(r)}{V_p} = 10^{-Z(X, X_0)} \quad (10)$$

Assuming α to be equal to unity implies the pore system has a certain degree of homogeneity. This does not imply that all the conducting pores are exactly the same but rather that the tortuosity and necking inherent to a given pore structure is uniform throughout the pore size distribution or otherwise stated that they are geometrically similar.

These relationships provide the theoretical background for conductometric phase transition porosimetry.

The Apple IIe computerized data acquisition system has been developed. The system consists of the Apple IIe computer (with two disk drives, a monitor, a parallel interface card, and a serial interface card), a cooling bath, a temperature interface box, and a Solomat conductance meter. The computer controls the bath and records the data for a typical experiment.

II. TUTORIAL

This section will take a step by step look at a typical multicycle conductance test. For further information on any step refer to the detailed descriptions in the program or interface sections.

1. Place program disk in drive 1 and raw data disk in drive 2, then reboot the computer by pressing the CTRL, open-Apple, and RESET keys simultaneously. The computer will restart and load the Main Menu program from the program disk. The programs used in this system are menu based. By choosing a number (to change a parameter) or a letter (to choose an action) the program is controlled. While inputting a selection or data, the left and right arrows on the lower right hand corner of the keyboard may be used to correct mistakes. Press the RETURN key to enter your selection.
2. Place the sample in an appropriate container and immerse in the cooling bath. The sample should be saturated with water and surrounded by bulk water. The container should be slightly larger than the sample and able to separate the cooling fluid from the sample. Connect the leads from the Solomat conductance meter to the probes mounted in the sample. If better resolution is needed at the lower end of the conductance scale place a resistor in series with the sample to lower the overall conductance below $160 \mu\text{S}$ and set the scale on the Solomat to $160 \mu\text{S}$ for the 1.0 cell. If a series resistor is used, the sample's conductance will be calculated by the data processing programs when the appropriate data are entered in the Main Menu before processing. If a series resistor is not used set the scale on the conductance meter to $16000 \mu\text{S}$ for the 1.0 cell. The thermistor should be placed adjacent to the sample. Turn the cooling bath on by using the green main switch first followed by the black compressor switch. Manually set the bath's temperature to the maximum temperature of the first cycle. It will take about 20 to 30 minutes for the temperature of the sample to reach that of the bath.
3. Start the controlling program by entering C for Run Conductance Test. The program will allow some notes to be entered. Then enter the number of cycle types and define each of the types by entering the maximum temperature, minimum temperature, rate of temperature change, and temperature interval between readings for each cycle type. If data collection is not needed for a particular cycle type enter the number 0 as the

temperature interval. Next enter the total number of cycles for the test sequence and the cycle type for each cycle. Finally, enter the sample's base file name and the starting cycle number for the test sequence. The test sequence will begin. Place the temperature preset switch on the rear of the cooling bath in the ON position. The computer will control the entire test and return to the Main Menu when finished. When the test sequence is completed, turn off the cooling bath and remove the sample.

4. The next step is to process the data. The parameters shown on the Main Menu control the processing. First choose a1 for File Name. If the name of the file you wish to process is unknown, press RETURN and a catalog of the disk in drive 2 will be displayed. Enter the base file name and the cycle number. For example, Mortar.25 would be entered for the 25th cycle of the sample named mortar. If a resistor was connected in series with the sample indicate so and input the CONDUCTANCE of the resistor (μS). The derivative programs use a moving average in their algorithms to smooth the data. Enter the desired moving average for each of the derivative steps. Some of the intermediate steps generate data that may or may not be of interest. Respond with a Y to a Save File option and the data for that file will be saved to a disk. The temperature of T0 is the normal freezing point of the pore solution in degrees Kelvin (usually 273.15 K). Once the parameters are set in the desired manner, enter A and calculations will be performed.

5 About one minute into the processing the program will prompt the user to remove the raw data disk and insert a Caldat disk. The processed information is stored on a different disk than the raw data and the raw data has been loaded and processed at this point. The total process takes roughly 7-8 minutes for a cycle with 350 points per portion (warming and cooling) and the program returns to the Main Menu when finished.

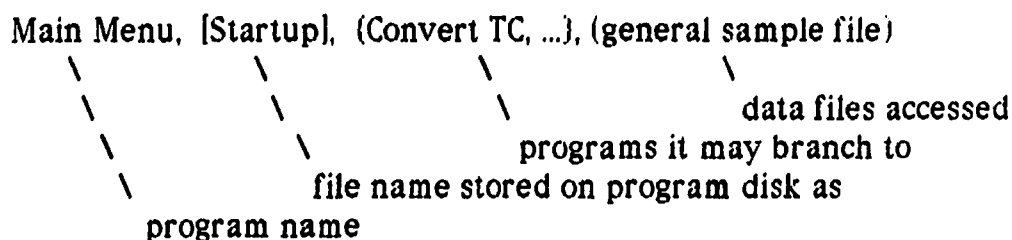
6. To make plots of the data choose B for Plot Curves. The computer will load the Plotting Program and display a menu of data file types (from X.Y through R.DVPDLR). If selection 6 or 7 is chosen, the option of plotting the radius on a log scale is available. The plotting menu will then be displayed. Both the warming and the cooling data, may be plotted simultaneously. Enter the file name(s) of the data (base filename, cycle number, and W for warming or C for cooling, i.e., Mortar .25C for the cooling portion of the 25 cycle of a sample named mortar). Plotting parameters may be adjusted to the user's preference and the range of the data. It is advisable to make a

plot on the computer screen to ensure the parameters are set up correctly before making a plot on the plotter. The computer may be stopped while plotting to the screen at any time by pressing the ESC key. The plot is finished when the computer beeps several times. Pressing any key will return the computer to the menu.

7. When all of the necessary plots have been made, enter E to Exit Program and return to the Main Menu. When finished, return the disks to the cases and turn off the computer monitor. The computer itself is left on to maintain the stability of the amplifier circuits.

III. PROGRAMS

In this section the purpose, operation, and the structure of each program will be discussed. The structure will be given to help understand the program and solve possible problems. Each program will be listed with the following information



Most programs are menu based, the user options are listed on the screen by number or letter. Choosing a number allows you to edit the data or a parameter and selecting a letter chooses a particular action. While inputting a selection or data, the left and right arrows on the lower right hand corner of the keyboard may be used to correct mistakes. Press the RETURN key to enter the selection.

There are three types of disks in this system. One of the disks contains the programs to control the tests, process the data and output the results. The other two types are data disks. The raw data disk contains the temperature vs. conductance information for all cycles and the general sample file. The CALDAT disk contains the processed data for the directories listed in the introduction section. Due to disk space limitations, only about two fully processed cycles (all directories saved) or 3 partially processed cycles (only saving X.Y, R.VP, and X.DYDX) may be saved on one CALDAT disk (assuming 340 points per portion of the cycle or 680 points per cycle). More cycles may be stored on a disk if the number of points per cycle is decreased.

DIRECTORY	CONTENT
X.Y	Converted temperature and Conductance
X.DYDX	$d(Y)/d(X)$
X.Z	Integral of Y
X.VP	Pore Volume
R.VP	Pore Volume vs. Radius
R.DVPDR	$d(\text{Pore Volume})/d(\text{Radius})$
R.DVPDLR	$d(\text{Pore Volume})/d(\text{Radius})$

The control program will be discussed first, followed by the data processing and output programs. Program listings can be found in volume 2 of this manual.

MAIN MENU, (Startup), (Plotter, Controller, Convert XY, Derivative XY, Integrate, Convert VP, Convert R, Derivative VP), (raw data file)

PURPOSE : Main Menu is the central program of the entire system. From Main Menu a new test sequence may be started, data plotted, or data calculations initiated.

OPERATION : Main Menu is the first program that will appear upon starting the system. From Main Menu a test may be started, data calculations initiated, or plots of data made. To start a new test press "C" and you will proceed directly to Controller. To make a plot press "B" and Plotter will be run. The format of the test may be printed on the plotter by entering "D" to run the Test Format program.

The items that appear on the screen initially are only used to control the calculation process. This data is only used for current control purposes and is not stored for each file. To load a file input a "1" for File Name and enter a file name consisting of a base file name and the cycle number. If the name or cycle number of the file is unknown, press RETURN and a catalog of the data disk will be shown. If a resistor was placed in series with the sample enter a "Y", if not a "N". If there was a resistor in series enter the conductance (not the resistance) of the resistor in uS. This data will be used to calculate the conductance of the sample. Both derivative programs use an algorithm that averages the data over a specified interval which may be adjusted. Several of the intermediate programs offer the option of saving the file. If the data from these programs is not important enter a "N" in the appropriate spaces. Not saving the intermediate data will speed the calculations and save disk space. When the temperature data is converted to radius the file may be thinned out so that there is at least one Angstrom between each of the data points. Thinning the file will cut the plotting time and also save disk space. The temperature T0 is the pore solution melting point in degrees Kelvin. A sample menu is shown on the next page.

CONDUCTANCE MENU

1.	File Name		:	Mortar.25
2.	T.C to X.Y	Series Resistor	:	Y
		Cond of Resistor	:	144.3
3.	X.Y to X.DYDX	Moving Average	:	3
		Save File	:	N
4.	X.DYDX to X.Z	Save File	:	N
5.	X.Z to X.VP	Save File	:	N
6.	X.VP to R.VP	Thin File Out	:	Y
		Temp of T0	:	273.15
7.	R.VP to R.DVPDR	Moving Average	:	3
8.	R.VP to R.DVPDLR	Moving Average	:	3

- A. Calculate the data
- B. Plot data
- C. Run Conductance test
- D. Print Test Format
- E. Exit to system

Enter your menu choice :

CONTROLLER, [Cond2], (Main Menu, Clock, Temp Timer), (raw data, general sample file, CAL)

PURPOSE : Controller controls the cooling bath and records data during a conductance test of the sample.

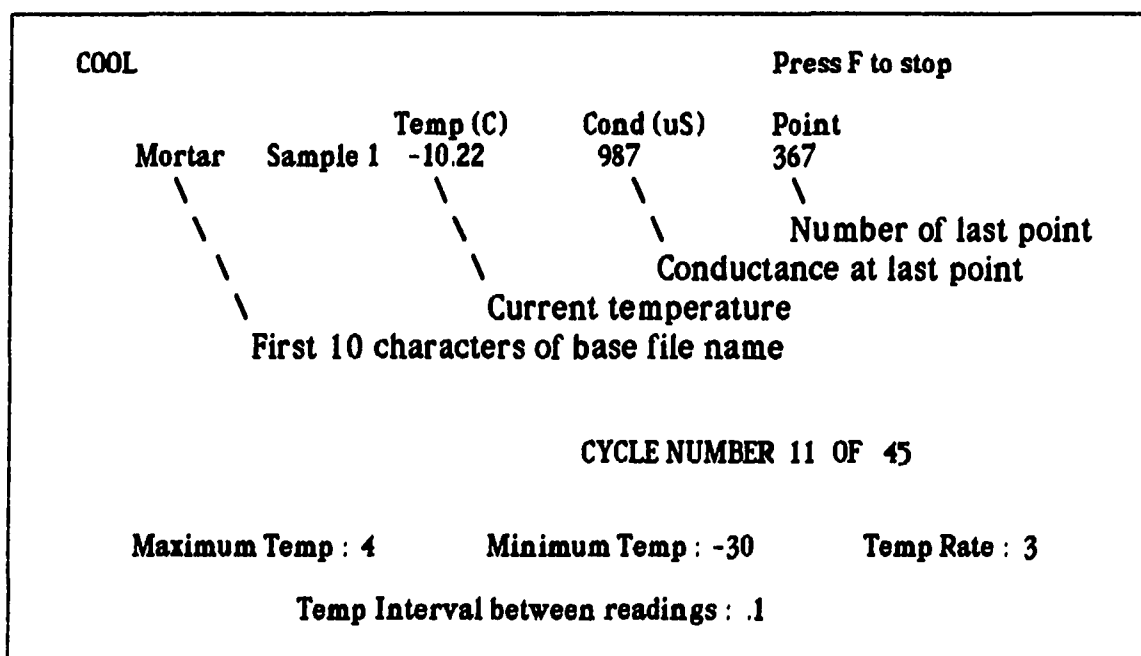
OPERATION : At the beginning of the program will prompt the user to enter some information about the sample and specify the testing parameters. The computer will first prompt the user to enter some notes on the sample. Typical information would be the date and sample material. Don't put any commas into the information or the computer will ignore the rest of the line after the comma. At the end of each line press RETURN.

Now the program prompts the user to enter the number of cycle types. Any number of cycle types may be specified. For each cycle type enter the maximum temperature, minimum temperature, rate of temperature change, and the temperature interval between data readings. The computer will check the input values to insure the minimum temperature is above the maximum, a negative rate of temperature change is not entered, a rate greater than 30 C/hr is not entered, and that the temperature interval for data readings isn't greater than the difference between the maximum and minimum temperature or negative. If a temperature interval of zero is entered the program will not record any data on that cycle. This feature is useful for multiple cycle runs when only data at certain cycle intervals is needed. All of the maximum temperatures must be the same so that when the computer switches from one cycle type to another there is a smooth transition. If the current cycle has a maximum lower than the next cycle's the computer will begin taking data as soon as the current cycle is over and continue until the temperature is above the level of the next reading.

Once the data for all of the cycle types has been entered the program will prompt the user to input a test profile consisting of a sequence of the cycle types. Input the total number of cycles in the test profile and the cycle type for each cycle. The cycle types are described by number, the first one entered being number one.

Now that the test profile is completed, input the base file name for the sample and the cycle number the test is starting on. For example if a previous test of 45 cycles was run on the sample, input the same base file name and then enter the starting cycle as 46 and the cycle number of the data will now be correct.

The test will now start and a status screen will be displayed. A sample status screen is shown below.



The display shows the current temperature of the sample, the conductance of the last data point taken, and the number of that point. The screen also displays the cycle parameters, the current portion of the cycle (heat or cool), and the progress of the overall test (in this case on the 11th of 45 cycles).

To discontinue the experiment at any time press F and the computer will stop at the end of the cycle. The computer must stop at the end of a cycle to save the data for that cycle.

The data is saved at the end of each cycle and when the test is complete the program returns to the Main Menu.

STRUCTURE : Controller allows the user to input notes on the sample, the parameters each cycle type, and the cycle types that make up the test profile. This information is stored on the raw data disk as the CT file.

The cycle type parameters are stored in the arrays TP, BTTM, INVL,, and RA for maximum temperature, minimum temperature, temperature interval and the rate of temperature change respectively. The TD\$ array is used to tell the program whether data should be taken during the cycle. If the temperature interval is set at 0 then TD\$ is set to tell the computer not to take data during the cycle, otherwise TD\$ is always set for taking data.

The test profile is stored in the array TC. Each element contains the cycle type corresponding to the cycle number.

The sample descriptions are used for the base file name. The starting cycle number will be one unless previous tests were run on the sample. If previous tests were run the starting cycle number is added to the current cycle number to get the cycle number the file is saved under (see the introduction for an explanation of the file names). For example if the starting cycle number is 10 the first cycle will be stored under cycle 10 and the second under 11 and so on.

When all of the data are entered the computer begins the test. There are two parts to each cycle, cooling and warming. During both portions of a cycle the same type of control cycle is performed. The temperature interval of the next reading is set and the temperature is continuously monitored. When the temperature passes the interval a conductance reading is taken and both the temperature and conductance are stored in the array SDT. The temperature of the interval is checked to see if it is past the end temperature for the current portion of the cycle. If the interval is past the end temperature then that portion of the cycle is completed. If the interval is not past the end temperature, then the next interval is set and the control cycle is repeated. While cooling a reading is taken when the temperature is lower than the temperature interval set and the cycle ends when the interval is less than the minimum temperature for the cycle. While warming, the temperature must be greater than the interval and the interval must be greater than the maximum temperature for the cycle.

The temperature of the cooling bath is controlled throughout the cycle. The bath is either cooled or warmed at the rate of temperature change specified. The temperature is controlled by sending a number between 0 and 255 to the temperature controller card. The card then converts the digital input to a voltage output that is sent to the cooling bath (for more information see the hardware section). A 0 represents 7 °C and 255 represents -35 °C. The temperature can be controlled in integer steps between these two values for a resolution of .164 °C/integer. For example 0 °C would be set by sending a 42.6 but really either a 42 or 43 would be sent for .112 or -.052 °C respectively. At the beginning of either portion of the cycle the starting and ending temperature points are set and the time (in seconds) between integer steps is calculated. If a cycle has end points, for example, of 4 and -16 °C, the corresponding integers would be 18 and 141. If the rate of temperature change was 20 °C/hr then the time between integer steps would be

$$(4 - (-16))\text{ }^{\circ}\text{C} / 20\text{ }^{\circ}\text{C/hr} = 1\text{ hr} * 3600\text{ sec/hr} = 3600\text{ sec}$$

$$3600\text{ sec} / (141 - 18)\text{ steps} = 29\text{ sec} / \text{step}.$$

Each time the sample temperature is read the time is checked to see if it is time for the next temperature step. The bath temperature will continue to change until the end of the cycle even if it passes one of the end points. The bath would only stop if one of the limiting end points (7 or -35 °C) is reached.

The clock for controlling the rate of temperature change is produced by the temperature controller card. An machine language program, Clock, is loaded from the program disk converts one of the timers on the temperature controller card to a clock.

The temperature of the sample is measured by a thermistor. The variable resistance of the thermistor is converted to variable frequency by an square wave oscillator. The period of the square wave is then measured by the interface card using a machine language program, Temp Timer, loaded from the program disk. The length of the period is then read by the program. The period is converted back to a temperature by linear interpolation of a temperature table read from the CAL file on the program disk.

The conductance is measured by the Solomat conductance meter. The meter's reading is transmitted in parallel to the interface card. The conductance may then be read directly from the interface card by the program.

When the cooling portion of a test cycle is completed the data are saved to the disk, the temperature controller parameters are set for the warming portion, and the controlling loop is started. At the end of a warming portion there is one extra thing checked. The data is saved, but then the cycle number is checked to see if the test is completed. If there are still cycles left to complete, the next cycle type is loaded, the temperature controller parameters for the cooling portion are set, and the controlling loop is started.

At the end of a test the program returns to the Main Menu.

PLOTTER, [Plot], (Main Menu, HP Plotter), (general sample file, V, PCV, DVDP, DVR, DSR, S)

PURPOSE : Plotter allows the user to load a sample's data files from any of the directories and plot the data on the computer screen. The plotting parameters may be changed. Plotter may also pass the data to HP Plotter to make a plot of the data on a Hewlett Packard plotter.

OPERATION : Plotter first prompts the user to enter the directory to plot from. The menu below will appear.

MENU OF SUBDIRECTORIES

1. X.Y
 2. X.DYDX
 3. X.Z
 4. X.VP
 5. R.VP
 6. R.DVPDR
 7. R.DVPDLR
 8. EXIT TO MAIN MENU
- ENTER NUMBER OF YOU CHOICE

Choose a type of data. If 6 or 7 is selected the user has the option of plotting the radius on a log scale. The plotting menu will then be displayed (a sample is shown).

Enter the letter of any value that you wish to change
and press RETURN.

1.	Title	Concrete Mortar	27 March 87
2.	Filename	Mortar.25W	
3.	Label	Warming	
4.	Linetype	7	
5.	Filename	Mortar.25C	
6.	Label	Cooling	
7.	Linetype	0	
8.	Minimum X	10	
9.	Maximum X	10000	
10.	X-Interval Size	10	
11.	Minimum Y	0	
12.	Maximum Y	1	
13.	Y-Interval Size1	
14.	Paper Length	8.5	

A - Onscreen Plotting	D - Clear Graphics Screen	False
B - HP722 Plotting	E - Exit Program	
C - View Graphics Screen	F - Find New Data Points	

Enter your choice and press RETURN.

Choices from this menu may either load data files, change plotting parameters, or choose an action. Two files may be plotted at the same time, usually the warming and cooling data from the same sample and cycle. To load a file enter 2 or 5, and enter the complete file name for the data. When a file name is entered, the computer does not load the file into an array but instead loads it during plotting. The label and the type of line associated with each file may be changed by choosing 3, and 4 or 6, and 7.

The other parameters are for adjusting the X and Y axis. The minimum and maximum value for each axis scale and also the spacing between scale markings may be adjusted. If the X axis is radius and a log scale is chosen, the interval for the X axis is not used so changes to the interval will have no effect on the plot. The paper size is normally set for the smaller plotters and 8.5 X 11 paper. If a larger plotter is used the paper length may be changed to 14 for 11 X 14 paper.

Now that the plotting parameters and file names have been entered, the data may be plotted. It is recommended to first plot the data on the computer screen to insure the parameters are set correctly before starting a more time consuming plot on a plotter. If a "TRUE" appears after "D - Clear Graphics Screen", the graphics screen will be cleared before the new data is plotted. To plot a second set of data on top of the previous plot, make sure

it says "FALSE". At any time after the axis are drawn pressing the ESC key will discontinue the plot. The computer will stop plotting and by pressing another key, the plotting menu will reappear. This escape feature only works while plotting on the computer screen.

To make the plot on paper, choose B to transfer the data to HP Plotter. HP Plotter will plot the data on the plotter and then return the data to Plotter.

To load a different sample or a different directory of files from the same sample, choose F. The program will start over by prompting the user to select a file directory.

Select E to return to the Main Menu.

STRUCTURE : The user is first prompted to select the directory of the files to plot. When the file name is entered (must include sample name, cycle number, and portion of the curve) data isn't loaded but it is ready to be loaded during plotting. Two files may be loaded and plotted at the same time. The top file is plotted first followed by the lower.

The plotting parameters set the X and Y axis. If a plot with a log scale X axis is selected the X interval is not used by the computer. The computer automatically sets up a log scale plot from the minimum X to the maximum X, but if either of these values are not a power of ten the computer rounds down to the next lowest power of ten. For example if the minimum X is set at 15 and the maximum is set at 15000 the computer will plot from 10 to 10000 with a log scale.

Line types for each of the files may be chosen from the following list.

Pattern #	Line Pattern
0	only the data points are plotted
1
2	— — — — —
3	— — — — —
4	— . — . — . — . — . — .
5	— — — — —
6	— — — — —
7	— — — — —

→| |← one pattern length

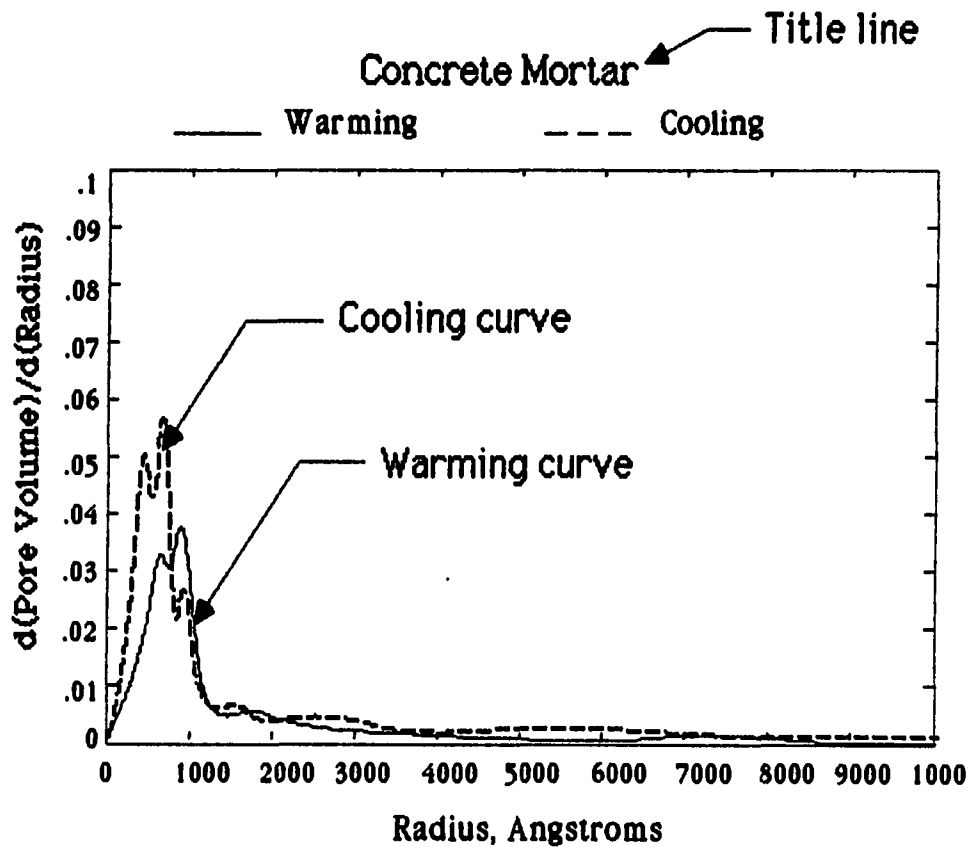
The arrays are passes to HP Plotter when a hard plot is desired.

HP PLOTTER, [Hpplot], (Plotter), (none)

PURPOSE : HP Plotter receives the parameters passed from Plotter and makes a plot on a plotter.

OPERATION : HP Plotter requires no user input.

STRUCTURE : The file names and parameters are passed to HP Plotter by Plotter. HP Plotter then initializes the Plotter and begins to draw the plot. A sample plot is shown below.



CONVERT XY, [T.C.to.X.Y], (Main Menu), (raw data file, X.Y)

PURPOSE : Convert XY reads in the raw temperature vs. conductance data from the disk and converts it to the X vs. Y format for further processing.

OPERATION : Convert XY requires no user input.

STRUCURE : Convert XY is the first in a string of data processing programs. Therefore Convert XY is responsible for loading the data from the disk and converting it into the format required for the rest of the processing programs. The Main Menu sends the file name to Convert XY. Convert XY loads the raw data from the warming and cooling files into the two-dimensional arrays A and B for the temperature and the conductance respectively. The first dimension of the array indicates whether the data are from the warming (0) or cooling (1) portion of a cycle. The second element is the numerical value of the data.

The raw data are converted to the X vs. Y format where

$$X = (-1 / T) * 10000$$

and T is the temperature in degrees Kelvin. During the conductance test you a resistor may be connected in series with the sample to narrow the conductance range of the test. This allows a lower scale on the conductance meter to be used and thus resolution is better at lower temperatures where low conductances occur. During the calculation of Y values the effect of the conductance of the series resistor is compensated for and the conductance of the sample alone is determined. The following relationships apply.

$$Y = \log (CT), \text{ and}$$

$$CT = (C * R) / (C - R),$$

where R is the conductance of the resistor in μS , and C is the conductance of the resistor and sample in μS . With the effect of the resistor is removed, CT is the conductance of the sample.

The new X vs. Y data are stored in the X.Y directory and the data are passed back to the Main Menu for further processing.

DERIVATIVE XY, [Der.X.Y], (Main Menu), (X.DYDX)

PURPOSE : Derivative XY calculates the derivative of Y with respect to X, dY/dX , for both the warming and the cooling data. The derivative algorithm also smooths the data using the moving average set in the Main Menu. The relationship between the derivative, dY/dX , and the samples pore volume is expressed in equations 7 and 8 in the Introduction.

OPERATION : Derivative XY requires no input by the user.

STRUCTURE : The data are passed from Main Menu through arrays A and B for X and Y respectively. The derivatives, dY/dX , are calculated and stored in the conductance array, therefore the conductance data in the array are lost, but the data are still in the disk file. The derivative is found with an algorithm that uses a moving average to smooth the data somewhat. The moving average (MA) is set in the Main Menu. If the user has chosen to save the data, they are stored in the X.DYDX directory. When the program is finished the data are passed back to the Main Menu for further processing.

INTEGRATE XDYDX, [Int], (Main Menu), (XZ)

PURPOSE : Integrate XDYDX perform 2 integrations on the X, DYDX data, resulting with the calculation of the parameter Z expressed in equation 8 of the Introduction.

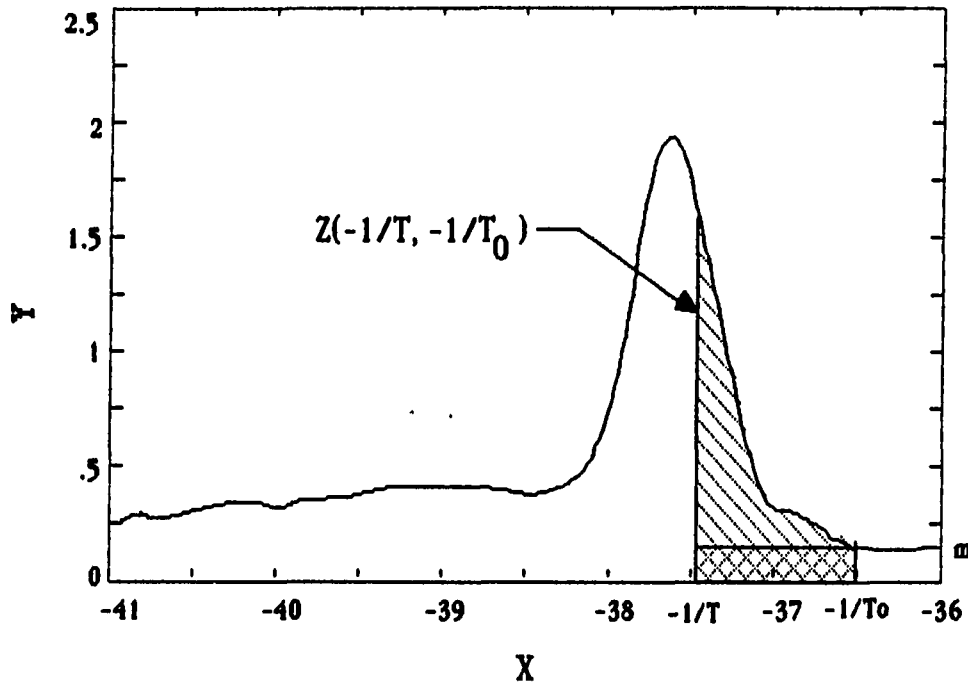
OPERATION : Integrate XDYDX requires no user input.

STRUCTURE : The arrays A and B are passed from the Main Menu containing the information X and dY/dX respectively. The Z values are cumulative totals of the integral and replace the dY/dX values in the array B.

The warming file is processed first with the data pairs processed sequentially from right to left on a X vs. dY/dX plot (the larger X value pairs first). The Z value for all pairs with an X greater than -36.5 are set equal to 0. The average value of Y on the dY/DX interval from -36 to -36.5 is found and stored in Y1 as the slope m. For the rest of the file the Z value is calculated using the following formula:

$$Z\left(\frac{-1}{T}, \frac{-1}{T_0}\right) = \underbrace{\int_{\frac{-1}{T}}^{\frac{-1}{T_0}} \frac{d(\log C)}{d\left(\frac{-1}{T}\right)} d\left(\frac{-1}{T}\right)}_{\text{where } \textcircled{\text{---}}} - \underbrace{\int_{\frac{-1}{T}}^{\frac{-1}{T_0}} m d\left(\frac{-1}{T}\right)}_{\text{where } \textcircled{\text{---}}}$$

The calculations are graphically shown on the graph below.



The cooling portion is processed next, in a left to right manner. The same base m is used on the cooling curve. The data pairs with a dY/dX value greater than -36.5 receive the Z value of the last pair less than -36.5 . When the cooling curve is complete the data are passed back to the Main Menu and stored in the directory $X.Z$ if the user has directed the computer to save the $X.Z$ File in the Main Menu.

CONVERT VP, {X.VP.from.X.Z}, {Main Menu}, (X.VP)

PURPOSE : Convert VP calculates the total conducting pore volume from the X vs. Z data. The total conducting pore volume is the ratio of V_{cp} / V_p as shown in equation 10 in the Introduction.

OPERATION : Convert VP requires no user input.

STRUCTURE : The data are passed into Convert VP via the arrays A and B for X and Z respectively. The pore volume is found using the equation,

$$V_p = 1 / 10 Z$$

and replaces Z in the B array. If the user chooses to save the data, they are stored in the X.VP directory. The data are passed back to the Main Menu when the program has completed the calculations.

CONVERT R, [R.VP.from.X.VP], (Main Menu), (R.VP)

PURPOSE : Convert R changes the X variable to a pore radius. Relative conducting pore volume is now expressed as a function of radius, as shown on the left hand sides of equations 9 and 10 in the Introduction.

OPERATION : Convert R requires no user input.

STRUCTURE : The data are passed to Convert R via two arrays, A and B containing X and VP respectively. The X data are converted from a temperature to radius using the equation

$$R = \frac{-2 * T_0 * (.25 - (29 * X) + (.25 * T_0 * X))}{(.9998 * 3.335 * 10^9) * (1 + T_0 * X)} * 10^8$$

where T_0 is the pore solution melting point (273.15 °K for water) which is entered in the Main Menu and X is the X variable. All data points with an original temperature greater than 0 °C are meaningless and removed from the arrays. If you have chosen to "thin" the file, the program only saves points that are spaced at least one Angstrom apart. The data are saved in the R.VP directory on the data disk and then passed back to the Main Menu for further processing.

DERIVATIVE VP, [Der.R.VP], (Main Menu), (R.DVPDR)

PURPOSE : Derivative VP calculates the derivative of V_p with respect to R , dV_p/dR , for both the warming and the cooling data. The derivative algorithm also smooths the data using the moving average entered in the Main Menu.

OPERATION : Derivative VP requires no input by the user.

STRUCTURE : The data are passed from Main Menu through arrays A and B for R and V_p respectively. The derivative, dV_p/dR , is calculated but not stored in the conductance array. The arrays are passed back to the Main Menu without any changes so that the derivative with respect to the $\log(R)$ can be computed. The derivative is found with an algorithm that uses a moving average to smooth the data somewhat. The moving average (MA) is set in the Main Menu. The data are stored in the R.DVPDR directory. When the program is finished the R and V_p data are passed back to the Main Menu for further processing.

DERIVATIVE LVP, [Der.LR.VP], (Main Menu), (R.DVPDLR)

PURPOSE : Derivative LVP calculates the derivative V_p with respect to $\log(R)$, dV_p/dLR , for both the warming and the cooling data. The derivative algorithm also smooths the data using the moving average entered in the Main Menu.

OPERATION : Derivative LVP requires no input by the user.

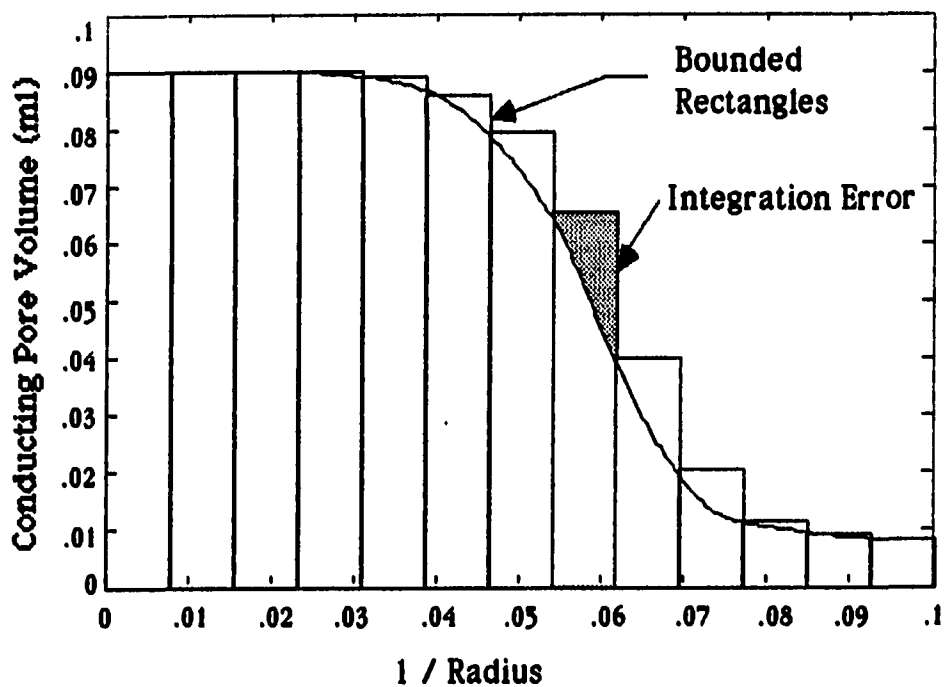
STRUCTURE : The data are passed from Main Menu through arrays A and B for R and V_p respectively. The radius is converted to the $\log(R)$ before any of the calculations and then the derivative of V_p is computed with respect to the $\log_{10}(R)$. The derivative, dV_p/dLR , is calculated and stored in the conductance array. The derivative is found with an algorithm that uses a moving average to smooth the data somewhat. The moving average (MA) is set in the Main Menu. The data are stored in the R.DVPDLR directory. This is the last step in the data processing cycle and the program returns to the Main Menu when finished.

SURFACE AREA, [Int.Sa], (none), (R.VP)

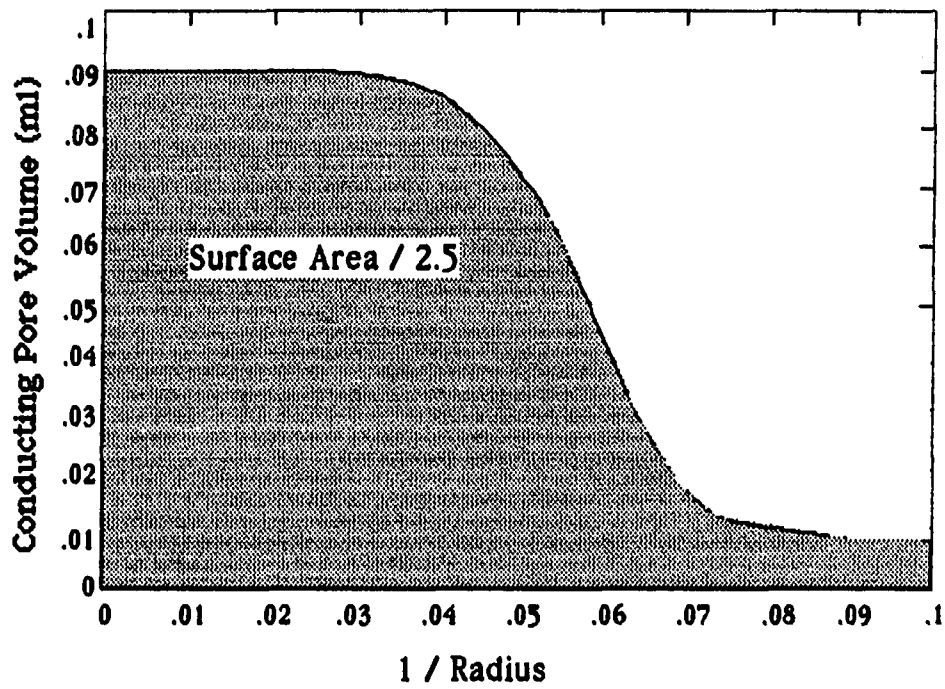
PURPOSE : Surface Area loads the radius vs. relative pore volume data and after the actual sample pore volume is entered calculates the surface area of the sample.

OPERATION : Surface Area is not offered as an option on the Main Menu, therefore to load and run the program, first exit the menu and type "RUN INT.SA,D1". Once the program is started, input a file name consisting of the base file name for the sample and the cycle number. After the R.VP file for the sample is loaded for the warming portion, enter the pore volume of the sample. The relative pore volume is multiplied by the actual total pore volume to get the pore volume at any given point. The computations will be performed and the surface area printed on the screen.

STRUCTURE : Surface area computations are only performed on the warming portion of the curve. Once the file name is entered, the pore volume data from the warming portion is loaded into an array for processing. Then a modified version of the data, changed to pore volume vs. $1/\text{radius}$, is integrated from 0 to $1/12$ using a method of rectangles. The change in $1/\text{radius}$ between points is multiplied by the height of the larger pore volume to get the area of a rectangle. The total integral is the sum of all of the rectangles. The area will be over estimated using this method but as the number of points increases and therefore the width of the rectangle decreases, the error becomes negligible. The integration method is illustrated on the next page. The width of the rectangles is exaggerated for clarity.



The surface area of the sample (excluding pores less than 12 Angstroms) is found to be the area under the curve multiplied by 2.5. The surface area is graphically represented on the next page.



The surface area is then printed to the screen and the program ended. To return to the Main Menu type "RUN STARTUP,D1"

TEST FORMAT, [Sample.Data], (Main Menu), (general sample file)

PURPOSE : Test Format prints out the test format, the cycle types and the cycles, for a test sequence.

OPERATION : To run Test Format enter a "D" in the Main Menu. Test Format will be loaded and the computer will prompt the user to insert the appropriate raw data disk. Enter the test format file name including the sample name, starting cycle number, and CT (for example Mortar.25.CT). The file will be loaded into memory and displayed on the computer screen. The user may choose to either quit and return to Main Menu or print the information on the plotter.

STRUCTURE : The data is loaded into an array and displayed on the screen. Due to limitations of screen size only 48 cycles may be displayed, the cycles over 48 will be loaded from the disk but not be shown on the screen. The program sends the data to the plotter.

CLOCK, [IO.Clock.Slot1], (Controller), (none)

PURPOSE : Clock generates a system clock that is accessed by Controller to maintain the rate of temperature change.

OPERATION : Clock requires no user input.

STRUCTURE : The operation of Clock is described in detail in Byte, March 1982 in an article by Ned Rhodes, the author of the program. The clock uses one of the 16-bit timers on the parallel interface card located in slot 1. The timer generates an interrupt every 1/60 of a second and calls Clock. Clock then displays the time on the screen and maintains a register which can be read by the program. The number in the register represents the number of interrupts since the clock was reset. Controller then divides the number in the register by 60 to get the number of seconds.

TEMP MEASUREMENT, [Cap.Res.2], (Controller), (none)

PURPOSE : Temp Measurement determines the period of the square wave generated by the thermistor's oscillator. The period is used to determine the temperature of the sample.

OPERATION : Temp Measurement requires no user input.

STRUCTURE : Temp Measurement uses one of the 16-bit timers from the parallel interface card located in slot 1. The program turns the timer on when it senses a rising edge on the signal. The timer then counts down until a falling edge is detected. The computer can then access the counter and subtract the initial value to get the half period of the wave.

CALIBRATE, [Calibrate], (none), (Cal)

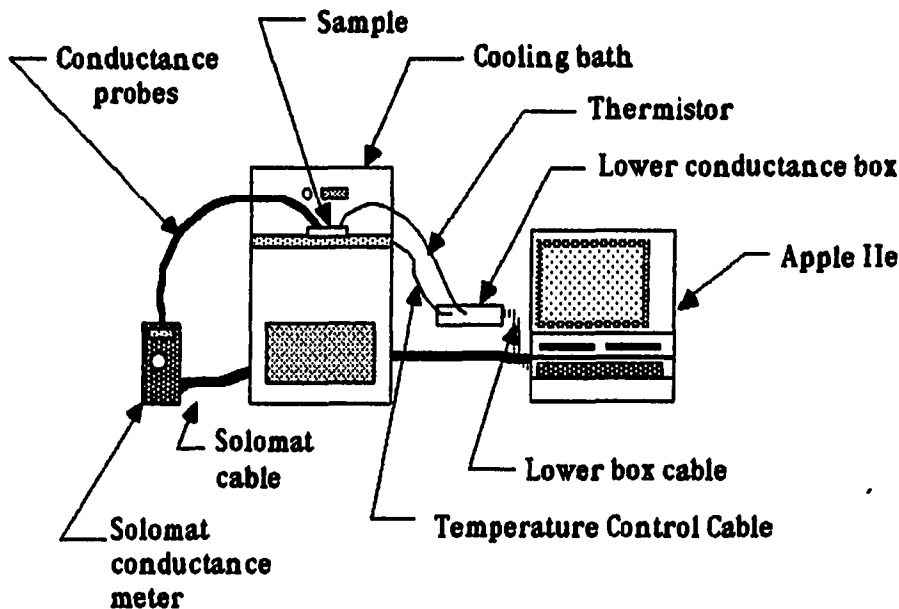
PURPOSE : Calibrate is a data editing program to allow the user to change the temperature calibration data for the thermistor.

OPERATION : Calibrate is not an option on the Main Menu. To load and run the program exit the Main Menu and type "RUN CALIBRATE,D1". The program will automatically load the Cal file and display it on the screen. The method of determining the calibration data is described in the Appendix. By using the arrows the cursor can be moved around the screen. When the cursor is at the desired location, type in the new data and press RETURN. The new data will then replace the old in the file located in the computer. When editing is completed the new version of the file in the computer may be saved to the disk, by pressing CTRL and S. If the user doesn't want the changes made to the data to be save pressing CTRL and Q will exit the program without saving the changes (NOTE : the data in the disk file is NOT changed in this case). In both cases you are returned to the Main Menu when finished.

STRUCTURE : The Cal file is automatically loaded into an array when the program is started. The user is then allowed to change the data in the array by moving the cursor on the screen to the data to be changed and entering the new data. The number of data points in the file is set at 22 and may not be changed using this program. When editing is finished the data is either written back to the file on the disk or discarded. The program returns to the Main Menu when completed.

IV. INTERFACE

The Conductance Porosimeter system consists of a Solomat conductance meter, an Apple IIe (with 2 disk drives, a monitor, and 2 parallel interface cards), an interface box, and a cooling bath. This section will mainly discuss the interface box but will also look at some of the aspects of the rest of the system. The main components are shown below.



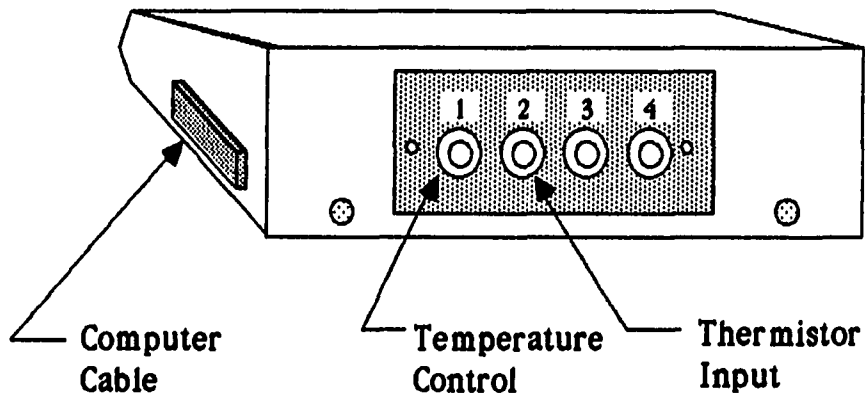
The Solomat measures the conductance of the sample and sends the digital data to the parallel interface card located in slot 4 of the Apple IIe. The temperature is converted from the variable resistance of the thermistor to a variable frequency square wave. The period of the square wave is found by a timer on the parallel interface card located in slot 1 of the Apple. Control of the cooling bath is accomplished by the parallel interface card in slot 1 sending a digital number to the temperature control circuit in the interface box. The temperature control circuit converts the digital data to a voltage level which is sent to the cooling bath. The voltage level represents the temperature level of the bath.

The following sections will take a closer look at each of these main functions of the interface.

INTERFACE BOX

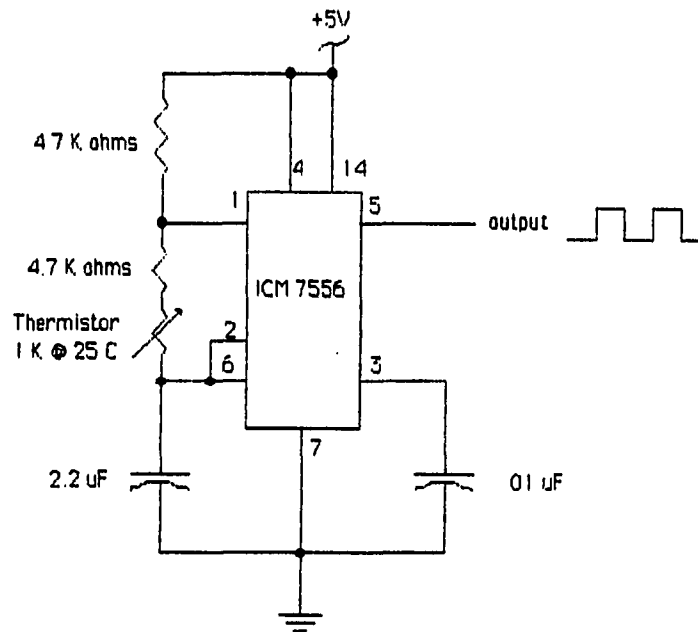
PURPOSE : The interface box provides the circuitry to control the cooling bath and record temperature readings.

OPERATION : The interface circuits are housed in a low profile box with connections for the computer cable, the thermistor, and the temperature control cable. The cables should be connected as shown.



STRUCTURE : There are two main components of the interface box, the thermistor and temperature control circuits. The thermistor circuit converts the variable resistance of the thermistor to a variable frequency square wave. The temperature control circuit converts the digital output from the computer to a voltage level that is sent to the cooling bath.

The main component of the thermistor circuit is a 556 timer chip. The chip is wired as a square wave oscillator with its frequency dependent on the capacitance and resistance on pins 6, 1 and 4. Since the resistance of the thermistor is dependent on temperature, and the frequency of the oscillator is dependent on resistance, the frequency is a function of the temperature of the thermistor. The square wave is sent to the parallel interface card in slot 1. The thermistor circuit is shown on the next page.

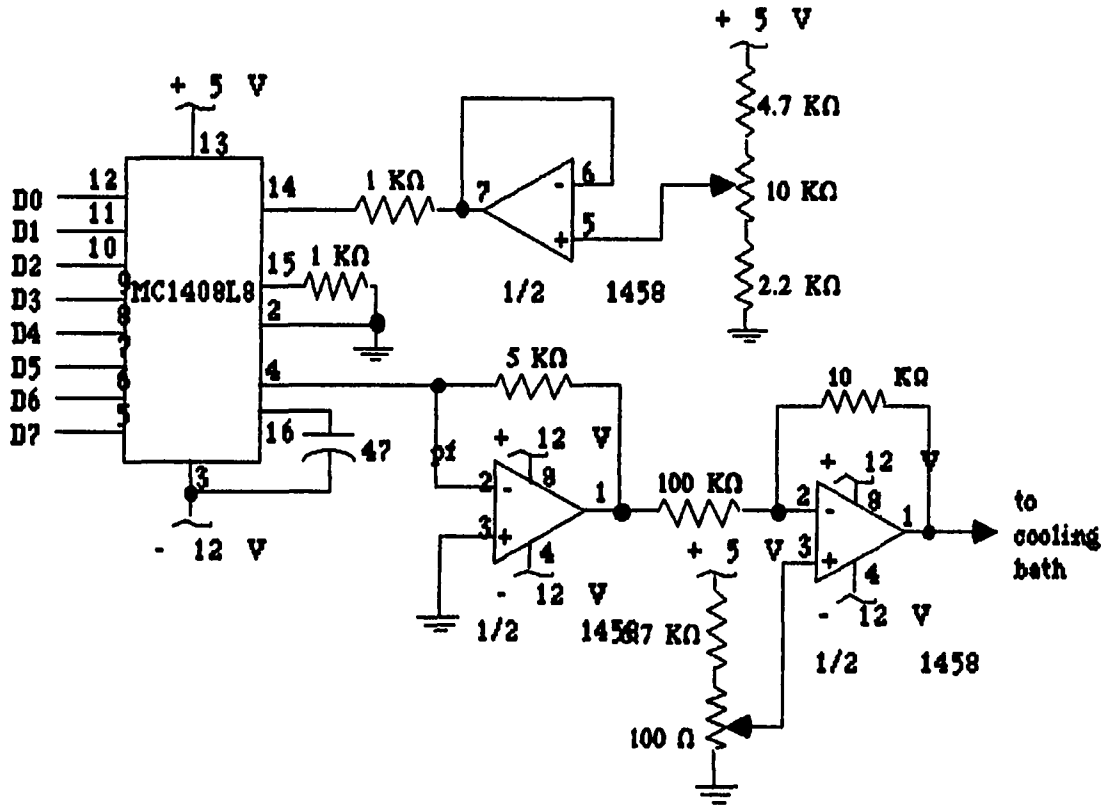


The temperature control circuit consists of a Digital-to-Analog (D/A) converter and three operational amplifiers (op amps). The temperature control output from the computer is in the form of a number from 0 to 255 corresponding to 7 and -33°C respectively. The 8-bit number is transferred through the parallel interface card in slot 1 of the Apple. Lines PA0-7 are used to represent the 8 bits (see the parallel interface card section). The 8-bit number is converted to an analog voltage output by the D/A converter (MC1408L8).

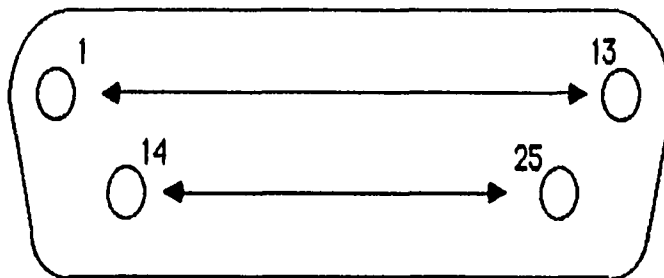
Adjusting the voltage on pin 14 of the D/A converter controls the range of voltage from the maximum to minimum inputs. The voltage on pin 14 is controlled a the voltage divider and op amp. Adjusting the 10 K Ω trim pot changes the voltage sent to the op amp. The op amp is configured as a voltage follower (just maintains the voltage input) and is connected to pin 14.

The D/A converter outputs a voltage on pin 4. The output is then amplified by the first op amp and reduced by the second one in line. The 100 Ω trim pot controls the offset voltage of the second op amp. The offset voltage is used to move the output voltage range to the proper level. The output should be .07 to -33 V corresponding to 7 and -33°C respectively. In practice, the 10 K Ω trim pot is used to set the low voltage and the 100 Ω for the high voltage (see the Appendix for calibrating the circuit).

The output is then connected by cable to a socket on the back of the cooling bath. The format of the socket may be found in the cooling bath's manual. The schematics of the temperature circuit are shown on the next page.



The cable connecting the interface box to the computer is a DB 25 connector configured as follows. The cable is connected to the other DB 25 connector on the back of the Apple.



Pin #	Name	Description
1	+ 5 V	Power supply
2	+ 12 V	Power
3	- 12 V	Power
4	GND	Common or Ground
5	PA0	Data line 0 for temperature control
6	PA1	" "1 "
7	PA2	" "2 "
8	PA3	" "3 "
9	PA4	" "4 "
10	PA5	" "5 "
11	PA6	" "6 "
12	PA7	" "7 "
13-21	Unused	
22	CA1	Square wave output of temp oscillator
23-25	Unused	

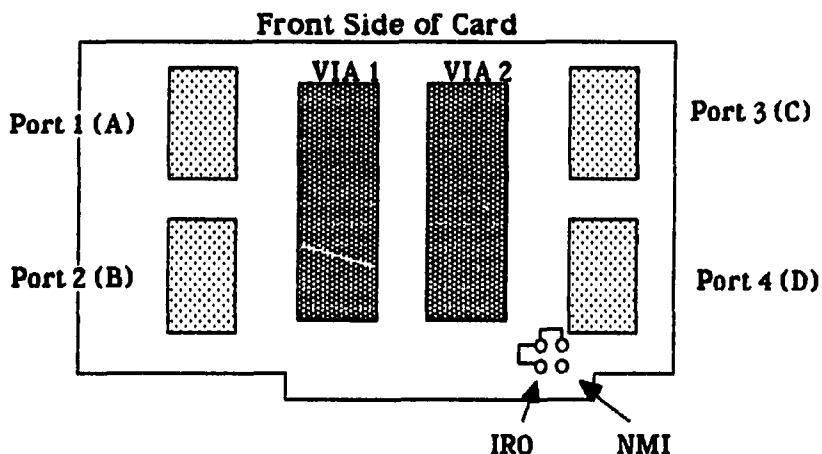
PARALLEL INTERFACE CARD

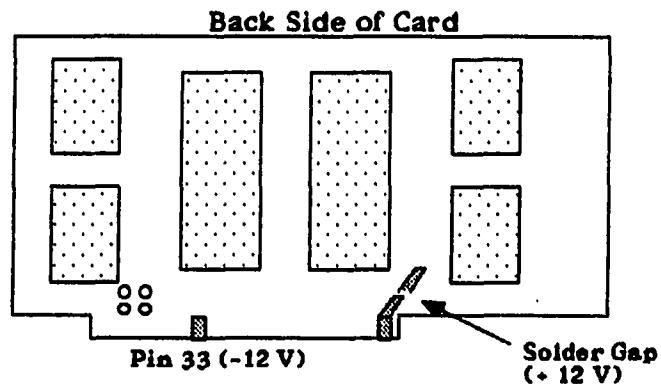
PURPOSE : The parallel interface card is used to input and output data in a parallel fashion (simultaneously on many lines rather than serially one bit at a time on one line), provide the system clock, and measure the period of the temperature signal square wave.

OPERATION : The parallel interface cards require no user intervention other than attachment of cables.

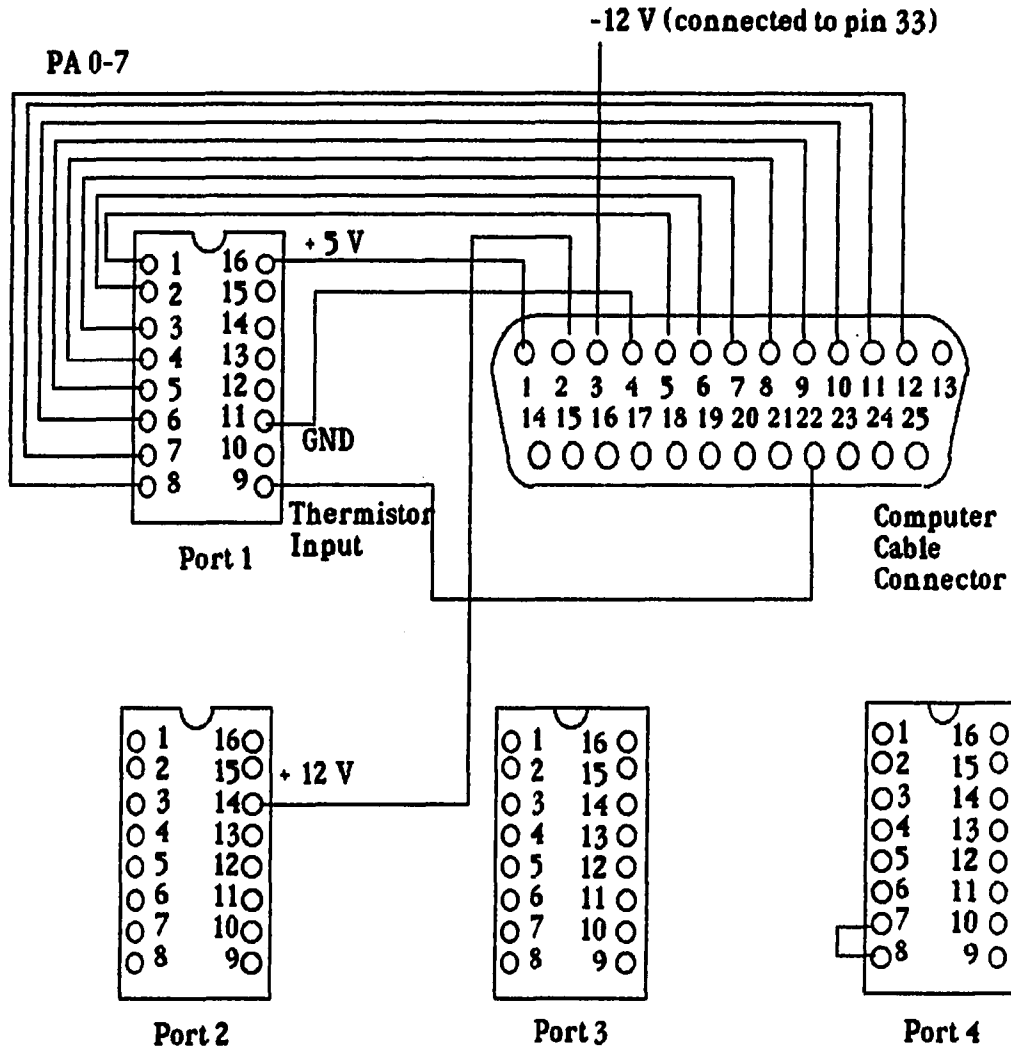
STRUCTURE : The parallel interface card used in this system is supplied by John Bell Engineering, 1014 Center St, San Carlos, CA 94070 (Tel: 415-592-8411). The card consists mainly of two 6522 Versatile Interface Adapters (VIA) with a total of four 8-bit bidirectional I/O ports and four independent 16-bit timers. More information may be found in the manuals supplied by the manufacturer.

Two parallel interface cards are used in the system. One of the cards in slot 1 of the Apple, controls the cooling bath, supplies the system clock, and measures the period of the thermistor's square wave. The other is located in slot 4 and is used to interface the Solomat conductance meter.



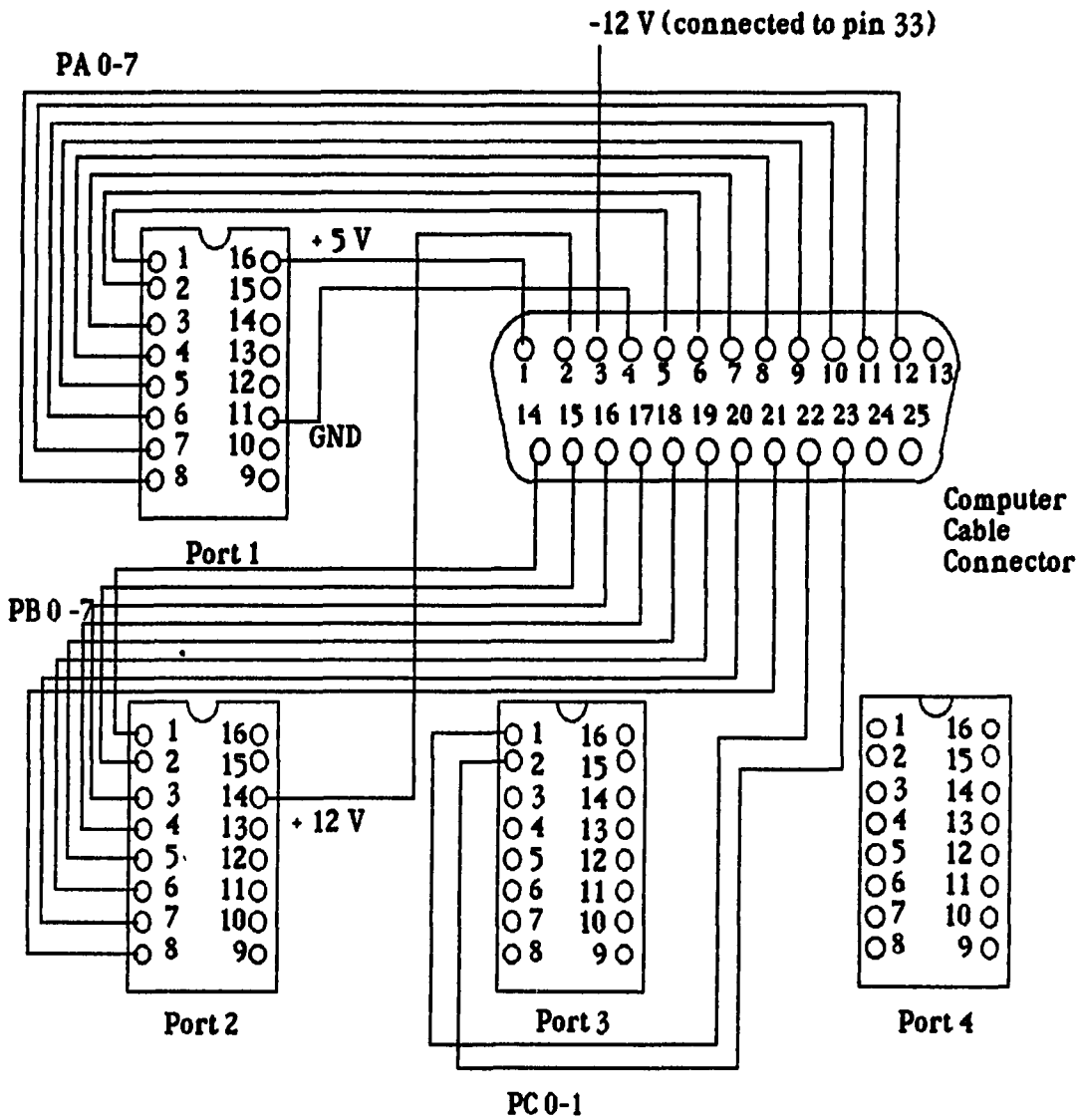


The card located in slot 1 generates the system clock through a program published in *Byte*, March 1982 by Ned W. Rhodes (Clock). Using the Apple's interrupts one of the 16-bit timers is triggered every 1/60 of a second. The program then reads the time from the timer. This is why the IRQ and NMI must be configured correctly. The period measurement for the thermistor is accomplished by triggering a timer on the rising edge and stopping on the falling edge. The timer counts down while operating and the program reads the timer. The temperature control data is simply passed in parallel through port 1. The connections are as shown, note the connector on port 4 and the fact that the -12 V line must be directly connected to pin 33.



Parallel Interface in Slot 1

The parallel interface card in slot 4 is used exclusively for interfacing with the Solomat. The Solomat is powered by the 5 volt line from the parallel interface card. The data is then sent back in parallel through ports 1, 2, and 3. The connections should be made as shown.



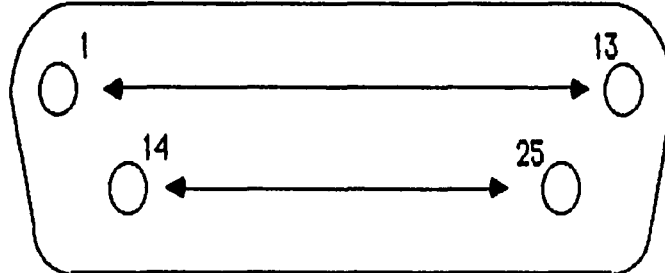
Parallel Interface Card in Slot 4

SOLOMAT CONDUCTANCE METER

PURPOSE : The Solomat conductance meter measures the conductance of the sample and sends the digital data to the Apple.

OPERATION : The conductance cables should be attached to the probes in the sample and the cable from the computer plugged into the socket on the side of the meter. The conductivity scale should be set to 16000 μS for the 1.0 cell unless a series resistor is used. A resistor is sometimes put in series with the sample to give better resolution at the lower end of the conductance cell. If a series resistor is used set the scale to 160 μS again for the 1.0 cell. The meter is powered by the computer and doesn't need to be switched on.

STRUCTURE : The Solomat is described in detail in the manufacturer's manual. The configuration of the socket is described in the section 6. All five digits are passed through the parallel interface. Power is supplied through pin 19. The configuration of the DB 25 connector on the back of the Apple is shown below. The digits are sent in Binary Coded Decimal (BCD) requiring 4 bits per digit except for digit 5 (digit 1 being the least significant) which only requires 1 bit as it can only be a one or a zero.

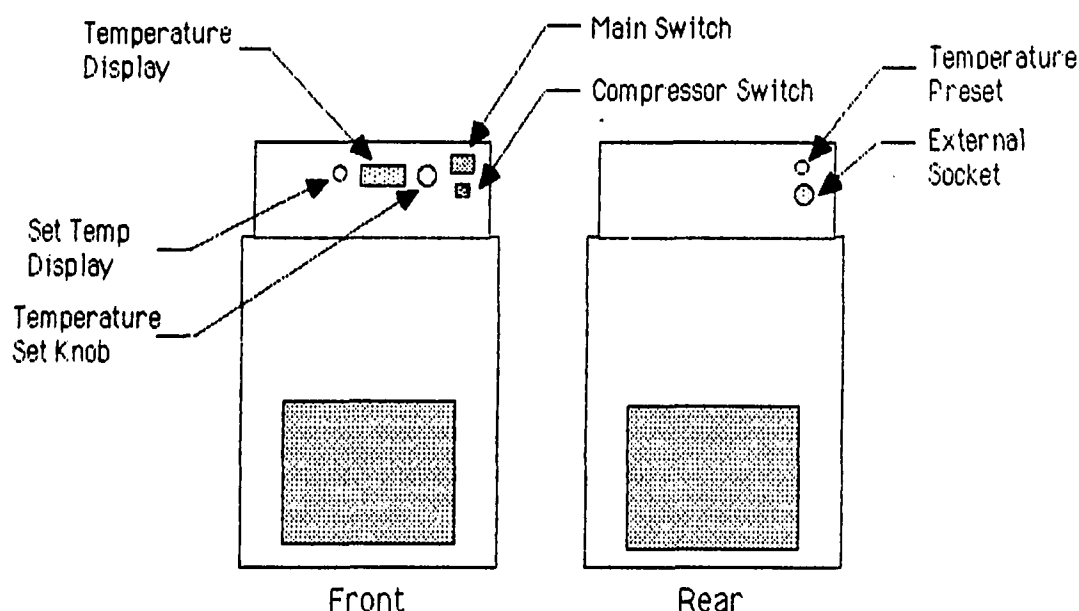


Pin #	Name	Description
1	+ 5 V	Power supply
2	+ 12 V	Power
3	- 12 V	Power
4	GND	Common or Ground
5	PA0	Digit 1 Bit 0
6	PA1	Digit 1 Bit 1
7	PA2	Digit 1 Bit 2
8	PA3	Digit 1 Bit 3
9	PA4	Digit 2 Bit 0
10	PA5	Digit 2 Bit 1
11	PA6	Digit 2 Bit 2
12	PA7	Digit 2 Bit 3
13	GND	Common or Ground
14	PB0	Digit 3 Bit 0
15	PB1	Digit 3 Bit 1
16	PB2	Digit 3 Bit 2
17	PB3	Digit 3 Bit 3
18	PB4	Digit 4 Bit 0
19	PB5	Digit 4 Bit 1
20	PB6	Digit 4 Bit 2
21	PB7	Digit 4 Bit 3
22	PC0	Digit 5 Bit 0
23-25	Unused	

COOLING BATH

PURPOSE : The cooling bath controls the temperature of the sample and allows the computer to cycle the sample through a freeze-thaw cycle while monitoring the conductance of the sample.

OPERATION : The controls of the cooling bath are shown below.



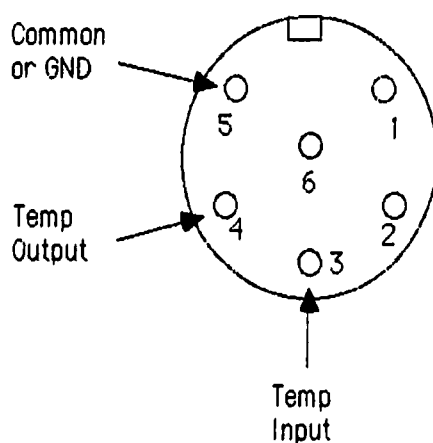
To turn on the bath, first place the main switch in the on position and then turn the compressor on. Turning them both on at the same time causes a large power surge on the power line. When turning the bath off turn off the compressor before turning the main power off.

The bath's display serves two purposes. It usually displays the current temperature of the bath, but when the set temp display button is pressed the display shows the temperature the bath has been set at. To set the bath temperature press the set temp display button and adjust the temperature set knob until the desired temperature is shown.

The external control socket is located on the rear of the bath. The cable from the temperature control circuit should be connected to the external socket. To place the bath under external control place the temperature preset switch in the ON position. The temperature of the bath will now be set by an external device, in this case the computer. A small mark will appear in the upper left hand corner of the bath's display indicating external

control. The temperature the external device is setting may be shown on the bath's display by pressing the set temp display button.

STRUCTURE : The cooling bath contains a refrigerating unit and a circulator for the bath's fluid. The bath currently contains isopropyl alcohol. The bath is a Haake model A 82 supplied by Haake Inc., 244 Saddle River Road, Saddle Brook, NJ 07662. For further information of the bath see the manufacturer's manual. The format of the external socket is as shown below.



The temperature voltage is sent into the bath using pin 3. The output pin is not used but could be used to externally monitor the bath's temperature. Both the output and the input represent temperature with voltages, 0.00 volts corresponding to 0 °C and 1.00 volts for 100 °C. Therefore for our range of 7 and -33 °C, we use .07 and -.33 volts respectively.

THERMISTOR

PURPOSE : The thermistor is a temperature dependent variable resistor used to measure the temperature of the sample.

OPERATION : The thermistor requires no intervention.

STRUCTURE : The thermistor used in model SP85DA102FA1 supplied by Thermometrics, 808 U.S. Highway 1, Edison N. J. 08817. It is .085" in diameter and 1/2" long with a nominal resistance of 1 k Ω at 25 °C and stable to \pm .02% per year.

V. APPENDIX

INITIALIZING DISKS

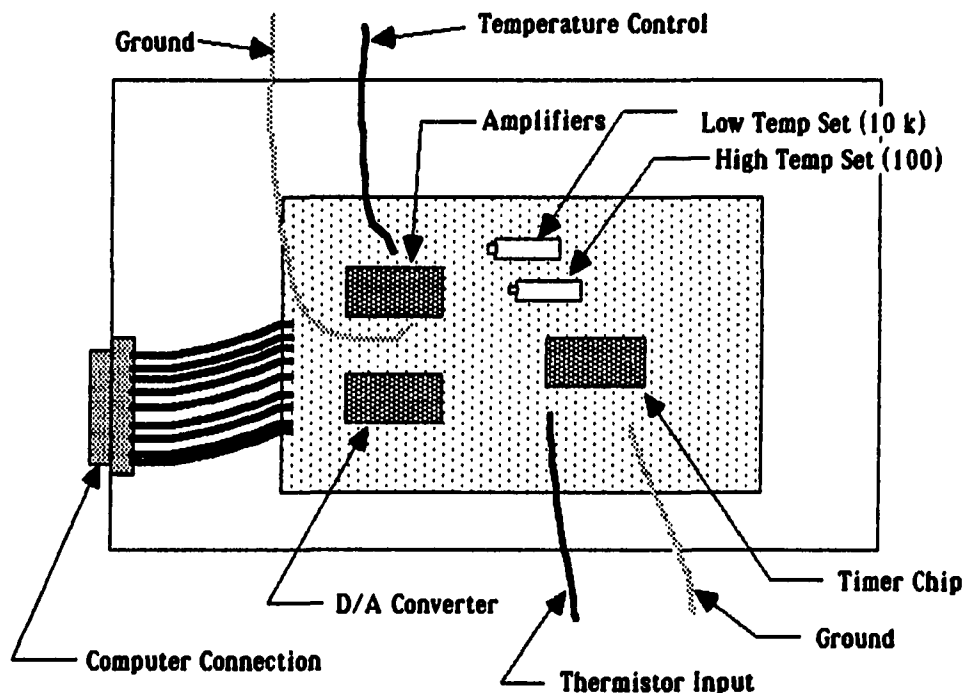
The data disks used by this system need to be initialized and directories established before data may be stored on them. To initialize a disk the user needs the ProDOS User's Disk and the /Caldat master disk. Insert the ProDOS User's Disk in drive one and press the CTRL and open-Apple, and RESET keys to reboot the system. The main menu for the ProDOS User's Disk will be displayed on the screen. Select F for ProDOS Filer (Utilities). This is a set of programs to allow the manipulation of disks. The next menu will be the ProDOS Filer menu. This time select V for Volume commands. This will further define the programs to those that manipulate entire disks rather than single files. When the Volume Commands menu is displayed select C for Copy a Volume. This program will copy the contents of an entire disk to another. This program will be used to copy the contents of the /Caldat master disk to the blank disk. The Copy a Volume screen will be displayed next. The program prompts the user to enter "THE VOLUME IN SLOT : " with a "6" being displayed. The program is asking what slot the disk drive controller card is in which in this case is the default value, slot 6. The copying will be done from drive 1 to drive 2 so answer the prompts respectively. Place the /Caldat master disk in drive 1, the blank disk in drive 2 and press RETURN. The volume name "/Caldat" will appear for the new volume name. Press RETURN to accept this as the new volume name for the blank disk. The program will format the blank disk and proceed to copy the contents from the /Caldat master disk to the blank disk. When the program is finished remove the disks and reboot the system with the porosimeter programs disk or initialize more disks. It is a wise idea to have a number of initialized data disks available.

CALIBRATING TEMPERATURE CONTROL

As described in the interface box section, the temperature of the bath is controlled by sending a number from 0 to 255 representing 7 and -33 °C respectively. The interface converts the number to a voltage level that is sent to the bath. To set the proper voltage levels the adjustment of two trim pots in the interface box will be necessary to send the appropriate maximum and minimum numbers.

The program used to calibrate the temperature control circuit isn't an option on the Main Menu. To run the program exit the menu and type "RUN I.O.CARD.TEST,D1". This program will allow the user to send a number from the parallel interface card to the interface box. The program will prompt the user to enter the port, enter 1 for port 1. Then enter the letter O for output on port 1. Only port 1 needs to be addressed as this is the port used to send the data to the interface box, so type a number 0 to begin the calibration.

The user will now need to remove the cover of the interface box to expose the trim pots that will be adjusted.



Turn the cooling bath on (the compressor doesn't need to be running) and place the external temperature preset switch (see cooling bath section)

to the ON position. By pressing the temperature set button on the front of the bath (and holding it in), the temperature the control circuit is sending in will be shown on the bath's display.

To set the trim pots, alternate between sending a 0 and a 255 to the temperature control circuit and adjusting the appropriate trim pot so that the correct temperature will be displayed on the bath. Enter 1/0 into the computer for port 1 with a 0 output. Then adjust the 100 Ω trim pot (High Temp Set) until 7 °C is shown on the display. Now enter 1/255 and adjust the 10 k Ω trim pot (Low Temp Set) until -35 °C is displayed. Repeat these steps until no adjustments to the trim pots need to be made.

Replace the cover on the interface box. To return to the Main Menu type "RUN STARTUP,D1" and the calibration is completed.

CALIBRATION OF THERMISTOR

New thermistors do not have the same temperature-resistance characteristics and individual thermistors themselves change over time. Therefore the calibration file may need to be updated occasionally. The thermistor circuit converts the variable resistance of the thermistor to a variable frequency square wave. The period of the square wave is then measured by the parallel interface card so that the program may read the period. The program converts the period back to temperature using the table of values in the Cal file.

The Cal file contains 22 period - temperature pairs. The controlling program reads a period and then searches the table. If the period is within the range of the table, the program linearly interpolates between the two closest points of the table to calculate the temperature of the sample. If the period is outside of the boundaries of the table, either the maximum temperature or minimum is assumed accordingly.

To find the 22 data points needed place a thermocouple and the thermistor in a test tube and immerse the tube in the cooling bath. The thermocouple is used to determine the temperature of the thermistor. Exit the Main Menu and type "RUN TEMP.PERIOD,D1" to load a program that will display the square wave's period. Two numbers will be shown on the screen, the top number is the current reading of the thermistor and the lower a moving average of 30 readings.

Turn the bath on and set the temperature at 7 °C. When the temperature of the thermocouple stabilizes record the temperature and the period of the thermistor's square wave shown on the screen. This is the first data pair. Continue changing the bath's temperature and making readings until you have taken 22 pairs. The readings should be spread over the range of 7 and -33°C with a higher concentration near zero where the temperature is critical in the data processing. Press the ESC key to exit the program.

The program Calibrate will automatically be loaded. As described in the software section, Calibrate allows you to view and edit the Cal file. Enter the new data points into the Cal file. Make sure to keep the order straight and put the highest temperature pair at the beginning of the file (point 1). The algorithms in the controlling program assume this order. When finished save the file and return to the Main Menu. The calibration is now completed.

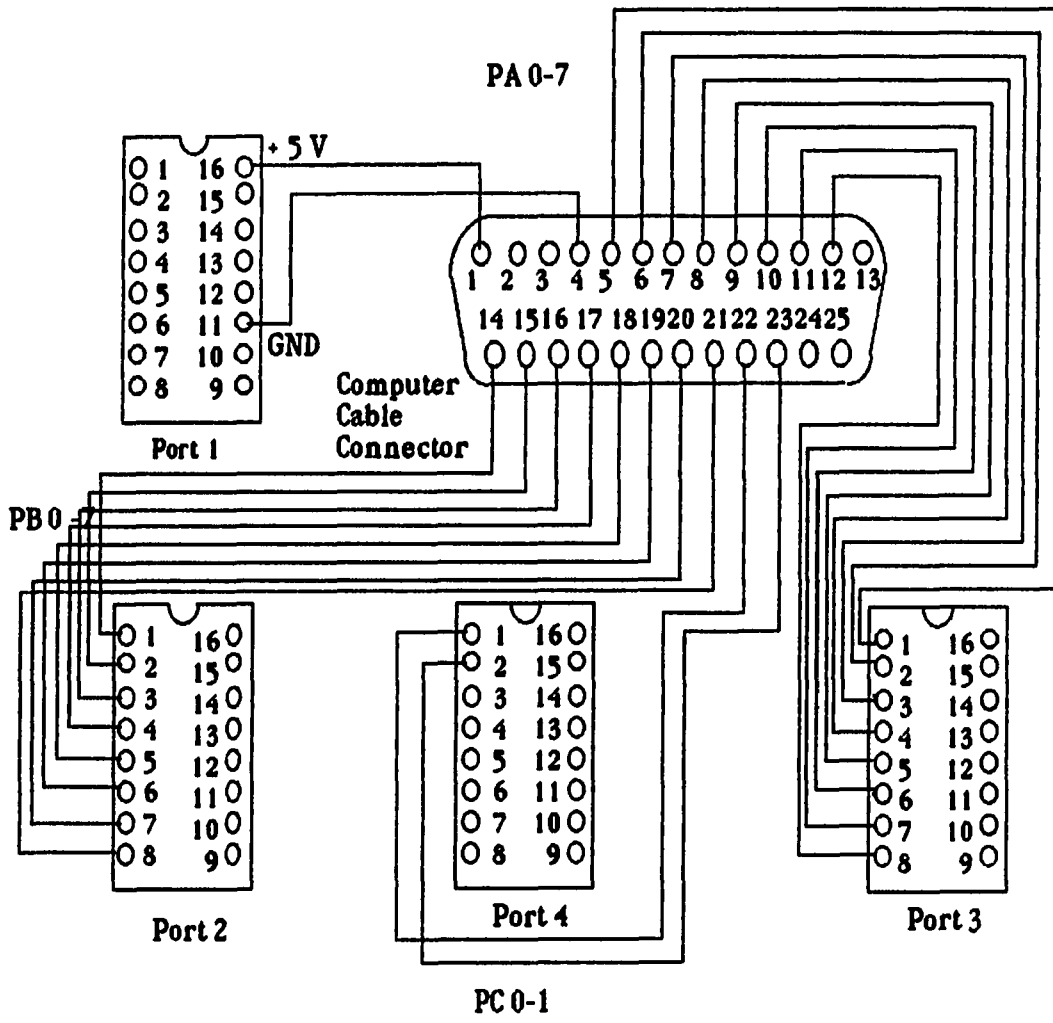
SIMPLIFICATIONS

The conductance porosimeter system uses two parallel interface cards because the original system was designed to handle both conductance PTP and phase transition porosimetry. In phase transition porosimetry the height of mercury in a capillary tube is monitored using capacitance. The variable capacitance was converted to a variable frequency square wave by a circuit similar to the one used for the thermistor. Because the amount of capacitance is so small, the noise from the temperature circuit necessitated two separate interface boxes and separate parallel interface cards. Therefore the conductance system can be set up so that only one parallel interface is used.

There are some changes to the programs and to the wiring to convert to only one interface card. The Solomat data must now be channelled through the card in slot 1 along with the thermistor input and the temperature control output. In the controlling program, Controller, change A1, B1 and C1 in lines 2040-2057 to these new values:

```
2040 A1 = -16256 + ISLOT * 256 + 1: B1 = -16384 + ISLOT * 256
2050 A3 = A1 + 2 : B3 = B1 + 2
2057 C1 = -16256 + ISLOT * 256 : C3 = C1 + 2
```

This changes the addresses the computer reads from the location in slot 4 to the card in slot 1. A1 is the location of port 1 and lines PA 0 - 7. It is used to send the digits 1 and 2 from the Solomat. Since the temperature control circuit uses port 1 of the card in slot 1 to send the control data to the temperature control circuit, A1 is also changed to port 3 and C1 (PC 0 or digit 5) is moved to port 4. The power line (+ 5V) remains with port 1. The +12 V line is not used and therefore eliminated. The computer connector for the interface box doesn't change but the new connector for the Solomat is as shown on the next page.



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**CONDUCTOMETRIC
PHASE TRANSITION POROSIMETER**

**USER'S MANUAL
Volume 2
Program Listings**

**CONDUCTOMETRIC PHASE TRANSITION POROSIMETER
USER'S MANUAL
Volume 2**

by
Todd Millard
and
Brett Gunnink

IOWA STATE UNIVERSITY, 1987

**CONDUCTOMETRIC PHASE TRANSITION POROSIMETER
USER'S MANUAL
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Program Listings**

LIST OF PROGRAMS

MAIN MENU
CONTROLLER
PLOTTERHP PLOTTER
CONVERT XY
DERIVATIVE XY
INTEGRATE XDYDX
CONVERT VP
CONVERT R
DERIVATIVE VP
DERIVATIVE LVP
SURFACE AREA
TEST FORMAT
TEMP MEASUREMENT
CALIBRATE

MAIN MENU

```

100 REM *****
110 REM * MAIN MENU FOR COND CALC *
120 REM * DOES ALL OF THE CALCULATIONS IN ORDER *
130 REM * BASED ON THE INPUT PARAMETERS SET HERE *
140 REM *****
150 D$ = CHR$(4): SPEED= 255: PRINT D$"PR#3"
160 SR$ = "Y":CR$ = "144.3":M1$ = "7":S1$ = "N"
170 S2$ = "N":S3$ = "N":K$ = "Y":M2$ = "3":C$(1) = "C":C$(0) = "W"
172 M3$ = "3"
175 TO$ = "273.15"
180 GOSUB 3000: REM PRINT MENU OUT
190 VTAB 21: INPUT "Enter your menu choice :";MN$
200 IF MN$ = "" THEN 190
210 IF LEN (MN$) > 1 THEN 190
220 IF ASC (MN$) > 56 OR ASC (MN$) < 49 THEN 250
230 ON ASC (MN$) - 48 GOSUB 400,700,850,1000,1100,1200,1300,1362
240 GOTO 190
250 IF ASC (MN$) < 65 OR ASC (MN$) > 68 THEN 190
260 ON ASC (MN$) - 64 GOTO 1400,1700,1800,270
270 END
400 REM *****
410 REM * ENTER THE FILENAME *
420 REM *****
430 VTAB 4: POKE 36,40
440 INPUT "";FIL$
450 IF FIL$ < > "" THEN 520
460 POKE 34,4
470 HOME
480 ONERR GOTO 600
490 PRINT D$"CATALOG /RAWDAT"
495 POKE 216,0
500 TEXT
510 GOTO 430
520 ONERR GOTO 650
530 PRINT D$"OPEN "FIL$;"C,D2"
535 PRINT D$"READ "FIL$;"C"
537 INPUT Z
540 PRINT D$"CLOSE"
545 POKE 216,0
550 GOSUB 3000
560 RETURN
600 REM NOT RAWDAT DISK
610 PRINT "Please insert the RAWDAT disk and press any key ...."
620 GET KB$
630 GOSUB 490
640 GOTO 190
650 REM FILE NOT FOUND
655 PRINT D$"CLOSE"

```

MAIN MENU

```

660 VTAB 23
670 PRINT "File not found on drive 2..."
675 PRINT D$;"DELETE "FIL$;"C"
680 GOSUB 430
690 GOTO 190
700 REM *****
710 REM * SET X.Y PARAMS *
720 REM *****
730 VTAB 5: POKE 36,40
740 INPUT "";SR$
750 IF SR$ < > "Y" AND SR$ < > "N" THEN 730
760 IF SR$ = "N" THEN CR$ = "": GOSUB 3000: GOTO 800
770 VTAB 6: POKE 36,40
780 INPUT "";CR$
790 IF VAL (CR$) < = 0 THEN 770
800 RETURN
850 REM *****
860 REM * SET DERIVATIVE PARAMS *
870 REM *****
880 VTAB 7: POKE 36,40
890 INPUT "";M1$
900 IF VAL (M1$) < 3 THEN 880
910 VTAB 8: POKE 36,40
920 INPUT "";S1$
930 IF S1$ < > "Y" AND S1$ < > "N" THEN 910
940 RETURN
1000 REM *****
1010 REM * SET X.Z PARAMS *
1020 REM *****
1030 VTAB 9: POKE 36,40
1040 INPUT "";S2$
1050 IF S2$ < > "Y" AND S2$ < > "N" THEN 1030
1060 RETURN
1100 REM *****
1110 REM * SET X.VP PARAMS *
1120 REM *****
1130 VTAB 10: POKE 36,40
1140 INPUT "";S3$
1150 IF S3$ < > "Y" AND S3$ < > "N" THEN 1130
1160 RETURN
1200 REM *****
1210 REM * SET R.VP PARAMS *
1220 REM *****
1230 VTAB 11: POKE 36,40
1240 INPUT "";K$
1250 IF K$ < > "Y" AND K$ < > "N" THEN 1230
1260 VTAB 12: POKE 36,40
1270 INPUT "";T0$

```

MAIN MENU

```

1280 RETURN
1300 REM *****
1310 REM * SET R.DVPDR PARAMS *
1320 REM *****
1330 VTAB 13: POKE 36,40
1340 INPUT "";M2$
1350 IF VAL (M2$) < 3 THEN 1330
1360 RETURN
1362 REM *****
1365 REM * SET R.DVPDLR PARAMS *
1370 REM *****
1375 VTAB 14: POKE 36,40
1380 INPUT "";M3$
1385 IF VAL (M3$) < 3 THEN 1375
1390 RETURN
1400 REM *****
1410 REM * RUN CALCULATIONS *
1420 REM *****
1422 IF FIL$ = "" THEN 190
1424 HOME
1428 PRINT "Performing the calculations..."
1430 REM CALL T.C.TO.X.Y
1440 CR = VAL (CR$)
1450 PRINT D$"CHAIN /CONPROG1/PROGRAMS/T.C.TO.X.Y"
1460 REM CALL DER.X.Y
1470 MA = VAL (M1$):SA = 1
1480 IF S1$ = "Y" THEN SA = 0
1490 PRINT D$"CHAIN /CONPROG1/PROGRAMS/DER.X.Y"
1500 REM CALL INT
1510 SA = 1
1520 IF S2$ = "Y" THEN SA = 0
1530 PRINT D$"CHAIN /CONPROG1/PROGRAMS/INT"
1540 REM CALL X.VP.FROM.X.Z
1550 SA = 1
1560 IF S3$ = "Y" THEN SA = 0
1570 PRINT D$"CHAIN /CONPROG1/PROGRAMS/X.VP.FROM.X.Z"
1580 REM CALL R.VP.FROM.X.VP
1590 PRINT D$"CHAIN /CONPROG1/PROGRAMS/R.VP.FROM.X.VP"
1600 REM CALL DER.R.VP
1610 MA = VAL (M2$)
1620 PRINT D$"CHAIN /CONPROG1/PROGRAMS/DER.R.VP"
1625 MA = VAL (M3$)
1630 PRINT D$"CHAIN /CONPROG1/PROGRAMS/DER.LR.VP"
1700 REM *****
1710 REM * CALL PLOT *
1720 REM *****
1730 PRINT D$"RUN /CONPROG1/PROGRAMS/P"
1800 REM *****

```

MAIN MENU

```

1810 REM * RUN CONDUCTANCE TEST *
1820 REM *****
1830 PRINT D$"RUN COND2,D1"
3000 REM *****
3010 REM * PRINT OUT THE MENU *
3020 REM *****
3030 HOME : PRINT
3040 HTAB 32: PRINT "CONDUCTANCE MENU"
3050 PRINT
3060 PRINT "1. Base Filename : ";FIL$
3070 PRINT "2. T.C to X.Y Series Resistor : ";SR$
3080 PRINT " Cond of Resistor : ";CR$
3090 PRINT "3. X.Y to X.DYDX Moving Average : ";M1$
3100 PRINT " Save File : ";S1$
3110 PRINT "4. X.DYDX to X.Z Save File : ";S2$
3120 PRINT "5. X.Z to X.VP Save File : ";S3$
3130 PRINT "6. X.VP to R.VP Thin File Out : ";K$
3135 PRINT " Temp of T0 : ";T0$
3140 PRINT "7. R.VP to R.DVPDR Moving Average : ";M2$
3150 PRINT "8. R.VP to R.DVPDLR Moving Average : ";M3$
3155 PRINT : PRINT
3160 PRINT " A. Calculate the data"
3170 PRINT " B. Plot data"
3180 PRINT " C. Run Conductance test"
3185 PRINT " D. Exit to system"
3190 RETURN

```

CONTROLLER

```

10 REM   CONDUCTANCE CONTROLLER FOR ONE SAMPLE
11 REM   MODIFIED TO ONE SAMPLE ON 2 OCT 86
12 HOME
15 D$ = CHR$(4)
18 HIMEM: 256 * PEEK (116) + PEEK (115) - 1024
20 PRINT D$"BLOAD CAP.RES.2,A$340,D1"
30 PRINT D$"BLOAD IO.CLOCK.SLOT1,A$9800"
45 POKE 1535,0
50 POKE 1407,0
55 POKE 1279,0
60 CALL - 26624
65 POKE 1919,1: REM   DISPLAYTIME
70 GOTO 2000
100 POKE 824,0: CALL 832
110 IF PEEK (822) > 3 THEN 140
120 TV = PEEK (823): IF TV < > 255 THEN POKE 823,TV + 1: GOTO 140
130 POKE 823,0
140 R = 65536 - ( PEEK (823) * 256 + PEEK (822))
150 R = R - 25000: IF R < 0 THEN R = R + 40536
160 RR = RR + R - AR(SR):AR(SR) = R: IF MR > SR THEN SR = SR + 1: GOTO 180
170 SR = 1
180 TEMP = INT (RR / MR + .5)
182 IF TEMP > P(NMPT) THEN TEMP = P(NMPT)
184 IF TEMP < P(1) THEN TEMP = P(1)
186 IF (TEMP < P(TI)) AND (TI > 1) THEN TI = TI - 1: GOTO 186
188 IF (TEMP > P(TI + 1)) AND (TI < NMPT - 1) THEN TI = TI + 1: GOTO 188
190 TEMP = INT (((TEMP - P(TI)) * (T(TI + 1) - T(TI)) / (P(TI + 1) -
P(TI)) + T(TI)) * 1E3 + .5) / 1E3
192 VTAB 6: HTAB 30
194 PRINT "           "": HTAB 30
196 PRINT TEMP;
200 OSEC = PEEK (1279) + 60 * PEEK (1407) + 3600 * PEEK (1535)
210 RETURN
300 REM SWITCH LEADS AND READS CONDUCTANCE
310 PO = (PO - INT (PO / 2) * 2) + INT (PO / 4) * 4
320 POKE C1,PO
340 CX = PEEK (C1):CY = PEEK (B1):CZ = PEEK (A1)
350 CX = (CX - INT (CX / 2) * 2) * 10000
360 CY = (CY - INT (CY / 16) * 6) * 100
370 CZ = CZ - INT (CZ / 16) * 6
380 CND = CX + CY + CZ
390 RETURN
400 REM CALCULATE TEMP CONTROL INFO
410 OT = INT ((M1 * S9 + B) + .5)
420 PT = INT ((M1 * F9 + B) + .5)
430 TS = 60 * ( ABS (F9 - S9) / (RA(TC(CCYCLE)) / 60)) / ABS (OT - PT)
440 SD = SGN (S9 - F9)
450 U6 = OSEC

```


CONTROLLER

```

460 RETURN
500 REM TURN 'FF HEATER
510 POKE D1,0
530 VTAB 2: POKE 36,10: PRINT "Off      "
540 RETURN
550 REM TURN ON COMPRESSOR, HEATER OFF
555 CCYCLE = CCYCLE + 1
560 S9 = TP(TC(CCYCLE)):F9 = BTM(TC(CCYCLE))
570 GOSUB 400: POKE D1,0T
585 VTAB 14: HTAB 40: PRINT "CYCLE NUMBER ";CCYCLE;" OF ";MCYCLE
590 VTAB 2: POKE 36,10: PRINT "Cool  "
595 RETURN
600 REM TURN ON HEATER, COMPRESSOR OFF
610 S9 = BTM(TC(CCYCLE)):F9 = TP(TC(CCYCLE))
620 GOSUB 400: POKE D1,0T
630 VTAB 2: POKE 36,10: PRINT "Heat  "
640 RETURN
700 REM INITIALIZE DATABASE AND FLAGS
710 REM SAMPLE DATA
730 FOR Y = 1 TO 300
740 SDT(Y,0) = 0
750 SDT(Y,1) = 0
760 NEXT Y
790 PNT = 1
800 FDNE = 0
840 INVL(TC(CCYCLE)) = - INVL(TC(CCYCLE))
850 COOL = NOT COOL
860 RETURN
1000 REM SAVE OFF DATA ON TEMPORARY DISK
1001 D$ = CHR$(4)
1002 IF COOL THEN W1$ = "C": GOTO 1010
1005 W1$ = "W"
1010 IF TD$(TC(CCYCLE)) = "N" THEN 1110
1020 PRINT D$;"OPEN  ";DIS$;".";CCYCLE + SCYCLE - 1;W1$;"D2"
1030 PRINT D$;"WRITE ";DIS$;".";CCYCLE + SCYCLE - 1;W1$
1040 PRINT PNT
1050 FOR N = 1 TO PNT
1060 PRINT SDT(N,0)
1070 PRINT SDT(N,1)
1080 NEXT N
1090 PRINT D$;"CLOSE ";DIS$;".";CCYCLE + SCYCLE - 1;W1$
1110 RETURN
1900 REM READ IN NUMBERS
1910 TMP$ = ""
1920 GET K$
1930 IF K$ = "." OR ASC (K$) = > 48 AND ASC (K$) < = 57 OR K$ = "-"
    THEN 960
1940 IF ASC (K$) = 13 THEN 1980

```

CONTROLLER

```

1950 IF ASC (K$) = 8 AND LEN (TMP$) = 1 THEN TMP$ = "": PRINT CHR$ (8);
    " "; CHR$ (8);
1952 IF ASC (K$) = 8 AND LEN (TMP$) > 1 THEN TMP$ = LEFT$ (TMP$, LEN
    (TMP$) - 1): PRINT CHR$ (8);" "; CHR$ (8);
1955 GOTO 1920
1960 TMP$ = TMP$ + K$
1965 PRINT K$;
1970 GOTO 1920
1980 TMP = VAL (TMP$)
1990 RETURN
2000 REM PRE MAIN PROGRAM
2005 GOSUB 4000
2020 PSLOT = 4:ISLOT = 1:COOL = 0:LCYCLE = 0:D$ = CHR$ (4):INVL = 0
2030 FSTP = 0:CCYCLE = 0
2035 M1 = 255 / ( - 35 - 7):B = 7 * - M1:D1 = - 16384 + ISLOT * 256 + 1:
OKE D1 + 2,255
2040 A1 = - 16384 + PSLOT * 256 + 1:B1 = - 16384 + PSLOT * 256
2050 A3 = A1 + 2:B3 = B1 + 2
2057 C1 = - 16256 + PSLOT * 256 + 1:C3 = C1 + 2
2058 POKE A3,0: POKE B3,0: POKE C3,254
2059 GOSUB 3500: REM GET NOTES
2060 HOME
2062 VTAB 16: HTAB 2: PRINT "# of Cycle Types : ";
2064 GOSUB 1900
2066 IF TMP = > 1 THEN 2072
2068 HTAB 21: PRINT " ";: HTAB 21
2070 GOTO 2064
2072 CT = INT (TMP)
2074 DIM TP(CT),BTTM(CT),INVL(CT),RA(CT),TD$(CT)
2076 FOR X = 1 TO CT
2080 HOME : VTAB 16: HTAB 2: PRINT "Cycle Type : ";X
2085 VTAB 18: HTAB 4
2087 PRINT "Maximum Temp : ";
2090 GOSUB 1900
2092 IF TMP < = 7 THEN 2100
2094 POKE 36,18: PRINT " ";: POKE 36,18
2096 GOTO 2090
2100 TP(X) = TMP
2110 HTAB 30
2120 PRINT "Minimum Temp : ";
2130 GOSUB 1900
2140 IF TMP < TP(X) THEN 2170
2150 POKE 36,44: PRINT " ";: POKE 36,44
2160 GOTO 2130
2170 BTTM(X) = TMP
2171 VTAB 18: POKE 36,58: PRINT "Temp Rate : ";
2172 GOSUB 1900
2173 IF TMP > 0 AND TMP < = 30 THEN 2178

```

CONTROLLER

```

2174 POKE 36,70: PRINT " ";: POKE 36,70
2175 GOTO 2172
2178 RA(X) = TMP
2250 VTAB 20: HTAB 10
2260 PRINT "Temp Interval Between Readings : ";
2270 GOSUB 1900
2280 IF TMP < = (TP(X) - BTM(X)) AND TMP > = 0 THEN 2302
2290 POKE 36,42: PRINT " ";: POKE 36,42
2300 GOTO 2270
2302 IF TMP = 0 THEN TD$(X) = "N":INVL(X) = .1: GOTO 2440
2304 TD$(X) = "Y"
2305 INVL(X) = TMP
2440 VTAB 24: HTAB 5
2450 PRINT "Is all information correct?";
2460 GET TMP$
2470 IF TMP$ = "Y" OR TMP$ = "y" THEN HTAB 5: CALL - 868: GOTO 2492
2480 IF TMP$ = "N" OR TMP$ = "n" THEN 2080
2490 GOTO 2460
2492 NEXT X
2493 GOSUB 6000: HOME
2496 CCYCLE = 0
2500 POKE 35,15: HOME : TEXT : VTAB 5: HTAB 30
2510 PRINT "Sample Description Starting Cycle"
2515 VTAB 6: HTAB 18: CALL - 868: PRINT "Sample 1";: INPUT DIS$
2516 VTAB 6: POKE 36,55: GOSUB 1900:SCYCLE = TMP
2520 VTAB 15: HTAB 5: PRINT "Is This name correct? ";: GET TMP$
2522 IF TMP$ = "Y" OR TMP$ = "y" THEN 2528
2525 IF TMP$ = "N" OR TMP$ = "n" THEN 2500
2527 GOTO 2520
2528 POKE 35,15: HOME : TEXT : GOSUB 7000
2530 VTAB 5: HTAB 30
2532 PRINT "Temp (C) Cond (uS) Point"
2534 VTAB 6: HTAB 5: PRINT LEFT$(DIS$,1)
2535 VTAB 6: HTAB 18: CALL - 868: PRINT "Sample 1"
2545 VTAB 3: POKE 36,50: PRINT "Press F to stop at cycle"
2550 DIM SDT(400,1)
2562 MR = 30
2565 DIM AR(MR)
2580 RTEMP = TP(TC(1))
2600 GOSUB 550: REM START COMPRESSOR
2610 GOSUB 6500
2620 GOSUB 700: REM INITIALIZE
2630 GOSUB 100: REM READ TEMP
2631 IF ABS (OSEC - U6) > = TS THEN U6 = OSEC:OT = OT + SD: IF OT > = 0
AND OT < = 255 THEN POKE D1,OT
2632 IF PEEK ( - 16384) = 195 THEN 2637
2633 IF PEEK ( - 16384) < > 198 THEN 2639
2634 FSTP = 1: VTAB 3: POKE 36,50

```

CONTROLLER

```

2635 CALL - 868: INVERSE : PRINT " Press 'C' to not halt ": NORMAL
2636 GOTO 2639
2637 FSTP = 0: VTAB 3: POKE 36,50
2638 PRINT "Press F to stop at cycle"
2639 POKE - 16368,0
2640 IF COOL AND TEMP < = RTEMP OR NOT COOL AND TEMP > = RTEMP THEN
GOTO 2655
2650 GOTO 2630: REM NEXT TEMP
2655 IF TD$(TC(CCYCLE)) = "N" THEN 2730
2660 GOSUB 300: REM TAKE CND READING
2670 SDT(PNT,0) = TEMP
2680 SDT(PNT,1) = CND
2705 HTAB 40: PRINT " ";
2710 HTAB 40: PRINT CND;
2715 POKE 36,51: PRINT " ";
2720 POKE 36,51: PRINT PNT
2730 IF RTEMP < = BTM(TC(CCYCLE)) AND COOL OR RTEMP = > TP(TC(CCYCLE))
AND NOT COOL THEN 3000: REM AT END OF CYCLE FOR SAMPLE
2740 RTEMP = INT ((RTEMP + INVL(TC(CCYCLE))) * 100) / 100
2750 PNT = PNT + 1
2760 GOTO 2630
3000 REM SAVE OFF DATA
3010 GOSUB 1000: REM SAVE TO TEMPORARY DISK
3020 IF NOT FSTP OR COOL AND FSTP THEN 3130
3030 GOSUB 500: REM SHUT OFF HEATER
3040 VTAB 15: HTAB 5
3050 PRINT "Program halted, do you wish to continue?";
3060 GET TMP$
3070 IF TMP$ = "Y" OR TMP$ = "y" THEN 3100
3080 IF TMP$ = "N" OR TMP$ = "n" THEN CALL - 868: GOTO 3140
3090 GOTO 3060
3100 HTAB 5: CALL - 868: REM ERASE LINE
3110 VTAB 3: POKE 36,50: PRINT "Press F to stop at cycle"
3120 FSTP = 0
3130 IF (CCYCLE - LCYCLE = NCYCLE OR CCYCLE = MCYCLE) AND NOT COOL THEN
3140
3135 GOTO 3180
3140 REM NO NEED TO TRANSFER DATA TO PERMANENT DISKS
3165 LCYCLE = CCYCLE
3170 IF (CCYCLE = MCYCLE) OR FSTP THEN 3210
3180 IF COOL THEN GOSUB 600: GOTO 3200: REM TURN ON HEATER
3190 GOSUB 550: REM TURN ON COMPRESSOR
3200 GOTO 2610: REM START NEXT CYCLE
3210 GOSUB 500: REM TURN OFF HEATER AND QUIT
3220 VTAB 24: HTAB 5
3230 PRINT "Program finished "
3240 PRINT : PRINT
3250 PRINT D$"RUN STARTUP,D1"

```

CONTROLLER

```

3500 REM NOTES ON SAMPLE
3510 HOME : VTAB 3: HTAB 29
3520 PRINT "Conductance Controller"
3530 VTAB 8: PRINT "Notes on Sample (press 'Return' at end of line) :"
3540 INPUT ">";N1$
3550 INPUT ">";N2$
3560 INPUT ">";N3$
3570 INPUT ">";N4$
3580 RETURN
4000 PRINT D$"OPEN CAL,D1"
4005 PRINT D$"READ CAL"
4010 INPUT NMPT
4020 DIM P(NMPT),T(NMPT)
4025 FOR TI = 1 TO NMPT
4030 INPUT P(TI),T(TI)
4040 NEXT TI
4042 TI = 1
4045 PRINT D$"CLOSE"
4050 RETURN
6000 REM NOW SPECIFY EACH CYCLE
6010 HOME : VTAB 16: HTAB 10
6020 PRINT "Number of Cycles : ";
6030 GOSUB 1900
6040 IF TMP = > 1 THEN 6070
6050 HTAB 29: PRINT " ";: HTAB 29
6060 GOTO 6030
6070 MCYCLE = INT (TMP)
6075 DIM TC(MCYCLE)
6080 FOR CCYCLE = 1 TO MCYCLE
6090 HOME : VTAB 16: HTAB 10
6100 PRINT "Cycle Number : ";CCYCLE;
6110 POKE 36,40
6120 PRINT "Cycle Type : ";
6130 GOSUB 1900
6140 IF TMP = > 1 AND TMP < = CT THEN 6160
6150 POKE 36,53: PRINT " ";: POKE 36,53
6155 GOTO 6130
6160 TC(CCYCLE) = INT (TMP)
6170 GOSUB 6500: REM PRINT OUT PARAMETERS
6180 VTAB 24: HTAB 5
6190 PRINT "Is this the correct cycle type ?";
6200 GET TMP$
6210 IF TMP$ = "Y" OR TMP$ = "y" THEN 6250
6220 IF TMP$ = "N" OR TMP$ = "n" THEN 6090
6230 GOTO 6200
6250 VTAB 24: HTAB 1: CALL - 868
6255 NEXT CCYCLE
6330 VTAB 16: HTAB 1: CALL - 868

```

CONTROLLER

```
6340 RETURN
6500 REM DISPLAY ITEMS
6510 VTAB 18: HTAB 4
6520 PRINT "Maximum Temp : ";TP(TC(CCYCLE));" ";
6530 HTAB 30
6540 PRINT "Minimum Temp : ";BTTM(TC(CCYCLE));" ";
6550 POKE 36,58
6560 PRINT "Temp Rate : ";RA(TC(CCYCLE));" ";
6570 VTAB 20: HTAB 10
6580 PRINT "Temp Interval between readings : ";
6590 IF TD$(TC(CCYCLE)) = "N" THEN PRINT "0": GOTO 6610
6600 PRINT INVL(TC(CCYCLE))
6610 RETURN
7000 REM SAVE OFF CYCLE TYPE DESCRIPTIONS ON DISK
7010 D$ = CHR$(4)
7030 PRINT D$;"OPEN ";DIS$;". ";SCYCLE;".CT,D2"
7040 PRINT D$;"WRITE ";DIS$;". ";SCYCLE;".CT"
7045 PRINT SCYCLE
7050 PRINT CT
7060 FOR N = 1 TO CT
7070 PRINT TP(N)
7072 PRINT BTTM(N)
7074 PRINT RA(N)
7076 PRINT INVL(N)
7078 PRINT TD$(N)
7080 NEXT N
7081 PRINT MCYCLE
7082 FOR N = 1 TO MCYCLE
7083 PRINT TC(N)
7084 NEXT N
7085 PRINT N1$: PRINT N2$: PRINT N3$: PRINT N4$
7090 PRINT D$;"CLOSE ";DIS$;". ";SCYCLE;".CT"
7110 RETURN
9999 POKE - 15986,64: HIMEM: 38400: END
```

PLOTTER

```

100 REM *****
110 REM * THIS PROGRAM PLOTS ONSCREEN PLOTS OR SAVES THE*
120 REM * VARIABLES IN A VARIABLE FILE AND PLOTS ON THE *
130 REM * HP7220 PLOTTER. *
140 REM *****
150 TEXT :CS$(1) = "TRUE":CS$(0) = "FALSE"
155 DIM AS%(30)
160 HOME :D$ = CHR$(4)
170 DR$ = "CALDAT"
180 L(1) = 7:L(2) = 2
200 CS = 1
235 REM *****
236 REM * CHECKING TO SEE IF DISK IN DRIVE TWO IS *
237 REM * CALDAT OR NOT. *
238 REM *****
240 K = 10: PRINT "MENU OF SUBDIRECTORIES"
250 PRINT : PRINT
260 HTAB (K): PRINT "1. X.Y"
270 HTAB (K): PRINT "2. X.DYDX"
275 HTAB (K): PRINT "3. X.Z"
280 HTAB (K): PRINT "4. X.VP"
290 HTAB (K): PRINT "5. R.VP"
300 HTAB (K): PRINT "6. R.DVPDR"
302 HTAB (K): PRINT "7. R.DVPDLR"
305 HTAB (K): PRINT "8. EXIT TO MAIN MENU"
310 INPUT " ENTER NUMBER OF YOUR CHOICE ";A$
320 IF (A$ < "1") AND (A$ > "8") THEN GOTO 240
330 A = VAL (A$)
332 IF A$ = "8" THEN 1040
333 IF A > = 2 AND A < = 4 THEN PRINT : INPUT "DO YOU WISH PORE RADIUS
MARKS ON TOP (Y/N DEFAULT = N) ";PZ$
334 K1$ = "Y": IF A$ = "6" THEN K1$ = "N"
335 IF A$ < > "6" AND A$ < > "7" AND A$ < > "5" THEN 338
336 PRINT : PRINT "DO YOU WISH TO PLOT THE RADIUS ON A LOG SCALE (Y/N
DEFAULT = ";K1$;") ";: INPUT TEMP$
337 KB$ = TEMP$: IF TEMP$ = "" THEN KB$ = K1$
338 IF A$ < "5" THEN L$(1) = "WARMING":L$(2) = "COOLING": GOTO 340
339 L$(2) = "NECK SIZES":L$(1) = "BODY SIZES"
340 ON A GOSUB 1180,1230,1330,1270,1320,1361,1361
390 HOME : PRINT "ENTER THE LETTER OF ANY VALUE THAT YOU WISH TO CHANGE."
391 L = 3:K = 2
400 PRINT "AND PRESS RETURN"
401 HTAB (K): PRINT "1. TITLE ..... "TI$
402 HTAB (K): PRINT "2. FILENAME ..... "F$(1)
403 HTAB (K): PRINT "3. LABEL ..... "L$(1)
404 HTAB (K): PRINT "4. LINETYPE ..... "L(1)
405 HTAB (K): PRINT "5. FILENAME ..... "F$(2)
406 HTAB (K): PRINT "6. LABEL ..... "L$(2)

```

PLOTTER

```

407 HTAB (K): PRINT "7. LINETYPE ..... "L(2)
410 HTAB (K): PRINT "8. MINIMUM X ..... "X1
420 HTAB (K): PRINT "9. MAXIMUM X ..... "X2
430 HTAB (K): PRINT "10. X-INTERVAL SIZE ..... "XINC
440 HTAB (K): PRINT "11. MINIMUM Y ..... "Y1
450 HTAB (K): PRINT "12. MAXIMUM Y ..... "Y2
460 HTAB (K): PRINT "13. Y-INTERVAL SIZE ..... "YINC
480 HTAB (K): PRINT "14. PAPER LENGTH ..... "PL
495 PRINT "-----"
500 PRINT "A-ONSCREEN PLOTTING";: HTAB 30: PRINT "D-CLEAR GRAPHICS
SCREEN";CS$(CS)
502 PRINT "B-HP722 PLOTTING ";: HTAB 30: PRINT "E-EXIT PROGRAM"
504 PRINT "C-VIEW GRAPHICS SCREEN";: HTAB 30: PRINT "F-FIND NEW DATA
POINTS"
510 PRINT
520 VTAB (23): INPUT "ENTER YOUR CHOICE AND PRESS RETURN.";SE$
525 IF SE$ = "" THEN 520
530 IF ( VAL (SE$) < 1 OR VAL (SE$) > 13) AND (SE$ < "A" OR SE$ > "F")
THEN 520
540 M = VAL (SE$)
550 ON M GOTO 3000,3050,3150,3200,3250,3350,3400,3450,3500,3550,3600,3650,
3700,3750
560 IF ASC ( LEFT$ (SE$,1)) = 65 THEN 720: REM A
570 IF ASC ( LEFT$ (SE$,1)) = 66 THEN PRINT D$"CHAIN
/CONPROG1/PROGRAMS/PLOT": REM B
580 IF ASC ( LEFT$ (SE$,1)) = 67 THEN GOTO 1200: REM C
590 IF ASC ( LEFT$ (SE$,1)) = 68 THEN CS = NOT CS: GOTO 390: REM D
600 IF ASC ( LEFT$ (SE$,1)) = 69 THEN PRINT D$"RUN STARTUP,D1": REM E
610 IF ASC ( LEFT$ (SE$,1)) = 70 THEN 160: REM F
720 PRINT CHR$ (17)
730 IF NOT CS THEN POKE - 16304,0: POKE - 16302,0: POKE - 16299,0:
POKE - 16297,0: GOTO 755
740 HGR2
750 POKE - 16368,0
755 YA = 159:YB = 0:XA = 1:XB = 279
760 HCOLOR= 3: IF NOT CS THEN GOTO 862
770 HPLOT 0,0 TO 279,0 TO 279,159 TO 1,159 TO 1,0
790 VARSCAL = (YA - YB) * YINC / (Y2 - Y1)
800 FOR K = 1 + VARSCAL TO YA STEP VARSCAL
810 HPLOT 1,K TO 5,K: HPLOT 279,K TO 274,K: NEXT K
815 IF A = 5 THEN 2100: REM PLOT X ON LOG SCALE
820 HARSCAL = (XB - XA) * XINC / (X2 - X1)
825 IF HARSCAL < 1 THEN HARSCAL = 1
830 FOR K = 1 TO XB - HARSCAL STEP HARSCAL
840 HPLOT K,159 TO K,154
850 K = INT (K)
860 NEXT K
862 FOR I = 1 TO 2

```


PLOTTER

```

865 IF F$(I) = "" THEN 1021
870 PRINT CHR$(4)"OPEN /"DR$/"FIL$/"F$(I)
880 ONERR GOTO 1090
890 PRINT CHR$(4)"READ/"DR$/"FIL$/"F$(I)
900 INPUT NMPTS
920 POKE 216,0
930 FOR T = 1 TO NMPTS
935 IF PEEK (- 16384) = 27 THEN T = NMPTS: GOTO 1010
936 POKE - 16368,0
940 INPUT X,Y
950 IF (X < = X2) AND (X > = X1) THEN J = FN CN(X)
960 IF (Y < = Y2) AND (Y > = Y1) THEN K = FN MOH(Y)
970 IF (X > X2) OR (X < X1) THEN J = 0
980 IF (Y > Y2) OR (Y < Y1) THEN K = 1
990 IF (J = 0) OR (K = 1) THEN J = 0:K = 1
1000 H PLOT J,K
1010 NEXT T
1020 PRINT CHR$(4)"CLOSE/"DR$/"FIL$/"F$(I)
1021 NEXT I
1022 PRINT CHR$(7); CHR$(7); CHR$(7)
1030 GET A$: PRINT CHR$(18): TEXT : PRINT D$"PR#3":K = 10: HOME :
GOTO 390
1040 PRINT CHR$(4)"RUN STARTUP,D1"
1050 REM *****
1060 REM *THIS IS AN ERROR HANDLING ROUTINE WRITTEN TO CLEAR*
1070 REM *ERRORS IN INPUTING FILE NAMES. *
1080 REM *****
1090 TEXT : PRINT CHR$(4)"CLOSE/"DR$/"FIL$/"SUB$
1100 PRINT CHR$(4)"DELETE /"DR$/"FIL$/"SUB$: POKE 216,0
1110 GET A$: PRINT D$"PR#3": GOTO 150
1130 END
1140 REM *****
1150 REM *THIS IS A LOOKUP TABLE FOR VARIABLE DEFAULTS AND *
1160 REM *EQUATIONS. *
1170 REM *****
1180 LAB$ = "Degrees Celcius":PL = 8.5:X1 = - 41:X2 = - 36:Y1 = - 1:
Y2 = 4:XNUM = .15:YNUM = .185:EL$ = "Log (Conductance in micro Mho)":
BT$ = "(-1/T), K x 10"
1190 SL = 2:XIN = 1:YIN = .5:LIN = 7:PEN = 1:FIL$ = "X.Y"
1200 DEF FN MOH(T) = 159 - (T - Y1) * (YA - YB) / (Y2 - Y1)
1210 DEF FN CN(R) = ((R - X1) * XB / (X2 - X1)) + 1
1220 RETURN
1230 LAB$ = "":PL = 8.5:YINC = .5:XIN = 1:PEN = 1:X1 = - 41:X2 = - 36:
Y1 = 0:LIN = 7:Y2 = 3:XNUM = .15:YNUM = .185:BT$ = "(-1/T), K x 10":
EL$ = "d(log(Conductance))/d(-1/T)"
1231 FIL$ = "X.DYDX"
1240 DEF FN MOH(T) = 159 - (T - Y1) * (YA - YB) / (Y2 - Y1)
1250 DEF FN CN(R) = ((R - X1) * XB / (X2 - X1)) + 1

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PLOTTER

```

1260 RETURN
1270 LAB$ = "":PL = 8.5:LIN = 7:PEN = 1:X1 = - 41:X2 = - 36:Y1 = 0:
      Y2 = 1:YIN = 1:XNUM = .15:YNUM = .185:XINC = 1
1280 EL$ = "Relative Conducting Pore Volume":BT$ = "(-1/T), K x 10":
      FIL$ = "X.VP"
1290 DEF FN MOH(T) = 159 - (T) * YA / (Y2 - Y1)
1300 DEF FN CN(R) = ((R - X1) * XB / (X2 - X1)) + 1
1310 RETURN
1320 LAB$ = "":PL = 8.5:X1 = 10:X2 = 10000:Y1 = 0:Y2 = 1.1:YIN = .1:
      XNUM = .15:YNUM = .185:XINC = 10:BT$ = "Radius, Angstroms":
      EL$ = "Relative Conducting Pore Volume":FIL$ = "R.VP"
1321 DEF FN MOH(T) = 159 - (T - Y1) * (YA - YB) / (Y2 - Y1)
1322 DEF FN CN(R) = ( LOG (R) / LOG (10) - LOG (X1) / LOG (10)) /
      ( LOG (X2) / LOG (10) - LOG (X1) / LOG (10)) * XB + 1
1324 IF KB$ = "N" THEN DEF FN CN(R) = ((R - X1) * XB / (X2 - X1)) + 1:
      A = 6:X1 = 0:XINC = 1000
1325 PEN = 1:LIN = 7
1326 RETURN
1330 LAB$ = "":PL = 8.5:YIN = 1:XIN = 1:X1 = - 41:X2 = - 36:Y1 = 0:
      Y2 = 5:PEN = 1:LIN = 2:XNUM = .15:YNUM = .185:BT$ = "(-1/T) K x 10":
      EL$ = "      Z      "
1333 PEN = 1:LIN = 7:FIL$ = "X.Z"
1340 DEF FN MOH(T) = 159 - (T - Y1) * (YA - YB) / (Y2 - Y1)
1350 DEF FN CN(R) = ((R - X1) * XB / (X2 - X1)) + 1
1360 RETURN
1361 LAB$ = "":PL = 8.5:X1 = 10:X2 = 10000:Y1 = 0:Y2 = 2:YIN = .2:
      XNUM = .15:YNUM = .185:XINC = 10:BT$ = "Radius, Angstroms "
1362 EL$ = "d(Pore Volume)/d(log Radius)":PEN = 1:LIN = 7:
      FIL$ = "R.DVPDLR"
1363 IF A = 6 THEN EL$ = "d(Pore Volume)/d(Radius)":FIL$ = "R.DVPDR":
      Y2 = .1:YIN = .01
1364 IF KB$ = "N" THEN X1 = 0:XINC = 1000
1365 DEF FN MOH(T) = 159 - (T - Y1) * (YA - YB) / (Y2 - Y1)
1366 IF KB$ = "Y" THEN DEF FN CN(R) = ( LOG (R) / LOG (10) - LOG (X1)
      / LOG (10)) / (LOG (X2) / LOG (10) - LOG (X1) / LOG (10)) * XB + 1:
      A = 5: GOTO 1368
1367 DEF FN CN(R) = ((R - X1) * XB / (X2 - X1)) + 1
1368 RETURN
1370 HOME
1380 PRINT D$"CATALOG /"DR$"/"FIL$
1390 INPUT "ENTER NEW FILE NAME      ";SUB$
1400 IF SUB$ = "" THEN GOTO 1370
1410 RETURN
1420 ONERR GOTO 1510
1430 PRINT D$"OPEN /CALDAT, TDIR"
1440 PRINT D$"CLOSE /CALDAT"
1450 POKE 216,0: GOTO 240
1470 REM *****

```

PLOTTER

```

1480 REM * CHECKING THE DISK IN DRIVE 2 FOR*
1490 REM * THE PREFIX OF /CALDAT. *
1500 REM *****
1510 PRINT "THE DISK IN DRIVE 2 IS NOT CALDAT. PLEASE REPLACE THE DISK IN
DRIVE 2 WITH CALDAT."
1520 INPUT "AND PRESS RETURN TO CONTINUE.";X$
1530 POKE 216,0: GOTO 100
2000 PRINT CHR$(17): POKE - 16304,0: POKE - 16302,0: POKE - 16299,0:
POKE - 16297,0
2001 GET A$
2010 POKE - 16303,0: POKE - 16300,0
2020 PRINT D$"PR#3": PRINT : GOTO 390
2100 REM SETS UP X AXIS ON A LOG SCALE
2105 IF X1 = 0 THEN X1 = 1
2110 FOR K = LOG (X1) / LOG (10) TO LOG (X2) / LOG (10) - 1
2120 FOR K1 = 1 TO 10
2130 XP = FN CN(K1 * 10 ^ K)
2135 IF XP = 280 THEN 2150
2140 H PLOT XP,159 TO XP,154
2150 NEXT K1
2160 NEXT K
2180 GOTO 862
2300 REM ROUNDS DOWN TO THE NEXT CYCLE
2310 VTAB (L): HTAB (MM)
2320 INPUT "";X1
2330 X1 = 10 ^ ( INT ( LOG (X1) / LOG (10)))
2340 VTAB (L): HTAB (MM)
2350 PRINT X1;" "
2360 GOTO 520
2400 REM ROUNDS TO NEAREST CYCLE
2410 VTAB (L + 1): HTAB (MM)
2420 INPUT "";X2
2430 X2 = 10 ^ ( INT ( LOG (X2) / LOG (10) + .5))
2440 VTAB (L + 1): HTAB (MM)
2450 PRINT X2;" "
2460 GOTO 520
3000 REM TITLE
3010 VTAB 3: HTAB 40
3020 INPUT "";TI$
3030 GOTO 520
3050 REM FILENAME
3060 VTAB 4: HTAB 40
3070 INPUT "";F$(1)
3080 IF F$(1) < > "" THEN 390
3090 POKE 34,9
3100 HOME
3110 PRINT D$"CATALOG /"DR$"/"FIL$
3120 TEXT

```

PLOTTER

```
3130 GOTO 3060
3150 REM LABEL
3160 VTAB 5: HTAB 40
3170 INPUT "";L$(1)
3180 GOTO 520
3200 REM LINETYPE
3210 VTAB 6: HTAB 40
3220 INPUT "";L(1)
3230 GOTO 520
3250 REM FILENAME 2
3260 VTAB 7: HTAB 40
3270 INPUT "";F$(2)
3280 IF F$(2) < > "" THEN 390
3290 POKE 34,9
3300 HOME
3310 PRINT D$"CATALOG /"DR$"/"FIL$
3320 TEXT
3330 GOTO 3260
3350 REM LABEL
3360 VTAB 8: HTAB 40
3370 INPUT "";L$(2)
3380 GOTO 520
3400 REM LINETYPE
3410 VTAB 9: HTAB 40
3420 INPUT "";L(2)
3430 GOTO 520
3450 REM MIN X
3460 VTAB 10: HTAB 40
3470 INPUT "";X1
3475 IF X1 = 0 AND KB$ = "Y" THEN X1 = 1
3477 VTAB 10: HTAB 40: PRINT X1
3480 GOTO 520
3500 REM MAX X
3510 VTAB 11: HTAB 40
3520 INPUT "";X2
3530 GOTO 520
3550 REM X INT
3560 VTAB 12: HTAB 40
3570 INPUT "";XIN
3580 GOTO 520
3600 REM MIN Y
3610 VTAB 13: HTAB 40
3620 INPUT "";Y1
3630 GOTO 520
3650 REM MAX Y
3660 VTAB 14: HTAB 40
3670 INPUT "";Y2
3680 GOTO 520
```

PLOTTER

```
3700 REM Y INT
3710 VTAB 15: HTAB 40
3720 INPUT "";YIN
3730 GOTO 520
3750 REM PAPER LENGTH
3760 VTAB 16: HTAB 40
3770 INPUT "";PL
3780 GOTO 520
```

HP PLOTTER

```

100 REM *****
110 REM * HARD PLOTS TO THE HP FROM PLOT PROGRAM *
120 REM *****
130 D$ = CHR$(4):LT = LOG(10):C$ = CHR$(3)
135 XNUM = .15:YNUM = .185
140 SP$ = " .00000":SN$ = " -.00000":BL$ = " "
150 GOSUB 1000: REM INITIALIZE HP
160 GOSUB 1200: REM DEFINE FUNCTIONS
170 PRINT "PA"XA", "YA";"
180 PRINT "PD;PA"XA", "YA", "XB", "YA", "XB", "YB", "XA", "YB", "XA", "YA",
  "XB", "YA", "XB", "YB", "XA", "YB", "XA", "YA";PU;"
190 GOSUB 2000: REM PLOTS SCALES
200 GOSUB 4500: REM PLOTS LABELS
210 GOSUB 4000: REM PLOTS POINTS
220 GOTO 5000: REM GRAPH TITLES
1000 REM *****
1010 REM * INIT FOR THE HP *
1020 REM *****
1030 PRINT D$"PR#2"
1040 PRINT CHR$(27)".("
1050 PRINT " IN;"
1060 PRINT CHR$(27)".I40;0;17:"
1070 PRINT "SP1;SI"XNUM", "YNUM";"
1080 PRINT "DI1,0;VS20;"
1090 IF PL = 14 THEN XA = 900:XB = 9470:YA = 900:YB = 6400: GOTO 1110
1100 XA = 900:XB = 9470:YA = 900:YB = 6400
1110 RETURN
1200 REM *****
1210 REM * DEFINE FUNCTIONS *
1220 REM *****
1230 IF A < > 5 THEN 1290
1240 REM LOG FUNCTION
1250 HARSCAL = ( LOG ( ABS ( X2 / X1 ) ) / LT ) / ( XB - XA )
1260 XT = LOG ( ABS ( X1 ) ) / LT
1270 DEF FN XL(R) = ( LOG ( ABS ( R ) ) / LT - XT ) / HARSCAL + XA
1280 GOTO 1380
1290 REM LINEAR FUNCTION
1300 HARSCAL = ( X2 - X1 ) / ( XB - XA )
1310 DEF FN XL(R) = ( R - X1 ) / HARSCAL + XA
1380 REM LINEAR FUNCTION
1390 VERSCAL = ( Y2 - Y1 ) / ( YB - YA )
1400 DEF FN YL(T) = ( T - Y1 ) / VERSCAL + YA
1410 RETURN
2000 REM *****
2010 REM * PLOTS SCALES *
2020 REM *****
2030 IF A < > 5 THEN GOSUB 2100: GOTO 2060
2040 GOSUB 2300: REM LOG X

```

HP PLOTTER

```

2060 GOSUB 2500: REM   LINEAR Y
2070 RETURN
2100 REM *****
2110 REM *   LINEAR X PLOT   *
2120 REM *****
2130 FOR XS = X1 TO X2 STEP XIN
2140 XN = FN XL(XS)
2150 GOSUB 3000: REM X LABEL
2160 PRINT "PA"XN - 550", "YA - 200";LB"XS$;C$
2170 PRINT "PA"XN", "YA";PD; PA"XN", "YA + 50";PU;"
2180 NEXT XS
2190 FOR XS = X1 TO X2 STEP XIN
2200 XN = FN XL(XS)
2210 PRINT "PA"XN", "YB";PD;PA"XN", "YB - 25";PU;"
2220 NEXT XS
2230 RETURN
2300 REM *****
2310 REM *   LOG X PLOT   *
2320 REM *****
2330 FOR K = LOG ( ABS ( X1 ) ) / LT TO LOG ( ABS ( X2 ) ) / LT - 1
2340 FOR K1 = 1 TO 10
2345 XS = K1 * 10 ^ K
2350 XN = ( LOG ( XS ) / LT - XT ) / HARSCAL + XA
2360 PRINT "PA"XN", "YA";PD;PA"XN", "YA + 50";PU;"
2370 IF ( K1 < > 1 ) AND ( K1 < > 5 ) AND ( ( K1 < > 10 ) OR
      ( K < > ( LOG ( ABS ( X2 ) ) / LT - 1 ) ) ) THEN 2400
2375 XIN = 10 ^ K
2380 GOSUB 3000: REM X LABEL
2390 PRINT "PA"XN - 550", "YA - 200";LB"XS$;C$
2400 NEXT K1
2410 NEXT K
2420 FOR K = LOG ( ABS ( X1 ) ) / LT TO LOG ( ABS ( X2 ) ) / LT - 1
2430 FOR K1 = 1 TO 10
2435 XS = K1 * 10 ^ K
2440 XN = FN XL(XS)
2450 PRINT "PA"XN", "YB";PD;PA"XN", "YB - 25";PU;"
2460 NEXT K1
2470 NEXT K
2480 RETURN
2500 REM *****
2510 REM *   LINEAR Y PLOT   *
2520 REM *****
2530 FOR YS = Y1 TO Y2 STEP YIN
2540 YN = FN YL(YS)
2550 GOSUB 3200: REM Y LABEL
2560 PRINT "PA"XA - 640", "YN";LB"YS$;C$
2570 PRINT "PA"XA", "YN";PD;PA"XA + 50", "YN";PU;"
2580 NEXT YS

```

HP PLOTTER

```

2590 FOR YS = Y1 TO Y2 STEP YIN
2600 YN = FN YL(YS)
2610 PRINT "PA"XB", "YN"; PD; PA"XB - 25", "YN"; PU;"
2620 NEXT YS
2630 RETURN
3000 REM *****
3010 REM * X LABELS *
3020 REM *****
3030 RF = 0
3040 IF XIN < > 0 THEN RF = - INT ( LOG ( ABS ( XIN ) ) / LT )
3050 XS = INT ( XS * 10 ^ RF + .5 ) / 10 ^ RF
3060 IF XS = 0 OR ABS ( XS ) > = .01 THEN XS$ = LEFT$
      ( BL$, 7 - LEN ( STR$ ( XS ) ) ) + STR$ ( XS ) : GOTO 3100
3065 XS$ = STR$ ( INT ( ABS ( XS ) * 10 ^ RF + .5 ) )
3070 IF XS < 0 THEN XS$ = MID$ ( SN$, 1 + RF, 7 - LEN ( XS$ ) ) + XS$ : GOTO
3100
3080 XS$ = MID$ ( SP$, 1 + RF, 7 - LEN ( XS$ ) ) + XS$
3090 XS$ = RIGHT$ ( XS$, LEN ( XS$ ) - 1 )
3100 RETURN
3200 REM *****
3210 REM * Y LABELS *
3220 REM *****
3230 RF = 0
3240 IF YIN < > 0 THEN RF = - INT ( LOG ( YIN ) / LT )
3250 YS = INT ( YS * 10 ^ RF + .5 ) / 10 ^ RF
3260 IF YS = 0 OR ABS ( YS ) > = .01 THEN YS$ = LEFT$ ( BL$, 7 - LEN
      ( STR$ ( YS ) ) ) + STR$ ( YS ) : GOTO 3300
3265 YS$ = STR$ ( INT ( ABS ( YS ) * 10 ^ RF + .5 ) )
3270 IF YS < 0 THEN YS$ = MID$ ( SN$, 1 + RF, 7 - LEN ( YS$ ) ) + YS$ :
      GOTO 3300
3280 YS$ = MID$ ( SP$, 1 + RF, 7 - LEN ( YS$ ) ) + YS$
3290 YS$ = RIGHT$ ( YS$, LEN ( YS$ ) - 1 )
3300 RETURN
4000 REM *****
4010 REM * PLOT FUNCTION *
4020 REM *****
4040 FOR K = 1 TO 2
4041 IF F$(K) = "" THEN 4225
4042 PRINT "SP1;LT"L(K)",2;"
4045 PRINT D$"OPEN /"DR$/"FIL$/"F$(K)
4047 PRINT D$"READ /"DR$/"FIL$/"F$(K)
4048 INPUT NMPTS
4050 I = 1
4055 INPUT X,Y
4060 XP = FN XL(X)
4070 YP = FN YL(Y)
4080 IF XP > XB THEN XP = XB
4090 IF (XP < XA) THEN XP = XA

```


HP PLOTTER

```

4095 IF YP > YB THEN YP = YB
4100 IF (YP < YA) THEN YP = YA
4120 PRINT "PA"XP", "YP";PD;PA"
4130 FOR I = 2 TO NMPTS
4135 INPUT X,Y
4140 XP = FN XL(X)
4150 YP = FN YL(Y)
4160 IF XP > XB THEN XP = XB
4170 IF (XP < XA) THEN XP = XA
4180 IF YP > YB THEN 4210
4190 IF (YP < YA) THEN 4210
4200 PRINT XP", "YP", "
4210 NEXT I
4215 PRINT D$"CLOSE"
4220 PRINT XP", "YP";PU;"
4225 NEXT K
4230 RETURN
4500 REM *****
4510 REM * PLOT AXIS LABELS *
4520 REM *****
4530 PRINT "SP1;SI.24,.38;"
4540 XP = (XB + XA) / 2 - LEN (BT$) / 2 * 150
4560 YP = YA - 600
4570 PRINT "PA"XP", "YP";LB"BT$;C$"
4575 IF A > 4 THEN 4580
4577 PRINT "SI.1,.12;PA"(XB + XA) / 2 - 50 + LEN (BT$) / 2 * 150",
"YA - 500";LB - 1"C$"
4578 PRINT "SI.24,.38;"
4580 YP = (YB + YA) / 2 - LEN (EL$) / 2 * 150
4590 PRINT "DIO,1;PA"XA - 700", "YP";LB"EL$;C$"
4595 PRINT "DI1,0;"
4600 RETURN
5000 REM *****
5010 REM * TITLES *
5020 REM *****
5025 PRINT "SI.30,.50;"
5030 XP = (XB + XA) / 2 - LEN (TI$) / 2 * 160
5040 YP = YB + 700
5050 PRINT "PA"XP", "YP";LB"TI$;C$"
5055 FOR K = 1 TO 2
5057 IF F$(K) = "" THEN 5120
5060 LIN = L(K)
5070 IF LIN = 0 THEN LIN = 2
5080 PRINT "SI.22,.30;LT"LIN",2;"
5085 IF LIN = 7 THEN PRINT "LT;"
5090 PRINT "PA"XA + 600", "YB + 170";PD;PA"XA + 1275", "YB + 170";PU;"
5100 PRINT "PA"XA + 1475", "YB + 170";LB"L$(K);C$"
5110 XA = (XA + XB) / 2

```

HP PLOTTER

5120 NEXT K

5130 PRINT "SPO;"

5200 PRINT D\$"PR#3": PRINT D\$"CHAIN /CONPROG1/PROGRAMS/P,@390,D1"

CONVERT XY

```

100 REM *****
110 REM * THIS PROGRAM CONVERTS RAW DATA OF TEMP AND *
120 REM * CONDUCTANCE INTO X,Y DATA PAIRS *
130 REM *****
140 SU = 1
142 IF SR$ = "Y" THEN SU = 10
145 DEF FN CN(X) = LOG (X / SU) / LOG (10)
170 PRINT "Converting Raw Data to X.Y"
180 DEF FN CN(X) = LOG ((X * CR / SU) / (CR - X / SU)) / LOG (10)
190 IF SR$ = "N" THEN 240
200 PRINT D$"OPEN "FIL$;C$(0)",D2"
210 PRINT D$"OPEN "FIL$;C$(1)
220 PRINT D$"READ "FIL$;C$(0)
230 INPUT NMPTS(0)
240 PRINT D$"READ "FIL$;C$(1)
250 INPUT NMPTS(1)
260 N = NMPTS(0)
270 IF NMPTS(1) > NMPTS(0) THEN N = NMPTS(1)
280 DIM A(1,N),B(1,N)
285 PRINT D$"READ "FIL$;C$(0)
287 FOR K = NMPTS(0) TO 1 STEP - 1
290 INPUT A(0,K),B(0,K)
295 NEXT K
300 PRINT D$"READ "FIL$;C$(1)
305 FOR K = 1 TO NMPTS(1)
307 INPUT A(1,K),B(1,K)
310 NEXT K
320 PRINT D$"CLOSE"
325 INPUT "Please insert the CALDAT disk in drive 2 and press
RETURN...";KB$
327 IF KB$ < > "" THEN 325
328 ONERR GOTO 1000
329 FOR L = 0 TO 1
330 PRINT D$"OPEN/CALDAT/X.Y/"FIL$;C$(L)
340 PRINT D$"WRITE/CALDAT/X.Y/"FIL$;C$(L)
344 PRINT NMPTS(L)
350 FOR K = 1 TO NMPTS(L)
360 A(L,K) = - 1 / (A(L,K) + 273.6) * 1E4
370 B(L,K) = FN CN(B(L,K))
380 PRINT A(L,K): PRINT B(L,K)
390 NEXT K
400 PRINT D$"CLOSE /CALDAT/X.Y/"FIL$;C$(L)
410 NEXT L
495 POKE 216,0
500 PRINT D$"CHAIN STARTUP,@1460,D1"
1000 PRINT "Check disk in drive 2"
1010 GOTO 325

```

DERIVATIVE XY

```

100 REM *****
110 REM * THIS PROGRAM FINDS THE DERIVATIVE *
120 REM * OF X W/RESPECT TO Y, DY/DX *
140 REM *****
160 PRINT "Converting X.Y to X.DYDX"
360 DIM TE(MA),VO(MA),TZ(MA),VZ(MA)
440 FOR K = 0 TO 1
445 FL = 0
446 FOR S = 1 TO MA:TE(S) = 0:VO(S) = 0: NEXT S
450 FOR S = 1 TO MA - 1:TE(S) = A(K,S):VO(S) = B(K,S)
460 NEXT S
465 PO = 0:PI = MA - 1
470 FOR I = INT (MA / 2 + 1) TO MA - 1:T = 0:V = 0
480 FOR J = 1 TO I:T = T + TE(J):V = V + VO(J): NEXT J
490 TS = T / I
500 VS = V / I
510 IF NOT FL THEN FL = 1: GOTO 570
520 DVDT = INT ((VS - OV) / (TS - OT) * 1E5 + .5) / 1E5
530 WT = INT ((TS + OT) / 2 * 1E3 + .5) / 1E3
540 FL = FL + 1
550 PO = PO + 1
560 A(K,PO) = WT:B(K,PO) = DVDT: IF FL = 2 THEN PO = PO + 1:
    A(K,PO) = WT:B(K,PO) = DVDT
570 OT = TS:OV = VS
580 NEXT I
590 FOR I = 1 TO NMPTS(K) - MA + 1
600 PI = PI + 1:TE = A(K,PI):VO = B(K,PI)
610 IF I > NMPTS(K) - 2 * MA + 2 THEN Z = I - NMPTS(K) + 2 * MA - 1:
    TZ(Z) = TE:VZ(Z) = VO
620 T = T + TE - TE(S):TE(S) = TE
630 V = V + VO - VO(S):VO(S) = VO
640 S = S + 1: IF MA < S THEN S = 1
650 TS = T / MA
660 VS = V / MA
680 IF TS = OT THEN DVDT = 00: GOTO 700
690 DVDT = INT ((VS - OV) / (TS - OT) * 1E5 + .5) / 1E5:00 = DVDT
700 WT = INT ((TS + OT) / 2 * 1E3 + .5) / 1E3
710 PO = PO + 1:A(K,PO) = WT:B(K,PO) = DVDT
720 OT = TS:OV = VS
730 NEXT I
740 FOR I = 2 TO INT (MA / 2 + 1):T = 0:V = 0
750 FOR J = I TO MA:T = T + TZ(J):V = V + VZ(J): NEXT J
760 TS = T / (J - I)
770 VS = V / (J - I)
790 DVDT = INT ((VS - OV) / (TS - OT) * 1E5 + .5) / 1E5
800 WT = INT ((TS + OT) / 2 * 1E3 + .5) / 1E3
810 PO = PO + 1:A(K,PO) = WT:B(K,PO) = DVDT
820 OT = TS:OV = VS

```

DERIVATIVE XY

```
830 NEXT I
835 NMPTS(K) = NMPTS(K) - 1
840 IF (SA) THEN 920
850 PRINT D$"OPEN /CALDAT/X.DYDX/"FIL$;C$(K)
860 PRINT D$"WRITE /CALDAT/X.DYDX/"FIL$;C$(K)
870 PRINT NMPTS(K)
880 FOR L = 1 TO NMPTS(K)
890 PRINT A(K,L): PRINT B(K,L)
900 NEXT L
910 PRINT D$"CLOSE"
920 NEXT K
930 PRINT D$"CHAIN STARTUP,@1500,D1"
```

INTEGRATE XDYDX

```

100 REM *****
110 REM * THIS PROGRAM INTEGRATES X, DY/DX DATA *
120 REM * TO CALCULATE THE PARAMETER Z *
130 REM *****
150 D$ = CHR$(4): SPEED= 255: PRINT D$"PR#3"
160 PRINT "Converting X.DYDX to X.Z"
260 REM PROCESS DATA
305 I = 1:Z = 0:TS = 0:TN = 0
330 IF A(1,I) < = - 36 THEN 370
340 B(1,I) = 0
350 I = I + 1
360 GOTO 330
370 IF A(1,I) < = - 36.5 THEN 420
380 TS = TS + B(1,I)
390 TN = TN + 1
400 I = I + 1
405 B(1,I) = 0
410 GOTO 370
420 Y1 = TS / TN:X1 = A(1,I - 1)
440 FOR K = I TO NMPTS(1)
450 Z = Z + (X1 - A(1,K)) * (B(1,K) - Y1)
460 B(1,K) = Z
475 X1 = A(1,K)
490 NEXT K
520 Z = 0:X1 = A(0,1)
540 FOR K = 1 TO NMPTS(0)
550 IF A(0,K) > - 36.5 THEN 570
560 Z = Z + (X1 - A(0,K)) * (B(0,K) - Y1)
570 B(0,K) = Z
580 X1 = A(0,K)
590 NEXT K
600 IF (SA) THEN 700
610 FOR K = 0 TO 1
620 PRINT D$"OPEN /CALDAT/X.Z/"FIL$;C$(K)
630 PRINT D$"WRITE /CALDAT/X.Z/"FIL$;C$(K)
640 PRINT NMPTS(K)
650 FOR L = 1 TO NMPTS(K)
660 PRINT A(K,L): PRINT B(K,L)
670 NEXT L
680 PRINT D$"CLOSE"
690 NEXT K
700 PRINT D$"CHAIN STARTUP, @1540,D1"

```

CONVERT VP

```

100 REM *****
110 REM * CONVERTS THE X.Z FILE TO X.VP *
120 REM * BASES THE LOW POINT ON THE AVERAGE OF *
130 REM * COOLING DATA FROM -36 TO -36.5 *
140 REM *****
150 PRINT "Converting X.Z to X.VP"
330 FOR K = 0 TO 1
340 FOR L = 1 TO NMPTS(K)
360 VP = 1 / 10 * B(K,L)
370 B(K,L) = VP
380 NEXT L
390 NEXT K
400 IF (SA) THEN 500
410 FOR K = 0 TO 1
420 PRINT D$"OPEN /CALDAT/X.VP/"FIL$;C$(K)
430 PRINT D$"WRITE /CALDAT/X.VP/"FIL$;C$(K)
440 PRINT NMPTS(K)
450 FOR L = 1 TO NMPTS(K)
460 PRINT A(K,L): PRINT B(K,L)
470 NEXT L
490 NEXT K
500 PRINT D$"CHAIN STARTUP,@1580,D1"

```

CONVERT R

```

100 REM *****
110 REM *THIS PROGRAM CONVERTS THE X VS PORE VOLUME FILE *
120 REM * TO RADIUS VS PORE VOLUME. *
130 REM *****
135 TO = VAL (TO$)
140 PRINT "Converting X.VP to R.VP"
320 FOR K = 0 TO 1
325 PO = 0:R1 = 0
330 FOR L = 1 TO NMPTS(K)
422 XO = A(K,L) * 1E - 4
425 R = (( - 2 * TO * (.25 - (29 * XO) + (.25 * TO * XO))) /
      ((.9998 * 3.335E9) * (1 + TO * XO))) * 1E8
427 IF R < 0 THEN 490
428 IF ( ABS (R - R1) < 1) AND (K$ = "Y") THEN 490
450 PO = PO + 1:R1 = R
460 A(K,PO) = R:B(K,PO) = B(K,L)
490 NEXT L
500 NMPTS(K) = PO
510 NEXT K
520 FOR K = 0 TO 1
530 PRINT D$"OPEN /CALDAT/R.VP/"FIL$;C$(K)
540 PRINT D$"WRITE /CALDAT/R.VP/"FIL$;C$(K)
550 PRINT NMPTS(K)
560 FOR L = 1 TO NMPTS(K)
570 PRINT A(K,L): PRINT B(K,L)
580 NEXT L
590 PRINT D$"CLOSE"
600 NEXT K
610 PRINT D$"CHAIN STARTUP,@1600,D1"

```


DERIVATIVE VP

```

100 REM *****
110 REM * THIS PROGRAM CALCULATES THE DERIVATIVE OF *
120 REM * OF THE PORE VOLUME WITH RESPECT TO *
130 REM * PORE RADIUS *
140 REM *****
160 PRINT "Converting R.VP to R.DVPDR"
360 DIM T1(MA),V1(MA),T2(MA),V2(MA)
440 FOR K = 0 TO 1
442 PRINT D$"OPEN /CALDAT/R.DVPDR/"FIL$;C$(K)
443 PRINT D$"WRITE /CALDAT/R.DVPDR/"FIL$;C$(K)
444 PRINT NMPTS(K) - 1
445 FL = 0
446 FOR S = 1 TO MA:T1(S) = 0:V1(S) = 0: NEXT S
450 FOR S = 1 TO MA - 1:T1(S) = A(K,S):V1(S) = B(K,S)
460 NEXT S
465 PI = MA - 1
470 FOR I = INT (MA / 2 + 1) TO MA - 1:T = 0:V = 0
480 FOR J = 1 TO I:T = T + T1(J):V = V + V1(J): NEXT J
490 TS = T / I
500 VS = V / I
510 IF NOT FL THEN FL = 1: GOTO 570
520 DVDT = INT ((VS - OV) / (TS - OT) * 1E5 + .5) / 1E5
530 WT = INT ((TS + OT) / 2 * 1E3 + .5) / 1E3
540 FL = FL + 1
560 PRINT WT: PRINT DVDT: IF FL = 2 THEN PRINT WT: PRINT DVDT
570 OT = TS:OV = VS
580 NEXT I
590 FOR I = 1 TO NMPTS(K) - MA + 1
600 PI = PI + 1:TE = A(K,PI):VO = B(K,PI)
610 IF I > NMPTS(K) - 2 * MA + 2 THEN Z = I - NMPTS(K) + 2 * MA - 1:
    T2(Z) = TE:V2(Z) = VO
620 T = T + TE - T1(S):T1(S) = TE
630 V = V + VO - V1(S):V1(S) = VO
640 S = S + 1: IF MA < S THEN S = 1
650 TS = T / MA
660 VS = V / MA
680 IF TS = OT THEN DVDT = 00: GOTO 700
690 DVDT = INT ((VS - OV) / (TS - OT) * 1E5 + .5) / 1E5:00 = DVDT
700 WT = INT ((TS + OT) / 2 * 1E3 + .5) / 1E3
710 PRINT WT: PRINT DVDT
720 OT = TS:OV = VS
730 NEXT I
740 FOR I = 2 TO INT (MA / 2 + 1):T = 0:V = 0
750 FOR J = I TO MA:T = T + T2(J):V = V + V2(J): NEXT J
760 TS = T / (J - I)
770 VS = V / (J - I)
790 DVDT = INT ((VS - OV) / (TS - OT) * 1E5 + .5) / 1E5
800 WT = INT ((TS + OT) / 2 * 1E3 + .5) / 1E3

```

DERIVATIVE VP

```
810 PRINT WT: PRINT DVDT
820 OT = TS:OV = VS
830 NEXT I
910 PRINT D$ "CLOSE"
920 NEXT K
930 PRINT D$ "CHAIN STARTUP,@1630,D1"
```

DERIVATIVE LVP

```

100 REM *****
110 REM * THIS PROGRAM CALCULATES THE DERIVATIVE OF *
120 REM * THE PORE VOLUME W/RESPECT THE LOG OF THE *
130 REM * PORE RADIUS *
140 REM *****
150 GOSUB 1000
160 PRINT "Converting R.VP to R.DVPDLR"
440 FOR K = 0 TO 1
445 FL = 0
446 FOR S = 1 TO MA:T1(S) = 0:V1(S) = 0: NEXT S
450 FOR S = 1 TO MA - 1:T1(S) = A(K,S):V1(S) = B(K,S)
460 NEXT S
465 PO = 0:PI = MA - 1
470 FOR I = INT (MA / 2 + 1) TO MA - 1:T = 0:V = 0
480 FOR J = 1 TO I:T = T + T1(J):V = V + V1(J): NEXT J
490 TS = T / I
500 VS = V / I
510 IF NOT FL THEN FL = 1: GOTO 570
520 DVDT = INT ((VS - OV) / (TS - OT) * 1E5 + .5) / 1E5
530 WT = INT ((TS + OT) / 2 * 1E3 + .5) / 1E3
540 FL = FL + 1
550 PO = PO + 1
560 A(K,PO) = WT:B(K,PO) = DVDT: IF FL = 2 THEN PO = PO + 1:A(K,PO) =
    WT:B(K,PO) = DVDT
570 OT = TS:OV = VS
580 NEXT I
590 FOR I = 1 TO NMPTS(K) - MA + 1
600 PI = PI + 1:TE = A(K,PI):VO = B(K,PI)
610 IF I > NMPTS(K) - 2 * MA + 2 THEN Z = I - NMPTS(K) + 2 * MA - 1:T2(Z)
    = TE:V2(Z) = VO
620 T = T + TE - T1(S):T1(S) = TE
630 V = V + VO - V1(S):V1(S) = VO
640 S = S + 1: IF MA < S THEN S = 1
650 TS = T / MA
660 VS = V / MA
680 IF TS = OT THEN DVDT = 00: GOTO 700
690 DVDT = INT ((VS - OV) / (TS - OT) * 1E5 + .5) / 1E5:00 = DVDT
700 WT = INT ((TS + OT) / 2 * 1E3 + .5) / 1E3
710 PO = PO + 1:A(K,PO) = WT:B(K,PO) = DVDT
720 OT = TS:OV = VS
730 NEXT I
740 FOR I = 2 TO INT (MA / 2 + 1):T = 0:V = 0
750 FOR J = I TO MA:T = T + T2(J):V = V + V2(J): NEXT J
760 TS = T / (J - I)
770 VS = V / (J - I)
790 DVDT = INT ((VS - OV) / (TS - OT) * 1E5 + .5) / 1E5
800 WT = INT ((TS + OT) / 2 * 1E3 + .5) / 1E3
810 PO = PO + 1:A(K,PO) = WT:B(K,PO) = DVDT

```

DERIVATIVE LVP

```
820 OT = TS:OV = VS
830 NEXT I
835 NMPTS(K) = NMPTS(K) - 1
850 PRINT D$"OPEN /CALDAT/R.DVPDLR/"FIL$;C$(K)
860 PRINT D$"WRITE /CALDAT/R.DVPDLR/"FIL$;C$(K)
870 PRINT NMPTS(K)
880 FOR L = 1 TO NMPTS(K)
885 A(K,L) = 106 A(K,L)
890 PRINT A(K,L): PRINT B(K,L)
900 NEXT L
910 PRINT D$"CLOSE"
920 NEXT K
925 FOR T = 1 TO 5: PRINT CHR$(7): NEXT T
930 PRINT D$"CHAIN STARTUP,@180 ,D1"
1000 REM *****
1010 REM * TO LOG R *
1020 REM *****
1025 LT = LOG(10)
1030 FOR K = 0 TO 1
1040 FOR L = 1 TO NMPTS(K)
1050 A(K,L) = LOG(A(K,L)) / LT
1060 NEXT L
1070 NEXT K
1080 RETURN
```

SURFACE AREA

```

100 REM *****
110 REM * PROGRAM TO CALCULATE SURFACE AREA *
120 REM *****
125 D$ = CHR$(4)
130 PRINT "Calculating the surface area..."
135 IF DT = 0 THEN GOSUB 2000
140 A = 0
150 FOR L = 2 TO NMPTS(0)
155 IF A(0,L) < = 12 THEN 170
160 A = A + (1 / A(0,L - 1) - 1 / A(0,L)) * (PV * B(0,L - 1)) * - 1E4
170 NEXT L
210 A = INT (A * 2.5 * 1E4) / 1E4
220 PRINT "Surface Area = ";A
230 PRINT : PRINT
260 END
2000 REM *****
2010 REM * FILE LOADER *
2020 REM *****
2030 INPUT "Base Filename : ";FIL$
2040 IF FIL$ = "" THEN PRINT D$"CATALOG /CALDAT/R.VP/": GOTO 2000
2050 PRINT D$"OPEN /CALDAT/R.VP/";FIL$;" ,D2"
2060 PRINT D$"READ /CALDAT/R.VP/";FIL$
2070 INPUT NMPTS(0)
2080 DIM A(1,NMPTS(0)),B(1,NMPTS(0))
2090 FOR I = 1 TO NMPTS(0)
2100 INPUT A(0,I),B(0,I)
2110 NEXT I
2120 PRINT D$"CLOSE"
2130 INPUT "Pore volume of Sample : ";PV$
2140 PV = VAL (PV$)
2150 RETURN
2160 INPUT DIL$,CO$,DT$
2165 INPUT MA$,N1$,N2$
2170 PRINT D$"CLOSE"
2180 RETURN

```

TEST FORMAT

```

10 REM READ DATA ON SAMPLE AND PLOT OUT
15 D$ = CHR$ (4)
20 HOME : VTAB 5
30 PRINT "Insert a /RAWDAT disk in drive 2 and hit a key...";
40 GET TMP$
50 PRINT D$"CATALOG,D2"
60 PRINT : INPUT "Enter new file name : ";F$
70 IF F$ = "" THEN 60
80 GOSUB 200: REM READ IN DATA
90 GOSUB 500: REM PRINT TO SCREEN
100 GOSUB 1000: REM PLOT OUT
110 GOTO 90
200 PRINT D$"OPEN ";F$;",D2"
210 PRINT D$"READ ";F$
215 INPUT SCYCLE
220 INPUT CT
230 FOR X = 1 TO CT
240 INPUT TP(X),BTM(X),RTE(X),IT(X),TD$(X)
245 IF TD$(X) = "N" THEN IT(X) = 0
250 NEXT X
251 INPUT MCYCLE
253 DIM TC(MCYCLE)
254 FOR X = 1 TO MCYCLE
256 INPUT TC(X)
258 NEXT X
260 INPUT N1$,N2$,N3$,N4$
270 PRINT D$"CLOSE ";F$
280 RETURN
500 REM PRINT TO SCREEN
510 HOME : VTAB 2
520 HTAB INT ( LEN (F$) / 2)
530 PRINT F$
540 VTAB 4
550 PRINT "Cycle Type";: POKE 36,15: PRINT "Top Temp";: POKE 36,30:
PRINT "Bottom Temp";: POKE 36,45: PRINT "Rate";: POKE 36,55:
PRINT "Sample Interval"
560 FOR X = 1 TO CT
570 PRINT TAB( 5);X;: POKE 36,20: PRINT TP(X);: POKE 36,34:
PRINT BTM(X);: POKE 36,46: PRINT RTE(X);: POKE 36,62: PRINT IT(X)
572 NEXT X
574 VTAB 9: PRINT "Cycle Type Cycle Type Cycle Type Cycle Type Cycle
Type Cycle Type"
576 FOR Y = 0 TO MCYCLE - 1
578 X = 2 + INT (Y / 8) * 11
580 VTAB 10 + (Y - INT (Y / 8) * 8): POKE 36,X
582 PRINT Y + SCYCLE;: POKE 36,X + 6: PRINT TC(Y + 1)
584 NEXT Y
590 VTAB 18: PRINT "NOTES : "

```

TEST FORMAT

```

600 PRINT N1$
610 PRINT N2$
620 PRINT N3$
630 PRINT N4$
640 VTAB 23: HTAB 10
650 PRINT " 1. Plot Out to HP"
655 VTAB 23: POKE 36,40
660 PRINT " 2. Quit"
670 INPUT "Enter menu choice (1-2) : ";TMP$
680 IF TMP$ > "2" OR TMP$ < "1" THEN 510
700 IF TMP$ = "1" THEN RETURN
710 PRINT "DONE"
720 END
1000 REM *****
1010 REM * PLOT OUT TO PLOTTER *
1020 REM *****
1030 REM INIT HP
1035 C$ = CHR$ (3)
1040 PRINT D$"PR#2"
1050 PRINT CHR$ (27)".I40;0;17:"
1060 PRINT "IN;"
1070 PRINT CHR$ (27)".I40;0;17:"
1075 PRINT CHR$ (27)".N;19:"
1080 PRINT "SP1;SI.4,.6;"
1090 PRINT "DI1,0;VS20;"
1095 PRINT "PA"5150 - LEN (F$) / 2 * 240;" ,6500;LB";F$;C$
1096 PRINT "SI.2,.3;"
1100 PRINT "PA1000,5650;"
1110 PRINT "LBCycle Type      Top Temp      Bottom Temp      Rate      Sample
Interval";C$
1120 PRINT "PA1000,5450;PD;PA9280,5450;PU;"
1130 FOR X = 1 TO CT
1140 PRINT "PA1600,";5450 - 200 * X;" ;LB";X;C$
1150 PRINT "PA3400,";5450 - 200 * X;" ;LB";TP(X);C$
1160 PRINT "PA5080,";5450 - 200 * X;" ;LB";BTTM(X);C$
1170 PRINT "PA6520,";5450 - 200 * X;" ;LB";RTE(X);C$
1180 PRINT "PA8440,";5450 - 200 * X;" ;LB";IT(X);C$
1182 NEXT X
1183 PRINT "PA1000,4250;PD;PA9280,4250;PU;"
1184 PRINT "PA1000,4050;"
1185 PRINT "LBCycle Type  Cycle Type  Cycle Type  Cycle Type  Cycle Type
Cycle Type";C$
1186 PRINT "PA1000,3850;PD;PA9280,3850;PU;"
1187 FOR Y = 0 TO MCYCLE - 1
1188 X = 2 + INT (Y / 8) * 12
1190 PRINT "PA"1000 + X * 120," ,3650 - 200 * (Y - INT (Y / 8) * 8)";"
1192 PRINT "LB"Y + SCYCLE;C$;"PA"1720 + X * 120," ,3650 - 200 *
(Y - INT (Y / 8) * 8)";"

```

TEST FORMAT

```
1194 PRINT "LB"TC(Y + 1);C$
1198 NEXT Y
1200 PRINT "PA1000,1750;LBNotes :";C$
1210 PRINT "PA1250,1550;LB";N1$;C$
1220 PRINT "PA1250,1350;LB";N2$;C$
1230 PRINT "PA1250,1150;LB";N3$;C$
1240 PRINT "PA1250,950 ;LB";N4$;C$
1250 PRINT D$"PR#3"
1260 RETURN
```


TEMP MEASUREMENT

VARIABLES

C100 BASE ADDRESS FOR I/O CARD
 C100 OUTPUT REGISTER B
 C101 OUTPUT REGISTER A
 C104 TIMER 1, LATCH LOW
 C105 TIMER 1, COUNTER HIGH
 C108 TIMER 2, LATCH LOW
 C109 TIMER 2, COUNTER HIGH
 C10B AUXILLIARY CONTROL REGISTER
 C10C PERIPHERAL CONTROL REGISTER
 C10D INTERRUPT FLAG REGISTER

PROGRAM

0340	A2 70	LDX #\$70	
0342	8E A2 70	STX \$70A2	CA1 POSITIVE ACTIVE EDGE
0345	8E 0C C1	STX \$C10C	CB1 NEGATIVE ACTIVE EDGE
0348	A0 00	LDY #\$00	
034A	8C 0B C1	STY #C10B	T2 TIMED INTERRUPTS
034D	A9 02	LDA #\$02	SWITCHING MASK
034F	AC 38 03	LDY \$0338	
0352	F0 02	BEQ \$0356	IF EQUAL THEN DON'T SWITCH
0354	A9 10	LDA #\$10	MASK
0356	A0 FF	LDY #\$FF	
0358	8C 0D C1	STY \$C10D	CLEAR INTERRUPT FLAGS
035B	2C 0D C1	BIT \$C10D	FIRST ACTIVE EDGE?
035E	F0 FB	BEQ \$0358	NO, THEN CONTINUE POLLING
0360	8C 08 C1	STY \$C108	
0363	8C 09 C1	STY \$C109	START COUNTING PERIOD
0366	8C 0D C1	STY \$C10D	CLEAR INTERRUPT FLAGS
0369	2C 0D C1	BIT \$C10D	SECOND ACTIVE EDGE?
036C	F0 FB	BEQ \$0369	NO, THEN CONTINUE POLLING
036E	AC 08 C1	LDY \$C108	GET LOW BYTE OF PERIOD
0371	AE 09 C1	LDX \$C109	GET MID BYTE OF PERIOD
0374	8C 36 03	STY \$0336	STORE PERIOD
0377	8E 37 03	STX \$0337	STORE PERIOD
037A	60	RTS	RETURN TO PROGRAM

CALIBRATE

```

100 REM *****
110 REM * DATA ENTRY PROGRAM *
120 REM *****
125 D$ = CHR$(4)
235 CO = 1
240 REM RETRIEVE AND DISPLAY FILE
250 HOME : GOSUB 1000
260 POKE 34,0
270 VTAB 1: HTAB 12
280 FIL$ = "CAL": PRINT "CAL"
285 VTAB 1: HTAB 40: PRINT "# of points : ";
290 GOSUB 2000
295 N = 1
300 VTAB 1: POKE 36,55: PRINT NMPT
310 VTAB 2: HTAB 16: PRINT XA$
320 VTAB 3: HTAB 16: PRINT YA$
330 GOSUB 1300
340 N = 1
350 GOSUB 1500
360 X = 1:Y = 5
370 GOSUB 1900
380 GOTO 500
400 REM INPUT NEW FILE
410 HOME : GOSUB 1000
415 POKE 34,0
420 VTAB 1: POKE 36,55
430 INPUT "";NMPT
440 DIM X(NMPT),Y(NMPT)
445 POKE 34,4
450 N = 1
460 GOSUB 1500
480 GOSUB 1300
485 X = 1:Y = 5
490 GOSUB 1900
500 REM *****
510 REM * MAIN CONTROLLER FOR EDITING AND ENTRY *
520 REM *****
530 GET KB$
540 IF (KB$ > = "0" AND KB$ < = "9") OR KB$ = "." OR KB$ = "-" THEN 3000
550 IF ASC (KB$) = 8 THEN 3200: REM LEFT ARROW
560 IF ASC (KB$) = 21 OR ASC (KB$) = 32 THEN 3300: REM RIGHT ARROW
570 IF ASC (KB$) = 10 THEN 3400: REM DOWN ARROW
580 IF ASC (KB$) = 11 THEN 3500: REM UP ARROW
590 IF ASC (KB$) = 19 THEN 3600: REM CONTROL S
610 IF ASC (KB$) = 17 THEN 4200: REM CONTROL Q
620 GOTO 530
1000 REM *****
1010 REM * DISPLAY HEADINGS *

```

CALIBRATE

```

1020 REM *****
1030 POKE 34,0: REM SET TOP TO LINE 0
1040 POKE 35,4: REM SET BOTTO TO LINE 3
1050 HOME
1060 PRINT "Filename : ";: HTAB 40: PRINT "# of points : "
1070 PRINT "X-Axis Label : "
1080 PRINT "Y-Axis Label : "
1090 PRINT "Point #"; TAB( 30);"X";: POKE 36,50: PRINT "Y";
1100 POKE 34,4: POKE 35,23
1110 RETURN
1300 REM *****
1310 REM * PRINTS BOTTOM LINE *
1320 REM *****
1330 POKE 34,23: POKE 35,24
1340 HOME
1350 PRINT "Control-S to Save      Control-Q to Exit";
1360 POKE 34,4: POKE 35,23
1370 RETURN
1500 REM *****
1510 REM * DISPLAY A PAGE OF DATA (19 LINES) *
1520 REM * N = THE DATA NUMBER TO START FROM *
1530 REM *****
1540 HOME
1550 FOR I = N TO N + 18
1560 IF I > NMPT THEN 1590
1570 PRINT TAB( 3);I; TAB( 25);X(I);: POKE 36,45: PRINT Y(I);
1580 IF I < > N + 18 THEN PRINT
1590 NEXT I
1600 X = 1:Y = 5
1610 GOSUB 1900
1620 RETURN
1700 REM *****
1710 REM * SCROLL UP ONE LINE *
1720 REM *****
1725 IF N + 18 > NMPT THEN 1800
1730 VTAB 23: POKE 36,78
1735 PRINT
1740 I = N + 18
1750 IF I > NMPT THEN 1780
1760 PRINT TAB( 3);I; TAB( 25);X(I);: POKE 36,45: PRINT Y(I);
1780 X = 1:Y = 23
1790 GOSUB 1900
1800 RETURN
1900 REM *****
1910 REM * MOVE CURSOR TO APPROPRIATE X,Y POSITION *
1920 REM *****
1930 VTAB Y
1940 IF X = 1 THEN HTAB 25

```

CALIBRATE

```

1950 IF X = 2 THEN POKE 36,45
1960 RETURN
2000 REM *****
2010 REM * RETRIEVE FILE *
2020 REM *****
2030 PRINT D$"OPEN "FIL$",D1"
2040 PRINT D$"READ "FILE$
2050 INPUT NMPT
2060 DIM X(NMPT),Y(NMPT)
2065 IF CO = 1 THEN 2080
2070 INPUT XA$
2072 IF XA$ = "" THEN INPUT XA$
2075 INPUT YA$
2077 IF YA$ = "" THEN INPUT YA$
2080 FOR I = 1 TO NMPT
2090 INPUT X(I),Y(I)
2100 NEXT I
2110 PRINT D$"CLOSE"
2120 RETURN
2300 REM *****
2310 REM * SAVE FILE *
2320 REM *****
2330 PRINT D$"OPEN "FIL$",D1"
2340 PRINT D$"WRITE"FIL$
2350 PRINT NMPT
2352 IF CO = 1 THEN 2370
2355 IF XA$ = "" THEN XA$ = "none"
2357 IF YA$ = "" THEN YA$ = "none"
2360 PRINT XA$: PRINT YA$
2370 FOR I = 1 TO NMPT
2380 PRINT X(I): PRINT Y(I)
2390 NEXT I
2400 PRINT D$"CLOSE"
2410 RETURN
3000 REM *****
3010 REM * ENTER THE DATA INTO ARRAY *
3020 REM *****
3025 PRINT KB$;
3030 GET IN$
3040 IF (IN$ > = "0" AND IN$ < = "9") OR IN$ = "." OR IN$ = "-" OR IN$ =
"E" THEN KB$ = KB$ + IN$: PRINT IN$;: GOTO 3030
3042 IF ASC (IN$) = 8 AND LEN (KB$) = 1 THEN KB$ = "": PRINT CHR$ (8);"
"; CHR$ (8);
3044 IF ASC (IN$) = 8 AND LEN (KB$) > 1 THEN KB$ = LEFT$ (KB$, LEN
(KB$) -1): PRINT CHR$ (8);" "; CHR$ (8);
3050 IF ASC (IN$) = 13 THEN 3070
3060 GOTO 3030
3070 IF X = 1 THEN X(N + Y - 5) = VAL (KB$)

```

CALIBRATE

```

3080 IF X = 2 THEN Y(N + Y - 5) = VAL (KB$)
3090 GOSUB 1900: REM RESET CURSOR
3100 PRINT VAL (KB$);" ";
3110 IF X = 1 THEN X = 2: GOTO 3140
3120 IF (Y - 5) + N < > NMPT THEN Y = Y + 1: X = 1
3130 IF Y > 23 THEN N = N + 1: GOSUB 1700
3140 GOSUB 1900
3150 GOTO 530
3200 REM *****
3210 REM * LEFT ARROW *
3220 REM *****
3230 IF X = 2 THEN X = 1: GOSUB 1900
3240 GOTO 530
3300 REM *****
3310 REM * RIGHT ARROW *
3320 REM *****
3330 IF X = 1 THEN X = 2: GOSUB 1900
3340 GOTO 530
3400 REM *****
3410 REM * DOWN ARROW *
3420 REM *****
3430 IF (Y - 5) + N < > NMPT THEN Y = Y + 1
3440 IF Y > 23 THEN N = N + 1: Y = 23: GOSUB 1700
3450 GOSUB 1900
3460 GOTO 530
3500 REM *****
3510 REM * UP ARROW *
3520 REM *****
3530 IF Y = 5 AND N < > 1 THEN N = N - 1: GOSUB 1500: GOTO 3560
3540 IF Y < > 5 THEN Y = Y - 1
3550 GOSUB 1900
3560 GOTO 530
3600 REM *****
3610 REM * CONTROL-S SAVE DATA *
3620 REM *****
3625 IF FIL$ = "" THEN 4000
3630 POKE 34,23: POKE 35,24
3640 HOME
3650 PRINT "Saving ";FIL$;" ...";
3660 GOSUB 2300
3670 POKE 34,0
3680 CLEAR
3690 GOTO 9000
4000 REM *****
4010 REM * TRIED TO SAVE WITHOUT A FILE NAME *
4020 REM *****
4030 POKE 34,23: POKE 35,24
4040 HOME

```

CALIBRATE

```
4050 PRINT "You must have a filename to save data.";
4060 GET KB$
4070 GOSUB 1300
4080 GOSUB 1900
4090 GOTO 530
4200 REM *****
4210 REM * ESCAPE SO EXIT OR NOT *
4220 REM *****
4230 POKE 34,23: POKE 35,24
4240 HOME
4250 PRINT "This will leave the data unchanged. Continue (Y/N Default =
N).";
4260 GET KB$
4270 IF KB$ = "Y" THEN 9000
4280 GOSUB 1300
4290 GOSUB 1900
4300 GOTO 530
9000 REM WE GONE BYE BYE
9010 PRINT D$"RUN STARTUP,D1"
```